

Liquid-phase exfoliated GeSe nanoflakes for photoelectrochemical-type photodetectors and photoelectrochemical water splitting

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The conversion of electromagnetic radiation into chemical fuels and electricity through photoelectrochemical (PEC) cells represents a powerful strategy for sustainable fuel and chemical generation,[1] as well as innovative self-powered photodetectors.[2] In this context, two-dimensional (2D) materials are attracting huge interest as potential advanced photo(electro)catalysts.[3] Recently, low-cost and environmentally friendly 2D group-IVA metal monochalcogenides have been theoretically predicted to be water splitting photocatalysts.[4] Among them, layered germanium selenide (GeSe), is a promising material candidate for optoelectronic devices due to its tuneable electronic structure, strong visible-light absorbance, photoresponse and environmental stability.[4] Here, we report the first experimental study of the photo(electro)catalytic activity of single-/few-layer GeSe flakes in different aqueous media, ranging from acidic to alkaline solutions: 0.5 M H₂SO₄ (pH 0.3), 1 M KCl (pH 6.5), 1 M KOH (pH 14). The PEC properties of the GeSe nanoflakes, produced by liquid-phase exfoliation approach[5] in non-toxic solvents (e.g., 2-propanol) and deposited by spray-coating technique,[5] are used to conceive PEC-type photodetectors, reaching responsivity up to 0.32 AW⁻¹ under 455 nm. Finally, we demonstrate the use of GeSe-based photoelectrodes as photoanodes or photocathodes for water splitting reactions under simulated sunlight, inspiring the use of 2D GeSe in innovative PEC systems.

References

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Figure

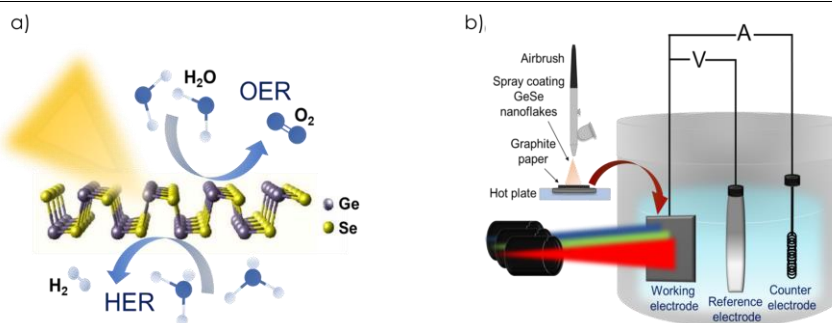


Figure 1: a) Schematic processes of photoelectrochemical water splitting on GeSe nanoflakes; b) Illustration of the experimental setup for photoelectrochemical characterization of the GeSe photoelectrodes, produced by spray coating deposition of the GeSe nanoflakes dispersion.