Graphene/ Mo₂C vertical heterostructures: Low temperature chemical vapor deposition growth and electrocatalytic properties on HER and OER.

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Thin 2D Mo₂C/graphene vertical heterostructures have attracted significant attention due to their potential application as electrodes in hydrogen evolution reaction (HER) and energy storage. Common drawback in the chemical vapor deposition synthesis of these structures is the demand for high temperature growth, which should be higher than the melting temperature of the metal catalyst. The most common metallic catalyst is Cu, which has a melting temperature of 1084 °C. Here, we report the growth of thin, ~ 200 nm in thickness, semitransparent micrometer-size Mo₂C domains and Mo₂C/graphene heterostructures at lower temperatures using liquid Sn-Cu alloys. Results demonstrate the capability to grow continuous nanometerthin Mo₂C films at temperatures as low as 880 °C, without sacrificing the growth rate. Mo2C films are proven to be efficient electrocatalyst for hydrogen evolution reaction (HER). Moreover, we demonstrate the beneficial role of araphene overarown on Mo₂C in reducing the HER overpotential values. We evaluate also the efficiency electrocatalytic of Mo₂C/ graphene on oxygen evolution reaction. Finally, we discuss on the enhanced electrocatalytic performance of the heterostructure when we apply a "flipped" transfer method which permits to reverse the vertical order of the heterostructure, allowing Mo₂C to stand on graphene ¹⁻³.

References

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Figure 1: a) OM images of Mo₂C domain nucleation and film growth. After 80 min, the domains reach the coalescence phase to form a continuous film. On the right end, a SEM image of the continuous Mo₂C film is demonstrated. b, C) Electrochemical performance Mo₂C/graphene of rode. Polarization curves and Tafel plots of Mo₂C and Mo₂C/graphene electrodes. d, e) Raman mapping of Mo₂C domain for Raman shift between 130-155 cm⁻¹ and Raman spectrum from zone covered with Mo₂C/araphene (black line) and graphene-only (red line).