

CVD synthesis of few-layered h -BC₂N and optical applications

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Arranging carbon, boron, and nitrogen atoms in a sp^2 network can give rise to tailored electronic properties from insulators (h -BN) to metals (graphene)[1,2]. For semiconductor applications, the construction of a ternary structure (h -B_xC_yN_z) is highly desirable, but its uniform and large-area synthesis has remained a great hurdle. This challenge has been attempted by a chemical vapor deposition method with a single molecular precursor, n -tri-methyl borazine where boron, carbon, and nitrogen atoms are covalently bonded, onto Ni catalysts in conjunction with the quenching method after the synthesis. The atomic structure closely resembles h -BC₂N as presented by XPS and nanometer resolution EELS mapping, and the photoluminescence and electroluminescence observed from the h -BC₂N film were in agreement, proving its well-established bandgap of 2.15 eV [3]. As optical application, the utilization of h -BC₂N film for 2D light emitting diodes was demonstrated. Though films might have impurities such as small h -BN fragments and h -B_xC_yN_z other than h -BC₂N phase, we believe that this work provide a starting point of controlling the ternary BCN compounds that retain sp^2 hybridized chemical bonds.

References

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Figures

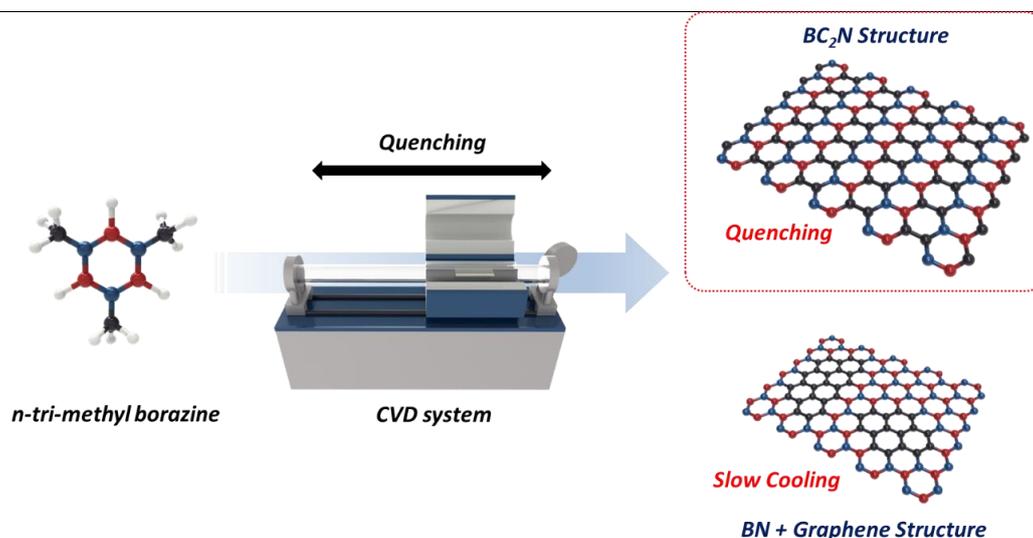


Figure 1: Schematic diagram of h -BC₂N synthesis.