## Exploring the photo-catalytic properties of Nb<sub>2</sub>O<sub>5</sub> nanoparticles synthesized by different procedures

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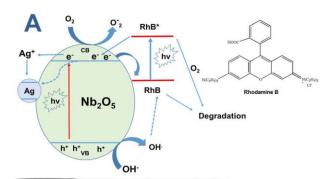
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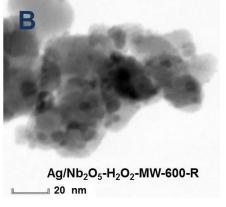
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## **Abstract**

Niobium-containing materials have been prominent in recent decades because of their special applications in the high technology industries, especially in the aerospace, metal super alloy and electroelectronics sectors. The properties of Nb<sub>2</sub>O<sub>5</sub> strongly depends on its synthesis procedure as well as on the conditions of ulterior thermal treatment [1]. We report the synthesis of Nb<sub>2</sub>O<sub>5</sub> powders prepared by the sol-gel precipitation method using niobium(V) ethoxide as precursor. Two chemical routes were chosen: presence of tryethyl amine (TEA) precipitant/template agent, or the oxidant peroxide method. Microwave-assisted activation (MW) was also used. Portions of Nb<sub>2</sub>O<sub>5</sub> powders synthesized by the above procedures were amorphous. Structural changes upon heating from temperature up to 800°C were investigated by XRD technique combined with thermogravimetric analysis. The sequential thermal treatment up to 800°C promotes the crystallization of hexagonal phase orthorhombic phase whereas the ulterior cooling to room temperature lead to a mixture of both phases. Samples calcined at selected temperatures of either 600°C or 800°C for 2 h, were characterized by a wide variety of techniques. The synthetic procedure as well as the combined MWactivation followed by ulterior thermal treatment lead to changes not only on particle size but also on the textural properties of the catalysts. The photoactivity of the synthesized catalysts has been evaluated using Rhodamine B (RhB) as a substrate, under both UV and visible lighting conditions. None of the catalysts synthesized showed activity in the visible. Under UV illumination, some of the catalysts exhibited a relatively low photo-activity in degradation of RhB, which associated with a photosensitizing effect. However, the addition of Aa+ considerably increased the activity of all the catalysts implementing considerably degradation of RhB under illumination leading to silver а new metallized sample.





**Figure 1: (A)** A proposal of the degradation mechanism of RhB in the presence of  $Nb_2O_5$  under UV-lighting conditions. **(B)** Selected TEM for the indicated sample.

## References

[1] A.M. Raba, J. Barba-Ortega, M.R. Joya, The effect of the preparation method of Nb<sub>2</sub>O<sub>5</sub> oxide influences the performance of the photocatalytic activity, Appl. Phys. A Mater. Sci. Process. 119 (2015) 923–928.