

Electrochemical Functionalization of 2H-Phase MoS₂ nanosheets towards Enhanced Catalytic Applications

A. Martínez-Jódar*

S. García-Dalí; J.I. Paredes; S. Villar-Rodil; A. Martínez-Alonso; J.M.D. Tascón.

Instituto de Ciencia y Tecnología del Carbono, INCAR-CSIC, 33011, c/ Francisco Pintado Fe, 26 Oviedo, Spain.

alberto.mj@incarcsic.es

Abstract

We describe a novel electrochemical strategy for the surface functionalization of 2H-phase MoS₂ nanosheets, which affords the grafting of different molecular groups (e.g., acetic acid groups) derived from organoiodides [1]. Upon cathodic treatment of a pre-expanded MoS₂ crystal in an organoiodide containing electrolyte, water-dispersible nanosheets with a derivatization degree of ~0.10 molecular groups per surface sulfur atom were obtained, expanding the scope of covalent molecular functionalization of two-dimensional MoS₂, typically restricted to the metastable 1T phase [2]. Functionalization is shown to be driven by the external supply of electrons to the MoS₂ nanosheets and to be controlled by the presence of S vacancies in the 2H-MoS₂ lattice, where the molecular groups can bind. The functionalized nanosheets were tested as a catalyst for the reduction of nitroarenes and organic dyes with NaBH₄, which is relevant in environmental remediation and chemical synthesis [3]. These derivatized 2H-MoS₂ nanosheets showed a remarkably enhanced catalytic activity compared to that of non-functionalized 1T- and 2H-phase MoS₂ nanosheets as well as non-noble metal catalysts previously reported for this application, and retained a good catalytic activity even at realistic reactant concentrations. The 2H-phase MoS₂ catalyst could also be immobilized on a polymeric scaffold in order to facilitate its manipulation and reutilization for several catalytic cycles. Analysis of the reduction kinetics revealed a correlation between the reaction order and the net electric charge of the organic substrates.

References

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Figures

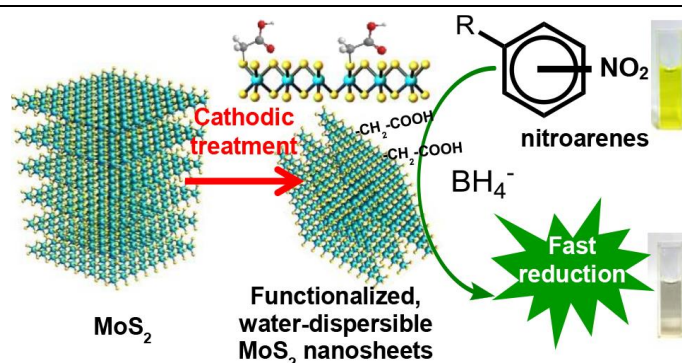


Figure 1: Scheme of the electrochemical cathodic strategy used to obtain derivatized 2H-MoS₂ dispersions and its application to the reduction reaction of several nitroarenes and organic dyes.