

Ionogel-based electrodes for non-flammable high-temperature operating electrochemical double layer capacitors

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Electrochemical double-layer capacitors (EDLCs) have garnered significant attention as suitable energy storage solution for a wide range of applications ranging from automotive, to smart grids, wind turbines, and power electronics[1]. However, their limited energy density and narrow operating temperature range pose challenges for wider adoption, especially in harsh environments [2, 3]. This study introduces a novel approach based on ionogels-type electrodes to enhance the energy density of EDLCs, while enlarging their operating temperature window. Ionogel-type electrodes were fabricated by depositing electrode material slurries based on ionic liquid as the liquid media, eliminating the use of organic solvents and costly drying processes [4]. The designed ionogel-type electrodes demonstrate exceptional electrolyte accessibility demonstrated by high energy density (up to more than 40 Wh kg⁻¹) and simplifying assembly processes in industrial manufacturing lines, while improving device performance compared to conventional EDLC. Electrochemical characterization of ionogel-type EDLCs exhibit a maximum power density of 479.6 kW/kg at a rated voltage of 2.7 V, remarking the superior rate capability of the ionogel-type EDLCs compared to that of conventional EDLCs (139.3 kW/kg maximum power density) [5]. At the power density of 12.96 kW kg⁻¹, the ionogel-type EDLC retains ~83% of energy density measured at the low power of 0.14 kW kg⁻¹. Contrarily, the energy density of conventional EDLCs drops by 65.4% with increasing power density from 0.14 kW kg⁻¹ (29.2 Wh kg⁻¹) to 8.67 kW kg⁻¹ (10.1 Wh kg⁻¹). Moreover, ionogel-type EDLCs demonstrate electrochemical stability and performance even up to 180°C, surpassing the limitations of traditional EDLCs based on organic electrolytes. Overall, this research presents a promising pathway for advancing EDLC technology through innovative electrode fabrication methods and IL-based electrolytes.

References

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Figures

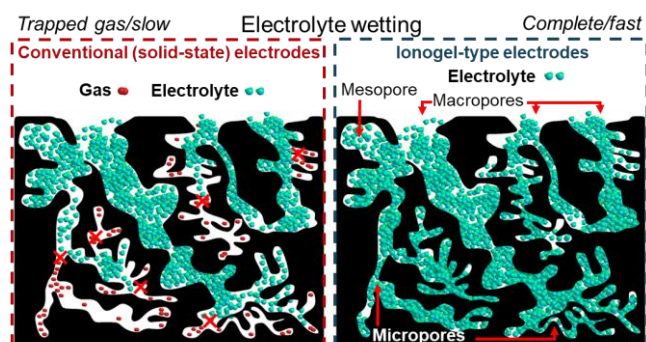


Figure 1. Sketch of the electrolyte wetting processes for conventional (solid-state) and ionogel-type electrodes.

This project received funding from the European Union's GREENCAP Horizon Europe research and innovation program under Grant Agreement No.101091572 and 2D-PRINTABLE Horizon Europe research and innovation program under Grant Agreement No.694101.