Core Shell Ternary Nanostructures based on Mo(W)S₂ For Hydrogen Evolution Reaction

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Water splitting, a vital process for clean energy production via hydrogen, relies on efficient catalysts. Transition metal dichalcogenides (TMDs), notably MoS₂ and WS₂, have garnered significant interest for their catalytic potential. Specifically, core@shell nanostructures like Au@MoS₂ have demonstrated enhanced activity in the hydrogen evolution reaction (HER) [1-2]. This work delves into the influence of W-based heteroatoms on HER performance within Au@MoS2 nanostructures. Synthesis involved the sodium borohydride reduction method, followed by thorough characterization using XPS and aberration TEM microscopy. Electrochemical analyses, including linear sweep voltammetry, cyclic voltammetry and impedance spectroscopy, were conducted to assess HER efficiency.

The STEM images (**Fig. 1**) vividly illustrate the core@shell structure, with a single $Mo(W)S_2$ layer enveloping the metallic Au core, indicating excellent crystallinity. Notably, tungsten presence was confirmed through both XPS and EELS analysis. The electrochemical data further underscore the efficacy of these hybrid structures, revealing reduced overpotential and Tafel slope compared to counterparts lacking W-based heteroatoms. Remarkably, the Au@Mo(W)S₂ sample containing a 3% molar ratio of the tungsten precursor, exhibited superior performance (**Fig. 2**).

These findings signify the potential for tailored synthesis strategies to fine-tune catalyst properties, thereby enhancing HER performance. The integration of W-based heteroatoms offers a promising avenue for advancing the efficiency of Au@MoS₂ nanostructures in hydrogen production applications, contributing to the quest for sustainable energy solutions.

References

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J.J. Quintana Gonzalez et al., Int. J. Hydrogen Energy. 51, 371-382 (2024)
Figures

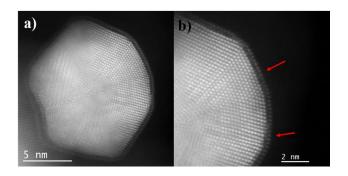


Figure 1: HAADF-STEM images of the $Au@Mo(W)S_2$ sample. The arrows highlight the presence of the $Mo(W)S_2$ layer.

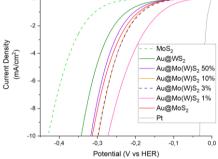


Figure 2: Polarization curves of the core@shell samples with different proportions of dichalcogenides.

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