

Cu₂Si : a promising 2D Dirac nodal line semimetal

Mathis Cameau^{1,2}, R. Yukawa³, C.-H. Chen⁴, A. Huang⁴, P. Moras⁵, P. M. Sheverdyeva⁵, M. Silló⁶, M. Hervé², S. Ito⁷, R. Ishibiki⁸, K. Horiba³, Y. Obata³, T. Kondo^{8,9}, H. Kumigashira³, H.-T. Jeng^{4,10,11}, I. Matsuda¹², and M. D'angelo²

1 Institut de Minéralogie, de Physique des Matériaux et de Cosmochimie, Sorbonne Université, France - **2** Institut des NanoSciences de Paris, Sorbonne Université, France - **3** Institute of Materials Structure Science, High Energy Accelerator Research Organization (KEK), Tsukuba, Japan - **4** Department of Physics, National Tsing Hua University, Taiwan - **5** Istituto di Struttura della Materia-CNR (ISM-CNR), Italy. **6** Synchrotron-SOLEIL, France - **7** Fachbereich Physik, Philipps-Universität Marburg, Germany - **8** Faculty of Pure and Applied Sciences, University of Tsukuba, Japan - **9** Materials Research Center for Element Strategy, Tokyo Institute of Technology, Japan - **10** Physics Division, National Center for Theoretical Sciences, Taiwan - **11** Institute of Physics, Academia Sinica, Taiwan - **12** Institute for Solid State Physics, University of Tokyo, Japan.

Contact e-mail : mathis.cameau@sorbonne-universite.fr

With the discovery of Dirac nodal-line (DNL) semimetals, the notion of the Dirac point has been extended to lines and loops in momentum space. As for Dirac cones, the energy dispersion of DNL is linear around the band crossing.

The first DNL two-dimensional material evidenced is Cu₂Si [1], which is composed of a honeycomb Cu lattice and a triangular Si lattice, both coplanar (see Figure 1, left). When prepared on a Cu(111) substrate, angle-resolved photoemission measurements revealed two concentric DNL close to the Fermi level and centred on the high-symmetry point Γ , in agreement with first principle calculations. These nodal lines are furthermore protected by inversion and time reversal symmetries, as well as a mirror crystalline symmetry. In this talk, we will focus on the experimental realization of Cu₂Si, address the challenges to overcome for devices applications, and discuss more specifically the changes in the electronic structure that can be induced by the choice of substrate and by induced spin-orbit coupling. We investigated the use of a Si(111) semiconductor substrate, more relevant for electronic devices, and showed that the electronic structure was substantially modified due to interactions with the substrate [2]. We also present our ongoing work aiming at observing the effect of a strong spin-orbit coupling induced by heavy element deposition (Pb) on a Cu₂Si monolayer, which would lead to a lift of degeneracy of the electronic bands (see Figure 1, right) and to a possible gap opening, as it has already been shown on graphene [3].

References

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- [2] M. Cameau et al., Phys. Rev. Materials 3, 4, 044004 (2019)
- [3] I. I. Klimovskikh et al., ACS Nano 11, 368 (2017)

Figures

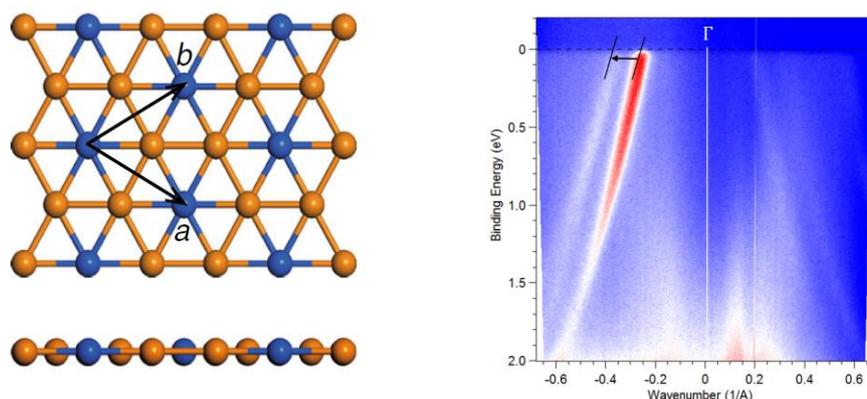


Figure 1: Left: Atomic structure of Cu₂Si, top and side view. Cu atoms are in orange, Si atoms in blue. **Right:** ARPES spectra at 30eV along Γ -M direction of Cu₂Si/Cu(111) after Pb deposition. The outermost band results from the splitting of the main one (in red), due to Pb interaction.