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Significant improved dehydrogenation of LiAlH₄ doped with two-dimensional Ti₃C₂

Lithium alanate (LiAlH₄) has attracted intense interest as one of the most promising candidates for hydrogen storage due to its high hydrogen storage capacity (10.5 wt%) and low cost. However, the drawbacks of its dehydrogenation process are the relatively high temperatures and the slow dehydrogenation kinetics. In this study, a two-dimensional Ti₃C₂ MXene was doped into LiAlH₄ to improve its hydrogen storage properties. The composition evolution and dehydrogenation performance of the as-prepared sample were characterized by means of XRD (X-ray diffraction), FT-IR (Fourier transform infrared spectroscopy) and Sieverts-type pressure-composition-temperature apparatus (PCT) measurements, respectively. LiAlH₄-5 wt% Ti₃C₂ samples start to release hydrogen at 48.8 °C, which is 154.2 °C lower than that of as-received LiAlH₄. Isothermal desorption measurements show that the 5 wt% Ti₃C₂-doped sample releases 5.3 wt% of hydrogen within 15 min at 200 °C.

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Figures

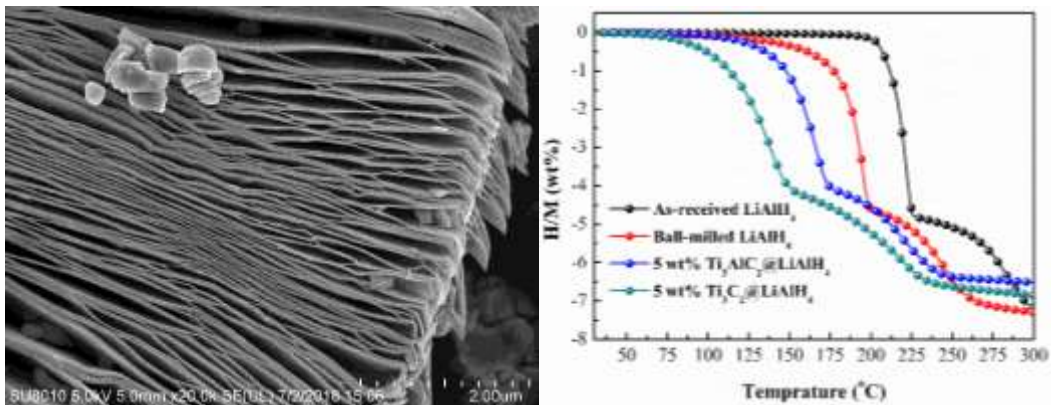


Figure 1: Fig.1 SEM images of Ti₃C₂

Figure 2: Volumetric release curves of the prepared samples.