

Development and Mechanistic Study of Mn(CO)₃ Single Sites in 2D-Covalent Organic Frameworks for Electrocatalytic CO₂ Reduction

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Global warming, climate change and our over-dependence on non renewable fossil fuels demand longterm solutions to reduce CO₂ emissions and develop sustainable energy technologies. The electrochemical CO₂ reduction has the potential to accomplish a "carbon-neutral energy cycle", which incorporates CO₂ as the unlimited carbon source for the production of high-density fuels. In this regard, this work entails a fundamental understanding of the CO2 mechanisms using Mn Single Atom Catalyst (SAC) within Covalent Organic Frameworks (COFs). The catalytic activity of the materials was benchmarked against other molecular supported catalysts reported in the literature. Compared to equivalent Mn derivates, COFs exhibited higher selectivity and activity towards CO₂ reduction. Additionally, mechanistic studies based on in situ / in operando spectroelectrochemical techniques together with DFT calculations were used to detect key catalytic intermediates and correlate the catalytic activity with the mechanical constraints impose to the $\{Mn(CO)_3\}$ active sites by reticular framework. Of particular note is the detection of a radical intermediate within a Mn based COFs avoiding the detrimental formation of a dimeric specie determined as a resting state in the catalytic cycle. In addition, a variation of the bipyridyl linker within the COF, by introducing larger phenanthroline ligand, tuned the structural and electronic properties of the catalyst, consequently influencing on the catalytic CO₂ reduction activity. The presence of both pyridyl and phenanthroline within the COFs was found to increase the activity respect the equivalent molecular catalysts. However, structural parameters like the crystallinity and porosity of the reticular frameworks proved to play an important role in the stability of the catalysts. Overall, {Mn(CO)₃}-COFs exhibited higher catalytic activity than the equivalent molecular complexes; thereby suggesting the directions for developing COFs as ligands for CO₂ reduction.

References

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Figures

