## Controlling photoluminescence of Molybdenum disulfide (MoS<sub>2</sub>) by molecular doping

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Atomically thin Molybdenum disulfide has been studied extensively because when the thickness of  $MoS_2$  is reduced to a single layer, the latter no longer exhibits the indirect band gap typical of the bulk, but a 1.8 eV direct band gap and gives rise to strong photoluminescence even at room temperature.<sup>1</sup> Chemical vapour deposition (CVD) has been the most effective method to obtain large domains of single layer MoS<sub>2.</sub><sup>2</sup>. However, during the growth process sulfur vacancies are formed and influence the electronic properties.<sup>3</sup> In this work we studied the photoluminescence (PL) of single layer  $MoS_2$  upon functionalization with thiol-terminated molecules. We demonstrate that the PL intensity increases when p-type doping chemisorb on single layer MoS<sub>2</sub>, while the PL intensity is reduced upon functionalization with an n-type dopant. This PL intensity variation is due to the switching between exciton and trion recombination PL, which depends on the carrier density in the single layer MoS<sub>2</sub>.<sup>4</sup> We confirmed the chemical environment of MoS<sub>2</sub> before and after functionalization using X-ray photoemission spectroscopy (XPS), while Raman spectroscopy was employed to monitor the functionalization via the vibrational modes of MoS<sub>2</sub>. Since the dopant molecules are covalently bonded, the PL intensity is not affected by solvent exposure. This straightforward and effective approach of controlling the PL by molecular doping enables a robust  $MoS_2$  system, which is essential for optoelectronic applications.

## References

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Figures

Figure 1: Schematic illustration of surface charge transfer between MoS<sub>2</sub> and dopant molecules