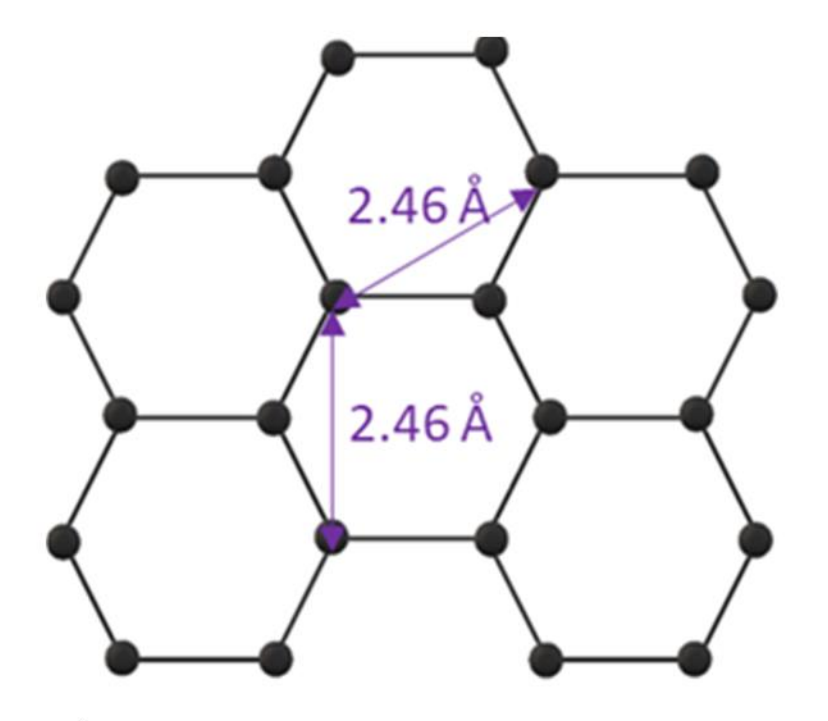


Two-steps CVD synthesis of Boron Nitride Islands on polycrystalline nickel substrate

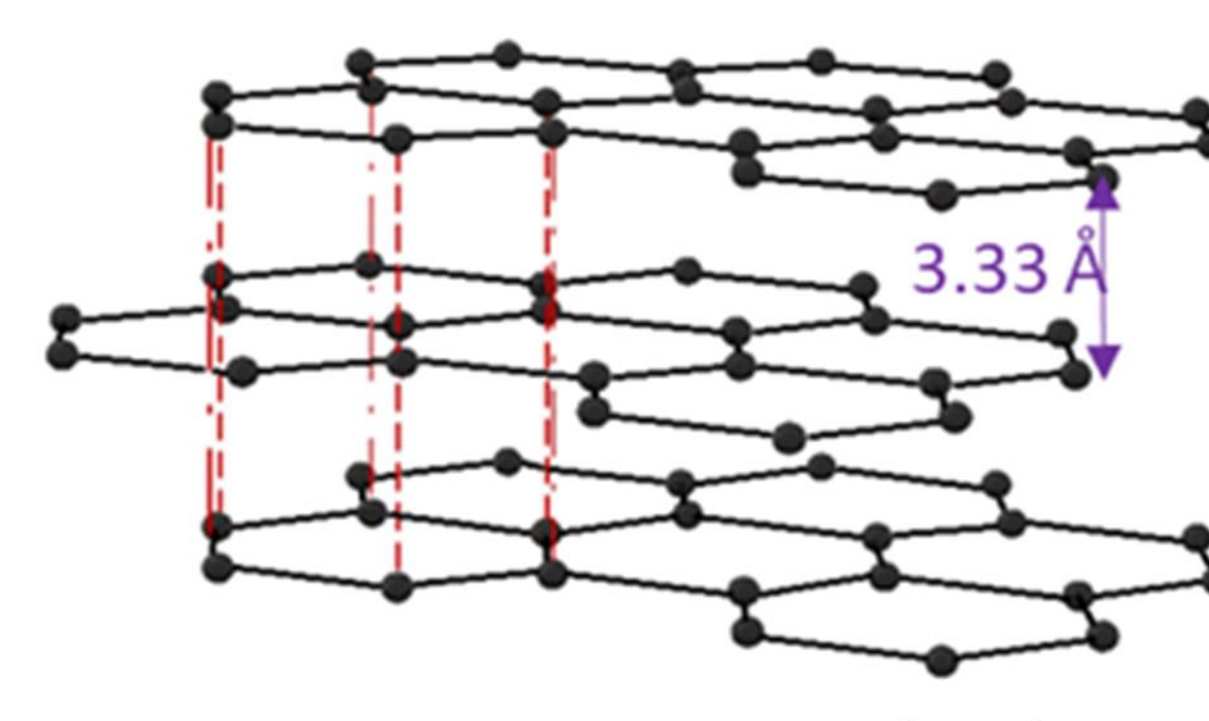
Tailpied Laure⁽¹⁾, Andrieux-Ledier Amandine⁽²⁾, Mérot Jean-Sébastien⁽¹⁾, Fossard Frédéric⁽¹⁾, Decams Jean-Manuel⁽³⁾, Loiseau Annick⁽¹⁾
(1) DMAS/LEM, (2) DPHY/CMT (3) Annealsys

Graphene and sp²-hybridized boron nitride (BN) : unique and complementary 2D materials

Graphene



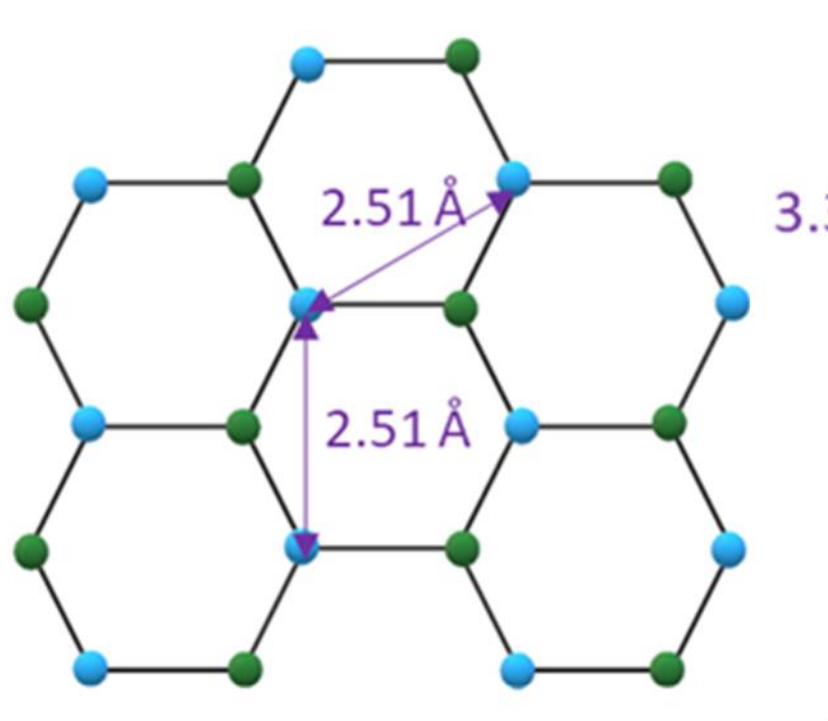
2.46 Å



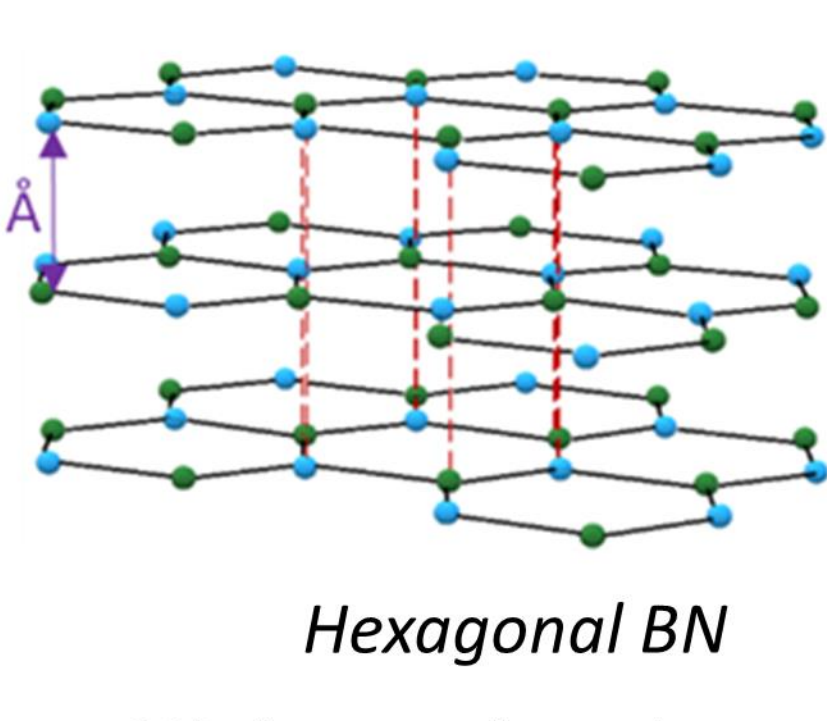
3.33 Å

- ✓ Excellent conductor : $\mu_{\text{Graphene}} = 200\,000\text{ cm}\cdot\text{V}^{-1}\cdot\text{s}^{-1}$ (102x μ_{Silicium})
- ✓ Record thermal conductivity : $\lambda = 5000\text{ W/mK}$

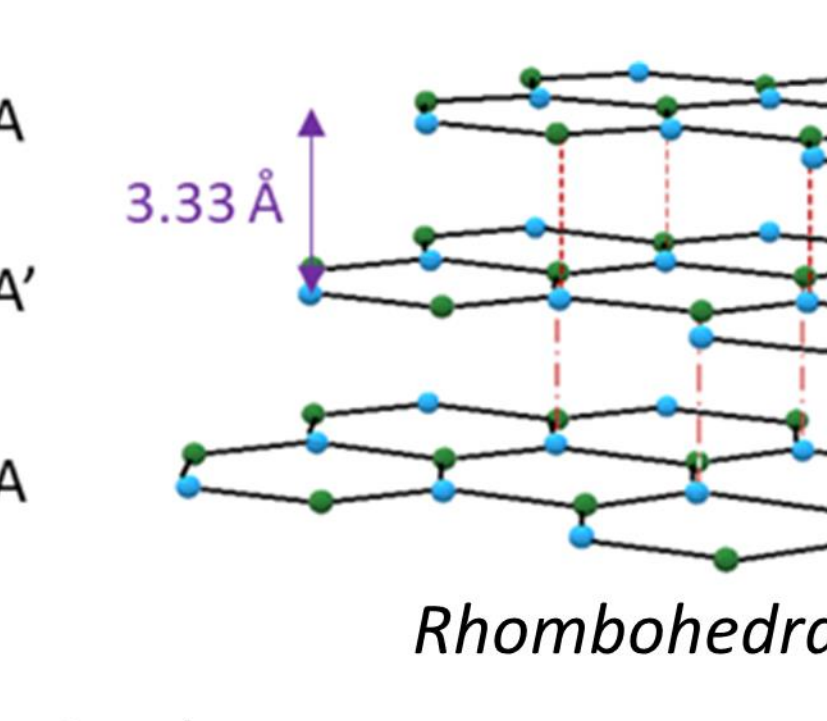
sp²-hybridized BN



2.51 Å



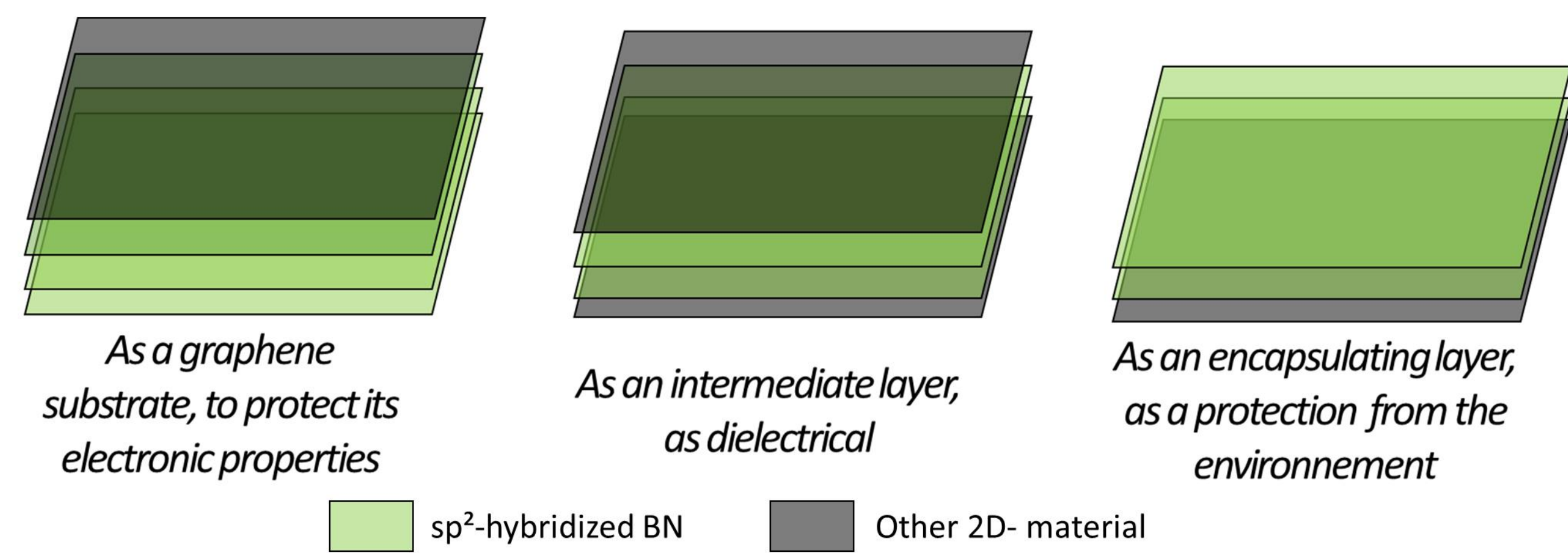
3.33 Å



3.33 Å

- ✓ High gap insulator (~6eV)
- ✓ Chemically and thermally stable

Role of BN in the Van der Waals heterostructures



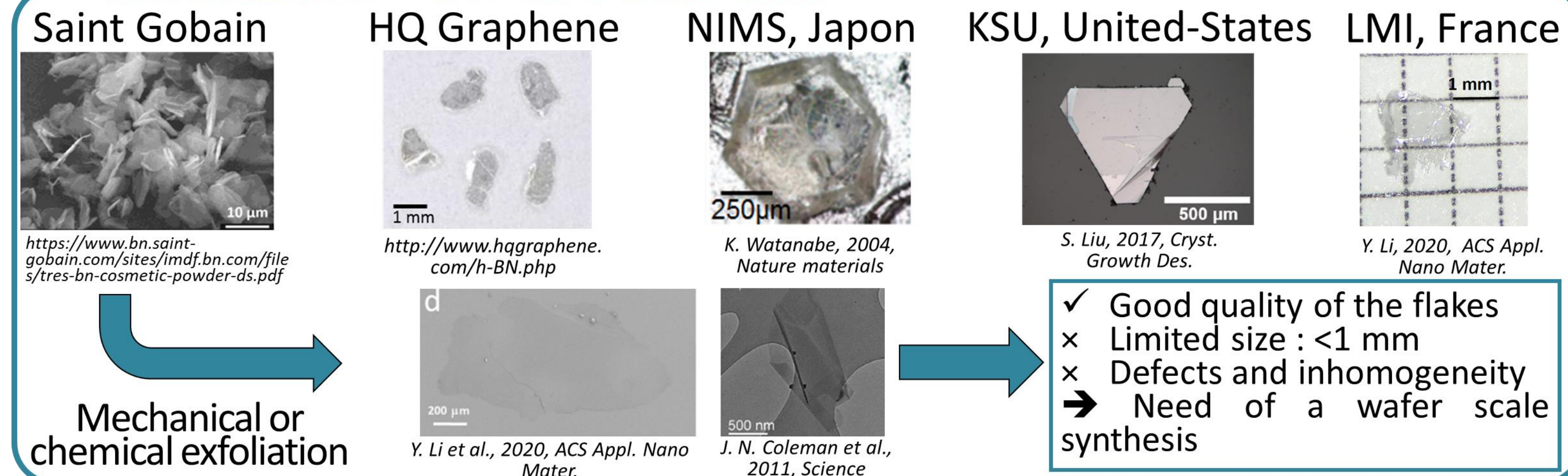
As a graphene substrate, to protect its electronic properties

As an intermediate layer, as dielectrical

As an encapsulating layer, as a protection from the environment

sp²-hybridized BN Other 2D- material

Limitation: Few BN sources



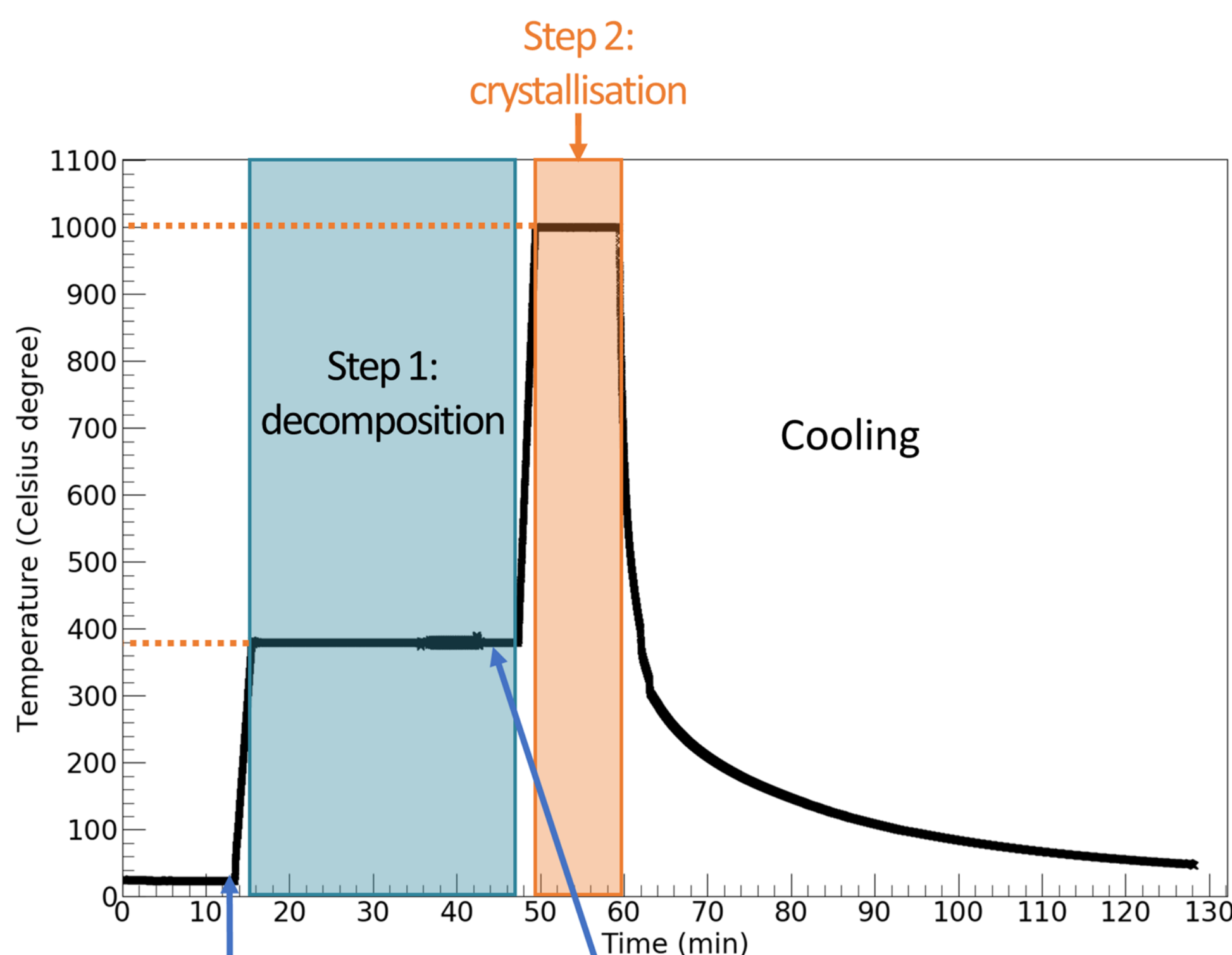
Saint Gobain HQ Graphene NIMS, Japon KSU, United-States LMI, France

Mechanical or chemical exfoliation

- ✓ Good quality of the flakes
- ✗ Limited size : <1 mm
- ✗ Defects and inhomogeneity
- ➔ Need of a wafer scale synthesis

BN synthesis

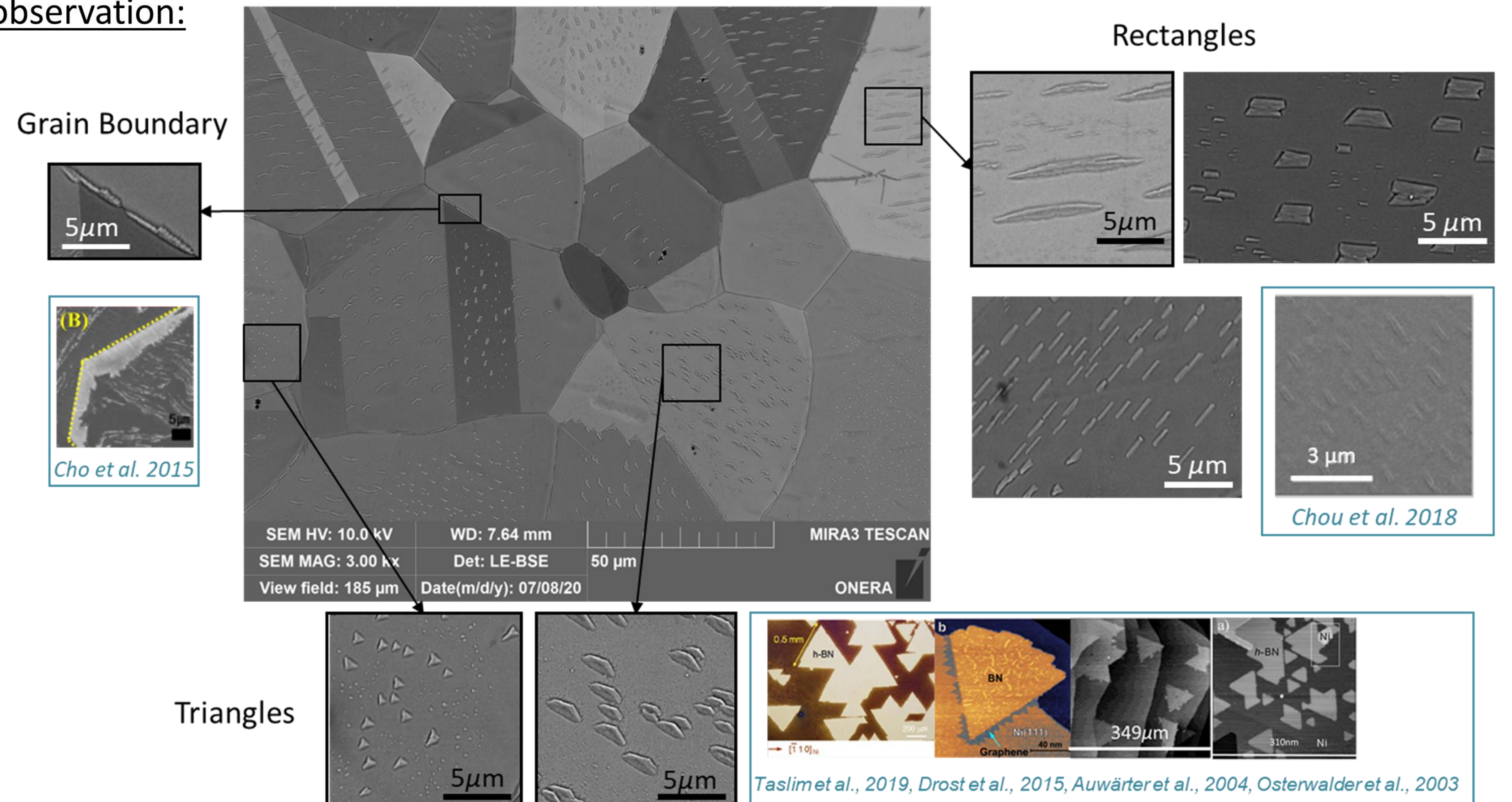
Set up: Low pressure chemical vapor deposition (LPCVD)-AS ONE Annealsys
Substrate : polycrystalline Ni, 99.99%, thickness=25 μm
Precursor : Borazine (B₃N₃H₆)
Process: Borazine reacts strongly under the halogens lamps used to heat the chamber ➔ two steps process used: Decomposition of the borazine at 380°C (Step 1) and crystallisation of the BN at 1000°C (Step 2) (Y. Shi, 2010, Nanoletters). The last step (step 3) is a fast cooling by extinction of the lamps under Ar/H₂.



Evolution of the temperature during a classical BN synthesis

Results: BN islands

SEM observation:



Grain Boundary

Rectangles

Triangles

SEM HV: 10.0 kV WD: 7.64 mm MIRA3 TESCAN

SEM MAG: 3.00kx Det: LE-BSE 50 μm

View field: 185 μm Date(m/d/y): 07/08/20 ONERA

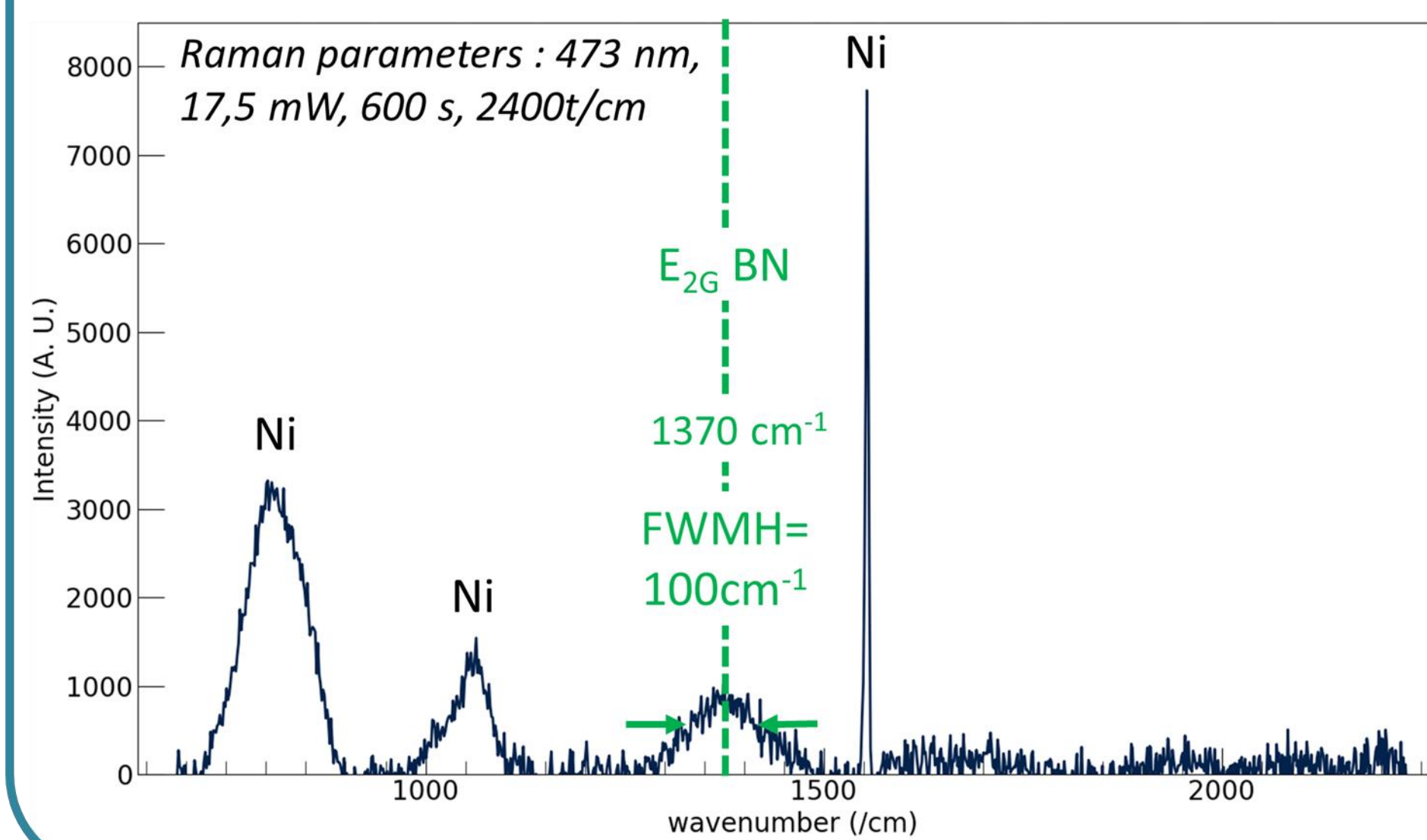
Cho et al. 2015

Chou et al. 2018

Taslim et al., 2019, Drost et al., 2015, Auwärter et al., 2004, Osterwalder et al., 2003

BSE/SEM (backscattered electron scanning electron microscopy) images exhibiting crystallographic contrast of the Ni substrate and related BN deposit

Raman spectra:



- ✓ Micrometer sized BN islands with shapes consistent with the data of the litterature
- ✓ Islands shapes different for each underlying Ni grain orientation (H. Prévost, 2020, 2D materials)
- Very broad peak in Raman spectroscopy ➔ Low crystallinity ?

Conclusion & Perspectives

- ✓ BN islands are obtained from borazine with a two-steps process. Their size and shape change as a function of the orientation of the underlying Ni grain
- To obtain a better crystallinity of the BN, the duration of the crystallization step will be extended
- Other parameters, such as the temperature of the decomposition step and the cooling speed will be modified to obtain a BN continuous film

Acknowledgement

The research leading to these results has received partial funding from the European Union's Horizon 2020 research and innovation program under grant agreement N° 881603 (Graphene Core 3) , ONERA and GDR-IGraphene&Co (for the subscription fees). Authors want to thanks the Annealsys society for the research collaboration. This work has benefited from SEM facilities and staff expertise of Laboratoire Castaing (DMAS/ONERA)

CONTACT PERSON

Laure Tailpied
 laure.tailpied-ext@onera.fr
 PHD student at Laboratoire d'Etude des Microstructures (LEM)
 UMR 104 Onera-Cnrs

REFERENCES

<https://www.bn.saint-gobain.com/sites/imdf.bn.com/files/tres-bn-cosmetic-powder-ds.pdf>

<http://www.hqgraphene.com/h-BN.php>

K. Watanabe et al., 2004, Nature materials

S. Liu et al., 2017, Cryst. Growth Des.

Y. Li et al., 2020, ACS Appl. Nano Mater.

J. N. Coleman et al., 2011, Science

A. Taslim et al., 2019, Nanoscale

H. Cho et al., 2015, Sci. Rep.

H. Chou et al. 2018, ACS Appl. Mater. Inter.

R. Drost et al., 2015, Sci. Rep.

W. Auwärter et al., 2004, Chem. Mater.

J. Osterwalder et al., 2003, e-J. Surf. Sci. Nanotech.

Y. Shi et al. 2010, Nano Lett.

H. Prévost et al. 2020, 2D Mater.