
Hisaki Sawahata

Mina Maruyama and Susumu Okada

Graduate School of Pure and Applied Sciences, University of Tsukuba, Tsukuba 305-8571, Japan

hsawahata@comas.frsc.tsukuba.ac.jp

Energetics and electronic properties of B₃N₃-doped graphene: Semiconducting graphene heterostructures

Electronic structures of porous sp² carbon networks strongly depend on their network topology and an arrangement of the pores, making them a semiconductor or metal. In addition to pores, interfaces with other two-dimensional materials also modulate the electronic structure of graphene depending on the interface shape and constituents. These facts open a question whether the energetics and electronic structure of porous graphene are further tunable by filling pores with nanoflakes of other two-dimensional materials. Among various two-dimensional materials, h-BN is an appropriate dopant. Thus, in this work, we aim to investigate the energetics and electronic structure of B₃N₃-doped graphene in which B₃N₃ (borazine) is embedded into the graphene nanomeshes consisting of triangularly arranged hexagonal vacancies, using the density functional theory with the generalized gradient approximation.

Figure 1 shows an optimized structure of the representative B₃N₃-doped graphene. The doped B₃N₃ rings form triangular lattice in graphene network by forming the heterobonds of BC and NC at their borders. Figure 2 shows the total energy of B₃N₃-doped graphene as a function of the spacing between adjacent B₃N₃ domains. The total energy is found to be inversely proportional to B₃N₃ spacing. Furthermore, the energy also weakly depends on the network topology of the graphene domain. The heterosheets with the Kekulé structure have lower total energy than the other heterosheets for the same B₃N₃ spacing. The network topology of the graphene region also affects the electronic structures of B₃N₃-doped graphene. Figure 3 shows the band gap of the heterosheets as a function of the spacing between adjacent B₃N₃ domains, indicating that all sheets are semiconductors with finite energy gap irrespective of the arrangements of B₃N₃ in graphene. The gap monotonically decreases with increasing the B₃N₃ spacing. The heterosheets with the Kekulé structure have a direct gap at the Γ point, while the remaining heterosheets have direct gap at K point. Calculated band gap for the heterosheet with the Kekulé structure is narrower than that with the non-Kekulé structure for the same B₃N₃ spacing.

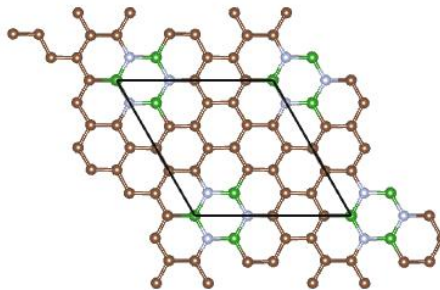


Figure 1: Optimized structure of B₃N₃-doped graphene. Brown, green, and cyan circles denote C, B, and N atoms, respectively.

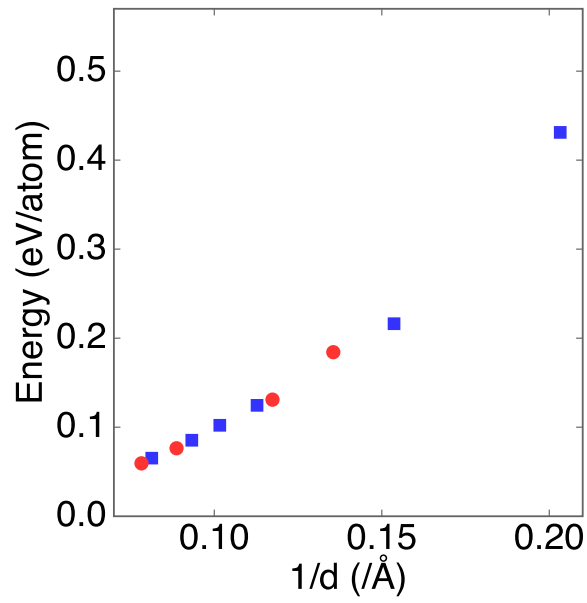


Figure 2: Total energy of B_3N_3 -doped graphene as a function of the inverse of the B_3N_3 spacing. Circles and squares denote the energy of the heterosheets with and without the Kekule structures, respectively.

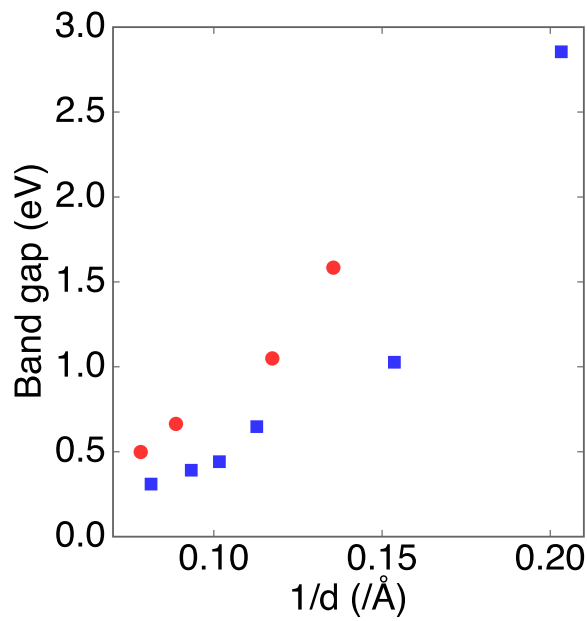


Figure 3: Band gap of B_3N_3 -doped graphene as a function of the inverse of the B_3N_3 spacing. Circles and squares denote the energy of the heterosheets with and without the Kekule structures, respectively.