

## High yield hydrogen production enabled by macroporous silicon monoliths

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### Abstract

A macroporous silicon (MPS) membrane of 210  $\mu\text{m}$  thickness was functionalized and evaluated for hydrogen production by steam reforming of various fuels. Performance is comparable to conventional reactors, but in a much compact device thanks to process intensification [1]. No blockage or damage was observed after extensive testing. MPS technology [2] is well suited for applications as demonstrated in [3], as it allows the fabrication of large arrays of regular pores in a silicon substrate. The fabricated samples consist of an array of ordered pores of 3.3  $\mu\text{m}$  in diameter and 210  $\mu\text{m}$  length, with a straight profile; the array periodicity is 4  $\mu\text{m}$ . The as-etched sample was post-processed to create an open membrane, and later, it was conformally covered with a  $\text{CeO}_2$  film, which was then coated with a layer of RhPd nanoparticles. Different fuels were studied for hydrogen generation at high temperatures (up to 1023 K): ethanol, propanol, acetone, acetic acid, 2-methoxyethanol, and a diesel surrogate [4]. Total tests duration was 80 h. The results show that no structural damage nor channel blockage is present. Results of  $\text{H}_2$  production are shown in **Fig. 1**. The reaction hydrogen yield and selectivity show the best results for 2-methoxyethanol at 923 K, with 53% selectivity,  $\theta_{\text{H}_2} = 0.4$  yield, and a  $\text{H}_2$  production density of  $110 \text{ L}_\text{N} \text{ H}_2 / \text{mL}_{\text{fuel,liq.}} \text{ cm}^3_{\text{react}}$ . Comparable results in hydrogen production with respect to conventional ceramic honeycombs have been obtained, but in a much smaller volume. Good performance was obtained and no damage or blockage was observed after extended testing, making this technology an outstanding candidate for energy production in embedded applications.

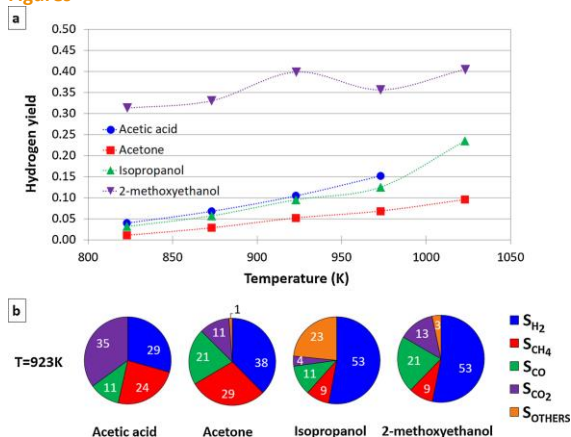
### Acknowledgements

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### References

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### Figures



**Figure 1:** (a) Hydrogen yield and (b) selectivity at 923 K for different tested fuels.