



Introduction

Growth of graphene using the CVD method (so-called CVD graphene) on mono- and polycrystalline copper surfaces give the possibility to obtain large-area graphene coatings [1]. The CVD graphene can be also transferred onto other metallic or non-metallic substrates. The results of many investigations indicate that graphene can provide a barrier substantially reduce the corrosion rate (e.g.[2,3]), however, the tightness of real graphene coating is a fundamental feature of barrier protection [1-4]. The experimental investigations of barrier properties of graphene layers included the development of methodical bases for characterization of obtained layers (optical microscopy, µ-Raman Spectroscopy and AFM).

Different techniques to obtain graphene films (examples)







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Mechanical exfoliation (Si/SiO₂)

Epitaxial growth (SiC)



HSMG - (transferred on the Si/SiO2 (grown on liquid metal CuNi)

CVD growth (Cu)

The barier properties of CVD graphene



cm-1 – large distance between defects (slightly	TYP	'E OF DEF
defected structure, Ld ≥ 12nm);	•	3.5 – grair
o $I(D)/I(G) \ge 1$: small distance between	•	7 – vaca
defects (defected structure, 1.5nm <ld <12nm);<="" td=""><td>•</td><td>13 – sp3</td></ld>	•	13 – sp3
o I(D)/I(G) <1 and FWHM of G band > 32		-
cm-1): very small distance between defects		
(strongly defected structure, Ld \leq 1.5 nm).		





E 0.0



Graphene layer on copper substrate (borate buffer, passive layer Cu₂O/B₂O₃) (pH=8, E=150 mV/SHE, E1)

Conclusions :

The obtained results indicate a strong influence of defects in CVD graphene materials (types, density), an important role of substrate preparation (topography, roughness, oxidation) and corrosion products as well as an influence of electrodeposition conditions on the barrier properties of graphene layers.

The damages of graphene coatings by copper substrate and improper electrodeposition conditions can be analysed by μ -Raman Spectroscopy (μ -RS), Atomic Force Microscopy (AFM) and Optical Microscopy (OM).

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E-pH, Cu-H₂O

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REFERENCES

- S. Maarof, A.A. Ali, A.M. Hashim, Nanoscale Research Letters, 14 (2019) 143. [1]
- N.T. Kirkland, T. Schiller, N. Medhekar, N. Birbilis, Corrosion Science, 56 (2012) 1. [2]
- K.S. Aneja, S. Bohm, A.S. Khanna, H.L.M. Bohm, Nanoscale, 42 (2015) 17879. [3]
- [4] S.P. Damari, L. Cullari, D. Laredo, R. Nadiv, E. Ruse, R. Sripada, O. Regev, Progress in Organic Coatings, 136 (2019) 105207.

