

Tailoring few-layers van der Waals crystals through galvanostatic molecular intercalation

Daniel Tezze¹

José M. Pereira¹, Covadonga A. García¹, Enrique A. García¹, Felix Casanova^{1,3}, Beatriz Martín-García^{1,3}, Maider Ormazá⁴, Luis E. Hueso^{1,3} and Marco Gobbi^{2,3}.

¹ CIC nanoGUNE BRTA, 20018 San Sebastian, Spain.

² Materials Physics Center CSIC-UPV/EHU, 20018 Donostia-San Sebastian, Spain.

³ IKERBASQUE, Basque Foundation for Science, 48013 Bilbao, Spain.

⁴ Departamento de Polímeros y Materiales Avanzados: Física, Química y Tecnología, Universidad del País Vasco, Paseo Manuel de Lardizabal 3, San Sebastián 20018, Spain.

d.tezze@nanogune.eu

Intercalation of guest species in van der Waals materials is an effective technique for customizing their properties.[1–2] The process can be achieved through different methods, that have been explored predominantly for bulk samples[3], which are impractical for the fabrication of novel nanodevices. The use of intercalated compounds in (quantum) electronic devices [4] demands controlled intercalation regimes in reduced-dimensions systems (thin flakes). This task is particularly challenging due to the complexity of the setup involved in conventional intercalation approaches. Herein, we present a paradigmatic novel route based on a galvanostatic electrochemical approach (**fig. 1a**) to directly intercalate pre-exfoliated 2H-TaS₂ flakes with cationic molecular species, revolutionizing its the state-of-the-art intercalation strategy. Our methodology yields intercalated 2H-TaS₂ flakes with unprecedented crystallinity (**fig. 1b**), crucial for integration in devices for electronics and quantum computing. Different molecular cations can be controllably intercalated, permitting to tune the on-set of superconductivity (**fig. 1c**). Moreover, we tailor the intercalation mechanism to spatially control phases within a single flake, potentially enabling van der Waals lateral heterojunctions.

References

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Figures

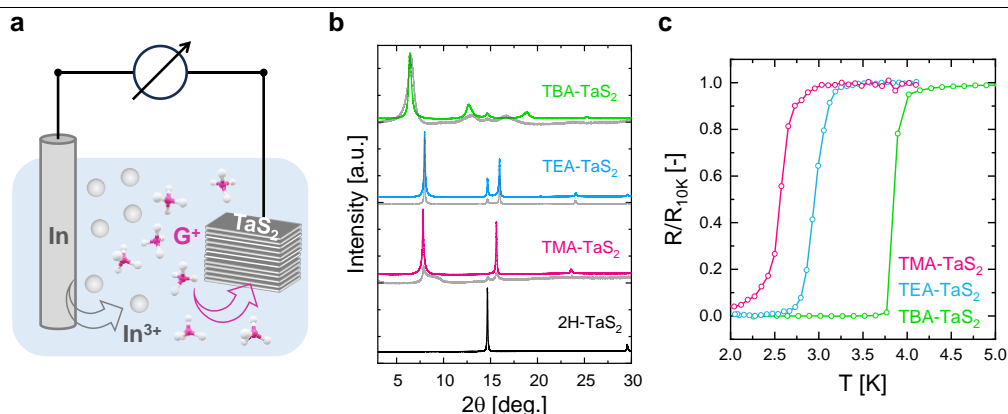


Figure 1: **a)** Schematic of the galvanostatic cell employed for the intercalation of 2H-TaS₂ with organic ions (G⁺). **b)** XRD patterns of pristine 2H-TaS₂ and three alkylammonium intercalates (TMA_xTaS₂, TEA_xTaS₂ and TBA_xTaS₂), obtained by the galvanic approach (coloured lines) and the XRD patterns of the same intercalates using the chemical approach (grey lines). **c)** Normalized resistance measured for three alkylammonium intercalated 2H-TaS₂ thin flakes (~10L).