Metal-free Cysteamine Functionalized Graphene Alleviates Mutual Interferences in Heavy Metals Electrochemical Detection

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Abstract:

The detection of heavy metal pollutants is of great concern in environmental monitoring due to the potent toxicity. Electrochemical detection, one of the foremost sensing techniques, is hindered by the mutual interference between target heavy metal ions. In particular, the sensitivity to Cd²+ (one of the most toxic heavy metals) is often overshadowed by other heavy metals (e.g., Pb²+ and Cu²+) when using carbon electrodes. [1,2] Strategies are frequently required using additional metallic particles embedded in the electrode, which faces recycling and reusability issues. In this study, a metal-free cysteamine covalently functionalized graphene (GSH), is employed to tackle this issue by selectively enhancing a 6-fold boost in Cd²+sensitivity of screen-printed carbon sensors (SPCE), while the sensitivities to Pb²+ and Cu²+ not being influenced in the simultaneous detection. The selective enhancement is attributed to grafted thiols on GSH, which have good affinity to Cd²+ based on Pearson's hard and soft acid and base principle. Moreover, GSH-SPCE features high reusable times (23 times) due to the covalent functionalization of thiols, surpassing the state-of-art SPCEs modified by non-covalently functionalized graphene derivatives. Finally, GSH-SPCE was validated in tap water.

References

- [1] Yang, Qiuyue, et al. "Development of a heavy metal sensing boat for automatic analysis in natural waters utilizing anodic stripping voltammetry." ACS ES&T Water 1.12 (2021): 2470-2476.
- [2] Borrill, Alexandra J., Nicole E. Reily, and Julie V. Macpherson. "Addressing the practicalities of anodic stripping voltammetry for heavy metal detection: a tutorial review." Analyst 144.23 (2019): 6834-6849.

Figures

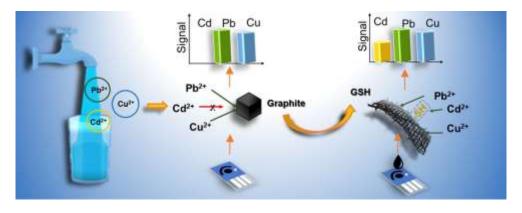


Figure 1: Schematic illustration of the significant effect of GSH relieving the mutual interference in simultaneous multiple heavy metals detection.