Reduced Porous Graphene Oxide Based Nanocomposite Materials as High Performance Lithium-ion Battery Electrodes

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Abstract

Reduced graphene oxide based nanocomposites attracted much attention as electrode materials for energy storage devices. To improve the electrochemical performance even further, recent research has focused on the preparation of porous structures. Reduced Porous graphene oxide allows for rapid ion diffusion and displays high real surface area. Here, our recent work in reduced porous graphene oxide based nanomaterials for lithium-ion battery electrodes is reported.

A composite of reduced porous araphene oxide and LiFePO₄ has been developed to improve the speed of charging-discharging, and the cycling stability of lithium ion batteries using LiFePO₄ as a cathode material.[1] Reduced porous graphene oxide has been successfully synthesized via activation by KOH. The as-prepared reduced porous graphene oxide was mixed with LiFePO₄ to prepare the composite. Microscopic observation and nitrogen sorption analysis have revealed the surface morphologies of reduced porous graphene oxide and the reduced porous graphene oxide/LiFePO4 composite. Electrochemical properties have also been investigated after assembling coin cells with the reduced porous araphene oxide/LiFePO₄ composite as a cathode active material. Interestingly, reduced porous graphene oxide/LiFePO₄ composite has exhibited better electrochemical properties than conventional graphene/LiFePO₄ composite as well as bare LiFePO₄, including exceptional speed of charging-discharging

and excellent cycle stability. That is because the reduced porous graphene oxide in the composite provides abundant porous channels for the diffusion of lithium ions. Moreover, it acts as a conducting network for easy charge transfer and as a divider, preventing the aggregation of LiFePO₄ particles. Owing to these properties of reduced porous graphene oxide, LiFePO₄ could demonstrate enhanced and stably long–lasting electrochemical performance.

Although MoS₂ has a high theoretical capacity than graphite as an anode material for Lithium-ion battery, its intrinsic poor electrical/ionic conductivity decreases rate property and lithium storage capacity. In this work, we prepared MoS₂/reduced porous araphene oxide composite by solvothermal method to complement conductivity and improve lithium storage properties of MoS₂.[2] Reduced porous araphene oxide was synthesized by convenient, low-cost, and mass-producible nitric acid treatment method. Due to its enlarged surface area and porous structure, MoS₂/reduced porous araphene oxide sample exhibited improved capacity and cyclic stability than other MoS₂/reduced graphene oxide and MoS₂ nanoparticle samples. After 100 cycles, MoS₂/reduced porous graphene oxide electrode shows improved capacity of 932 mA h g⁻¹ while capacity of MoS₂/reduced graphene oxide electrode is 395 mA h g⁻¹ at 200 mA g⁻¹.

References

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