

Graphene and Boron-doped Graphene by Pulsed Laser Deposition

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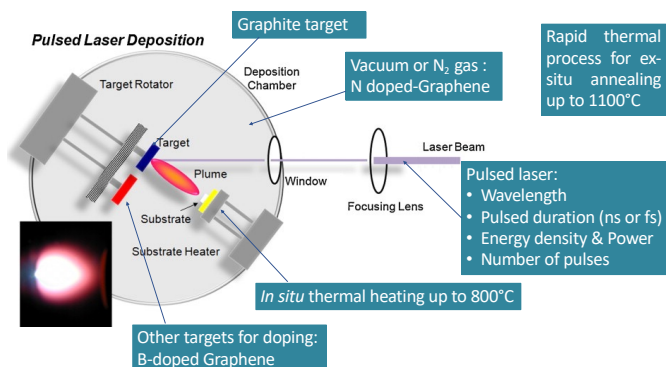
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1. Graphene synthesis from Pulsed Laser Deposition



Very few study on graphene synthesis using PVD process.

PVD advantages are:

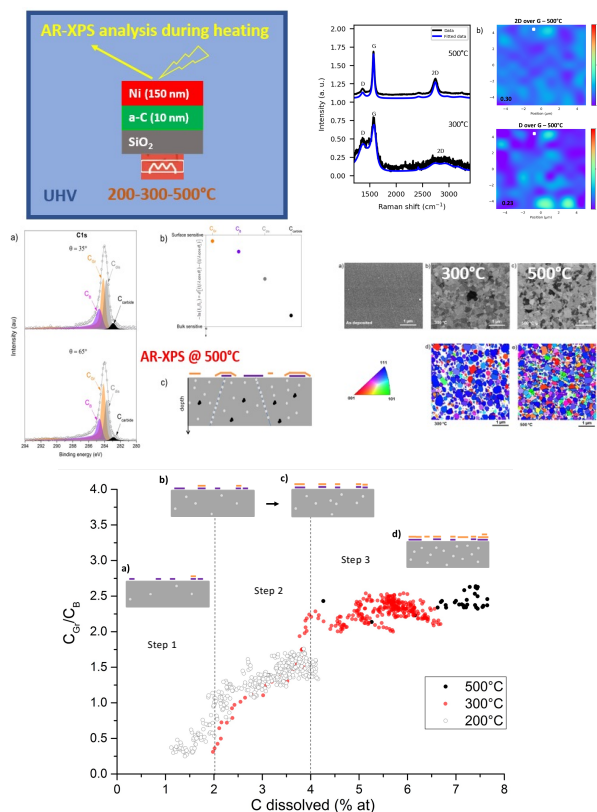
- Thickness control of carbon/metal coatings.
- Doping (B, N...) control in the carbon coating.
- Lower Temperature (vs CVD).

PLD as a PVD Versatile Process for graphene growth:

- Control of Doping (B, N...) at all concentrations, by reactive PLD or co-PLD.
- Interest of thermal heating in situ during PLD.

Review: Bleu et al, Frontier in Chemistry, 2018

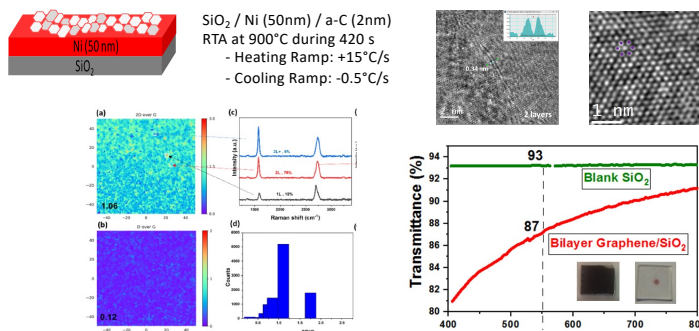
2. Growth mechanism of Graphene from in situ XPS during thermal heating



A three-step process, depending on the dissolved carbon concentration in Ni:

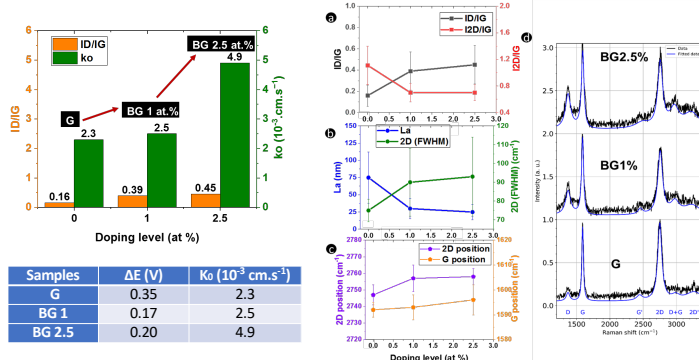
- Step 1: Islands of monolayer epitaxial graphene; $C_{diss} < 2$ at.%
- Step 2: Additional dissolved C promotes progressive growth of bilayer
- Step 3: Formation of weakly interacting few-layer graphene, $C_{diss} > 4$ at.%

3. Transfer-free undoped graphene via PLD and RTP



- Dominant bilayer graphene with few defects at 900°C
- Graphene domains of 130 nm deduced from D/G
- Transmittance of 87% @ 550 nm, corresponding to 2 layers on SiO₂.

4. B-G produced by PLD for the first time



- Boron doping increases defect level and affects the graphene layer formation.
- Upshifts of G and 2D peaks with the boron doping.
- Higher electrochemical reversibility ΔE of graphene with B incorporation
- Higher kinetic electronic transfer compared to the undoped graphene
- The electron transfer capability k_0 is dependent on the boron concentration

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REFERENCES

BLEU et al. in:

- Part 1: Frontier in Chemistry 6 (2018) 572 (Review)
Part 2: Carbon 155 (2019) 410
Part 3: Mat. Chem. & Phys. 238 (2019) 121905
J. Raman Spectroscopy 50 (2019) 1630
Appl. Surf. Sci. 555 (2021) 149492

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- Part 4: Appl. Surf. Sci. 513 (2020) 145843
Diam. Rel. Materials 115 (2021) 108382