## Structure and dynamics of water at feldspar surfaces from machine learning augmented molecular simulation

Wanqi Zhou<sup>1</sup>, Pablo M. Piaggi<sup>1,2</sup> (Arial 10) <sup>1</sup>CIC Nanogune, Tolosa Hiribidea 76, 20018 San Sebastian, Spain <sup>2</sup>Ikerbasque, Basque Foundation for Science, Bilbao 48013, Spain

w.zhou@nanogune.eu

## Abstract:

Over the last decade, significant evidence has been gathered that points to feldspar minerals as the most important ice nucleating particles in our planet's atmosphere [1]. There have been efforts to understand the formation of ice at feldspar surfaces using experimental tools such as optical microscopy [2], electron microscopy [3], and atomic force microscopy (AFM) [4], as well as computational tools such as density-functional theory (DFT) calculations [5], and molecular dynamics (MD) simulations [6, 7]. In spite of these efforts, the microscopic characteristics of the nucleation site and the atomicscale mechanism of ice nucleation at feldspar surfaces have not yet been fully elucidated.

Here, we develop a machine-learning potential the (MLP) model interactions at the to water/microcline feldspar interface, using a dataset generated from density-functional theory (DFT) calculations based on the SCAN exchangecorrelation functional [8]. The MLP accurately reproduces the energies and forces obtained from calculations. We then performed DFT MD simulations, driven by the MLP, to investigate the structural and dynamic properties of water on various fully hydroxylated terminations of the (100), (010), and (001) surfaces of microcline feldspar. With these simulations, we explore the effects of ions and surface structure on ice nucleation. Our results provide important insights into the effects of solid-liquid interface on the structure and dynamics of liquid water. These insights contribute to unravelling the microscopic mechanism of ice nucleation on feldspar.

## References

[1] Atkinson, J. D., et al., *Nature*, **498** (2013), 355-358.

- [2] Kiselev, A., et al., Science, 355 (2017), 367-371.
- [3] Whale, T. F., et al., *Phys. Chem. Chem. Phys.*, **19** (2017), 31186-31193.
- [4] Franceschi, G., et al., *J. Phys. Chem. Lett.*, **15** (2024), 15-22.
- [5] Pedevilla, P., et al., *J. Phys. Chem. C*, **120** (2016), 6704-6713.
- [6] Soni, A.; Patey, G. N., *J. Chem. Phys.*, **150** (2019), 214501.

[7] Kumar, A.; Bertram, A. K.; Patey, G. N., ACS *Earth Space Chem.*, **5** (2021), 2169-2183.
[8] Piaggi, P. M., et al., *Faraday Discuss.*, **249** (2024), 98-113.

## Figures



Figure 1. Background of ice nucleation on feldspar surfaces: Feldspar exhibits the highest efficiency for ice formation, yet the underlying mechanism remains unknown.



Figure 2. Schematic summary of the training procedure for the machine-learning potential based on active learning.