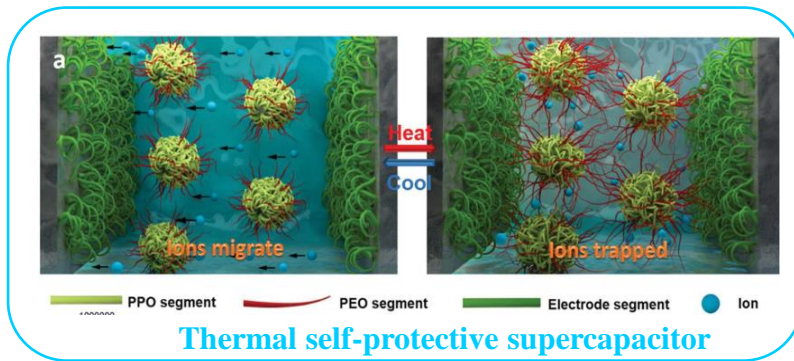
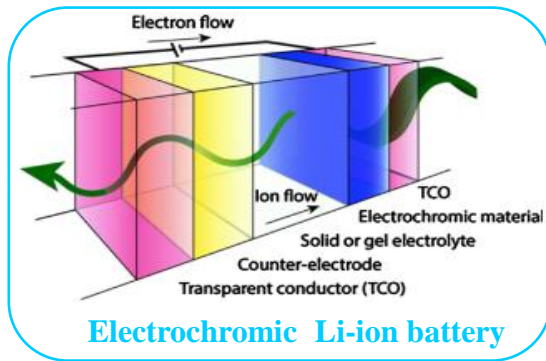
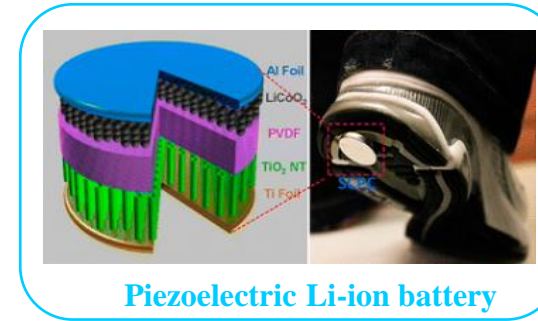
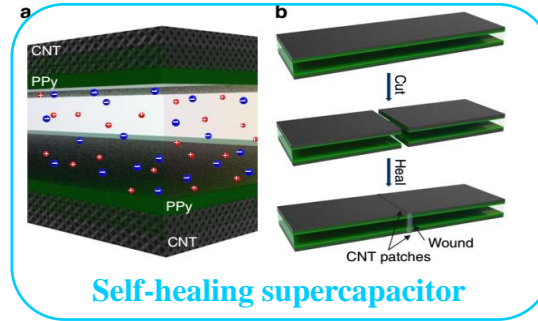
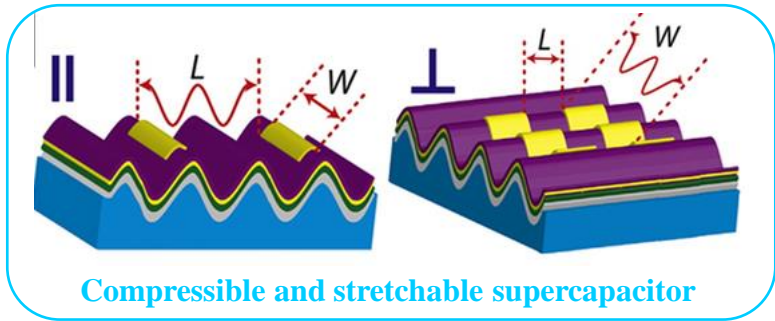


A Dual-Stimuli-Responsive Sodium Bromine Battery with Ultra-High Energy Density

Faxing Wang

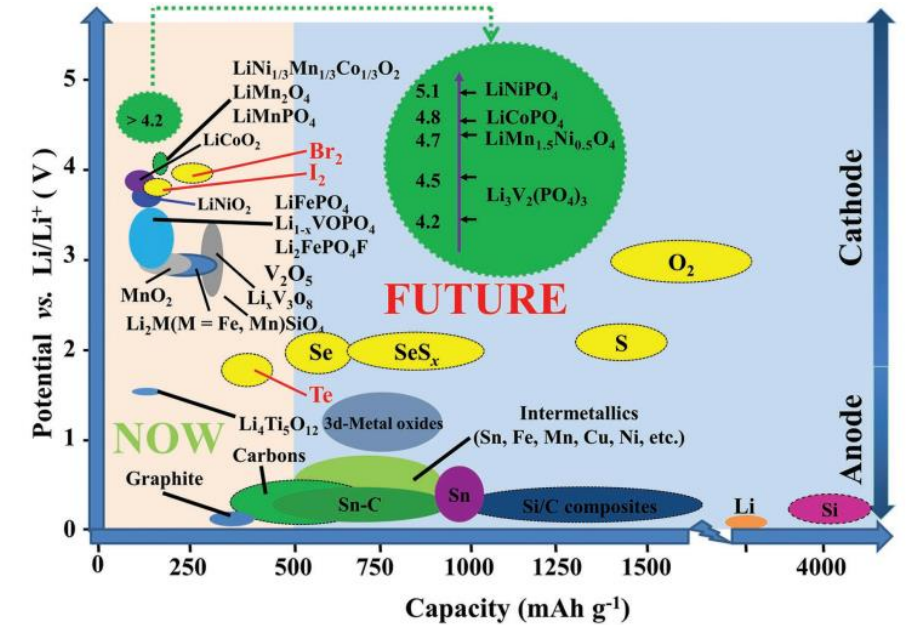
Prof. Xinliang Feng

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Technische Universität Dresden, Germany



Adv. Mater. 2016, 28, 8344–8364. (Review)

Energy storage devices	Electrode materials	Type of stimuli	Energy density	References
Supercapacitor	WO ₃ //WO ₃	Electrochromic	3.6 Wh kg ⁻¹	Angew. Chem. Int. Ed. 2014, 53, 11935.
	PANi@CNT//PANi@CNT	Electrochromic	12.8 Wh kg ⁻¹	Adv. Mater. 2014, 26, 8126.
	CNT//CNT	Self-healing	<10 Wh kg ⁻¹	Adv. Mater. 2014, 26, 3638.
	PANi@CNT//PANi	Self-healing	<30 Wh kg ⁻¹	Angew. Chem. Int. Ed. 2014, 53, 9526.
	PPV@CNT//PPV@CNT	Self-healing	< 30 Wh kg ⁻¹	Nat. Commun. 2015, 6, 10310.
	CNT//CNT	Thermal protection	<10 Wh kg ⁻¹	Adv. Mater. 2015, 27, 5593.
	AC//AC	Thermal protection	<20 Wh kg ⁻¹	Adv. Mater. 2016, 28, 7921.
Rechargeable battery	Li//V ₂ O ₅	Electrochromic	52 Wh kg ⁻¹	Nano Lett. 2012, 12, 1857.
	Al//prussian blue	Electrochromic	80 Wh kg ⁻¹	Nat. Commun. 2014, 5, 4921.
	LiTi ₂ (PO ₄) ₃ @CNT//LiMn ₂ O ₄ @CNT	Self-healing	32 Wh kg ⁻¹	Angew. Chem. Int. Ed. 2016, 55, 14384.
	Zn//LiMn ₂ O ₄	Cooling- Recovery	130 Wh kg ⁻¹	Angew. Chem. Int. Ed. 2017, 56, 7871.
	Cu//prussian blue	Thermal self-charge	30 Wh kg ⁻¹	Nat. Commun. 2014, 5, 3942.
Li ₄ Ti ₅ O ₁₂ //LiFePO ₄	Photo self-charge	230 Wh kg ⁻¹	Nat. Commun. 2015, 6, 8103.	



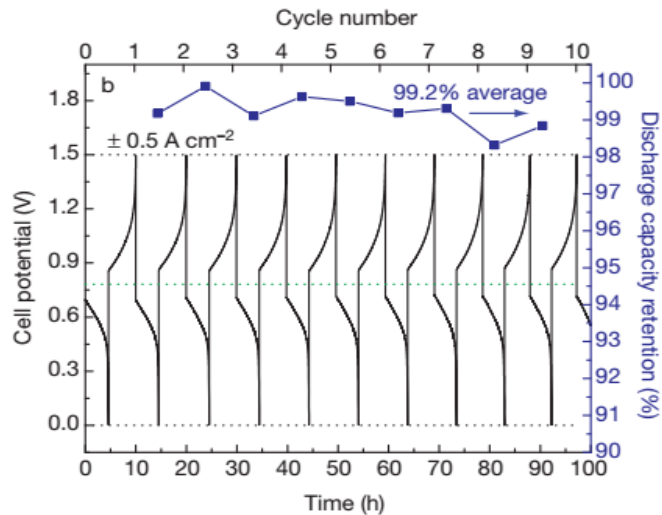
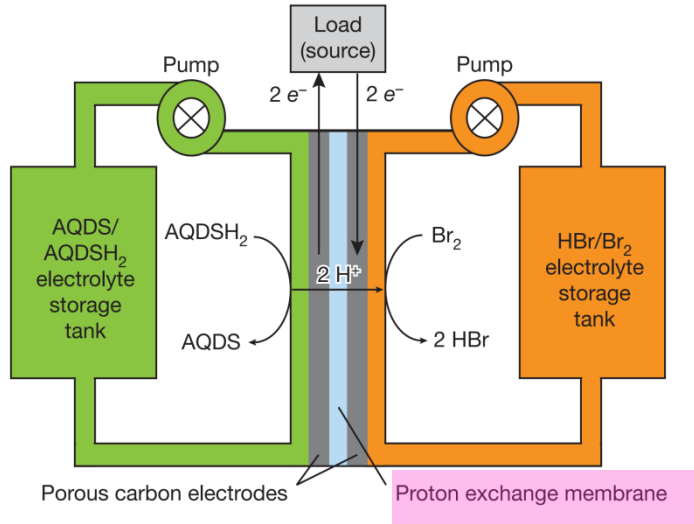
Theoretically Na⁺/Na: 1200 mAh g⁻¹ (0.26)
 Br⁻/Br₂: 310 mAh⁻¹, 3.7V vs. Na⁺/Na (1)

Theoretical energy density (310 ÷ 1.26) mAh g⁻¹ × 3.7V = 910 Wh kg⁻¹

Challenges: (1) single stimuli response; (2) Low energy density (<250 Wh kg⁻¹).

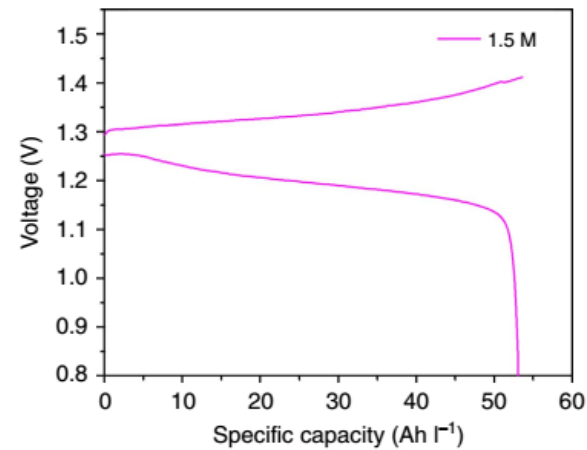
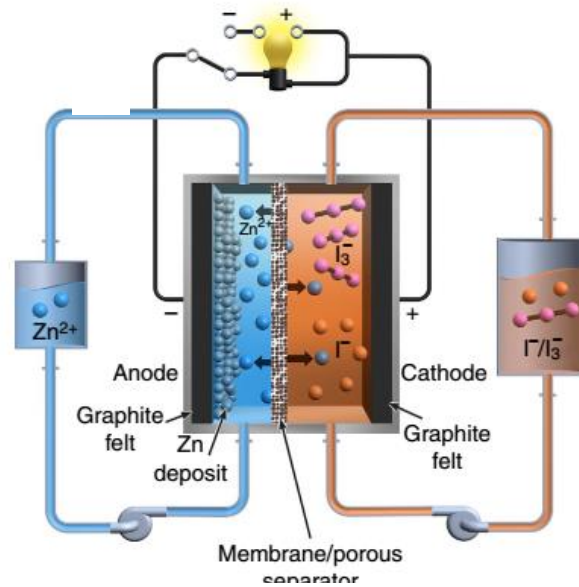
Flowing design strategies

Redox molecular//Br₂ flow battery



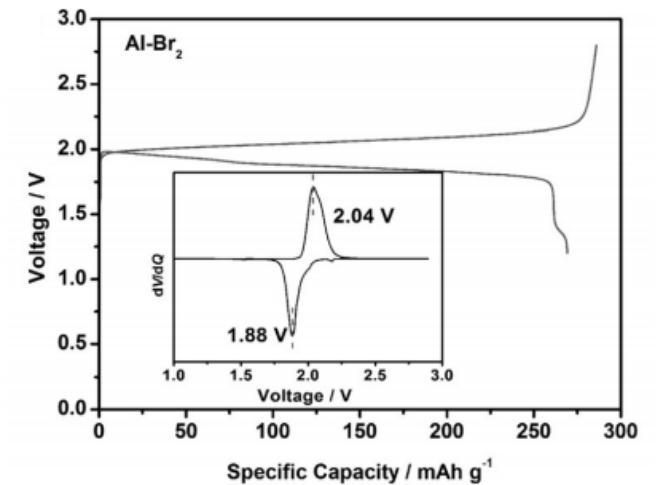
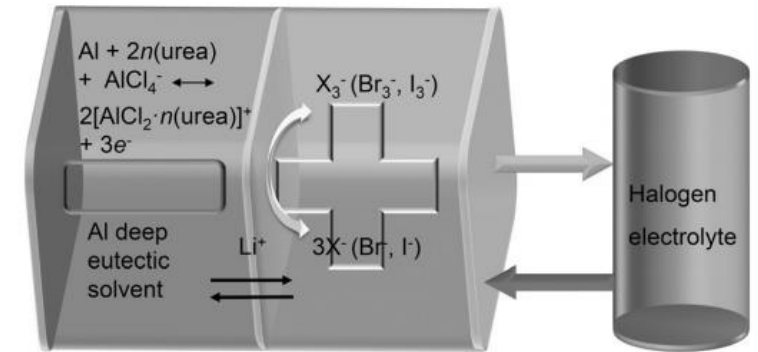
Nature 2014, 505, 195–198.

Zn//I₂ flow battery



Nat. Commun. 2015, 6, 6303

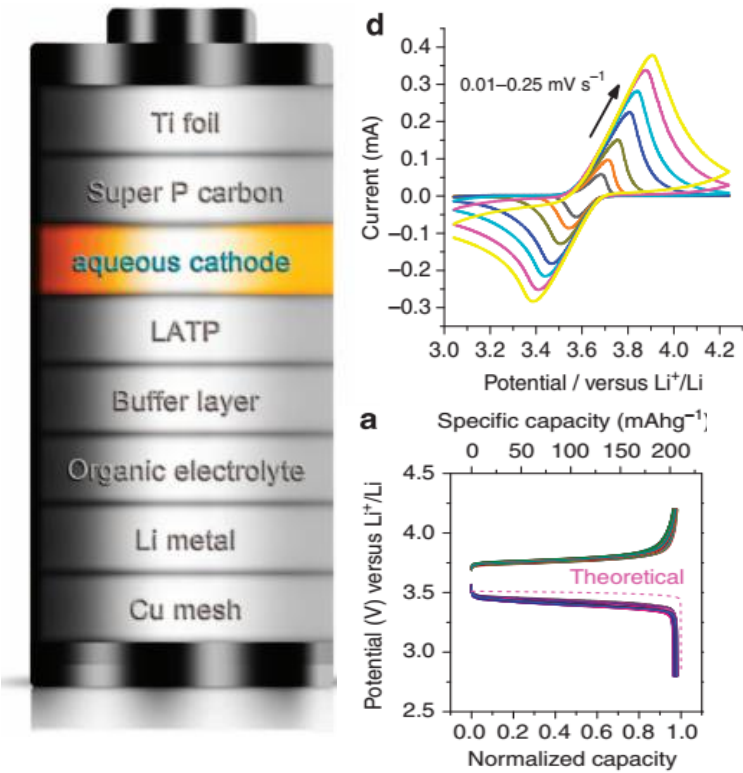
Al//Br₂ flow battery



Angew. Chem. Int. Ed. 2017, 56, 7454

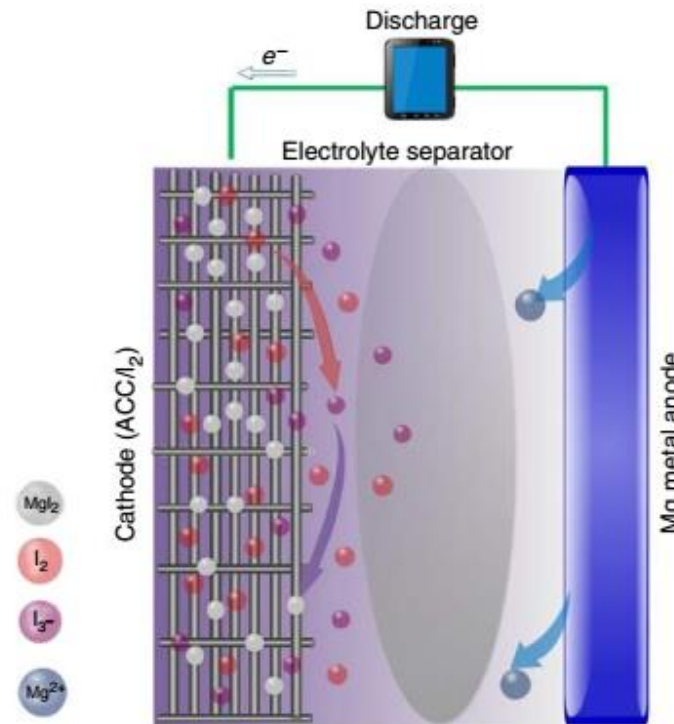
Static design strategies

Li//I₂ battery



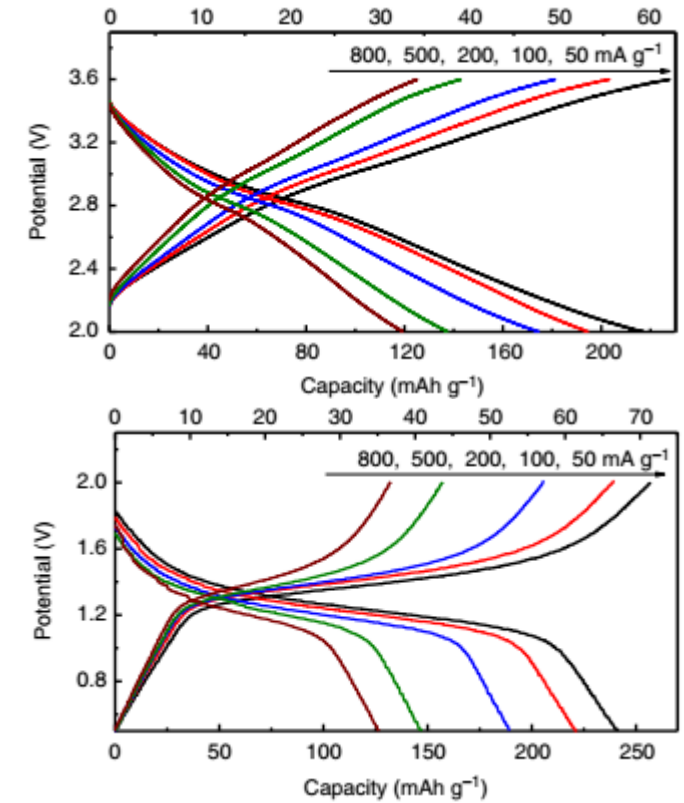
Nat. Commun. 2013, 4, 1896.

Mg//I₂ battery



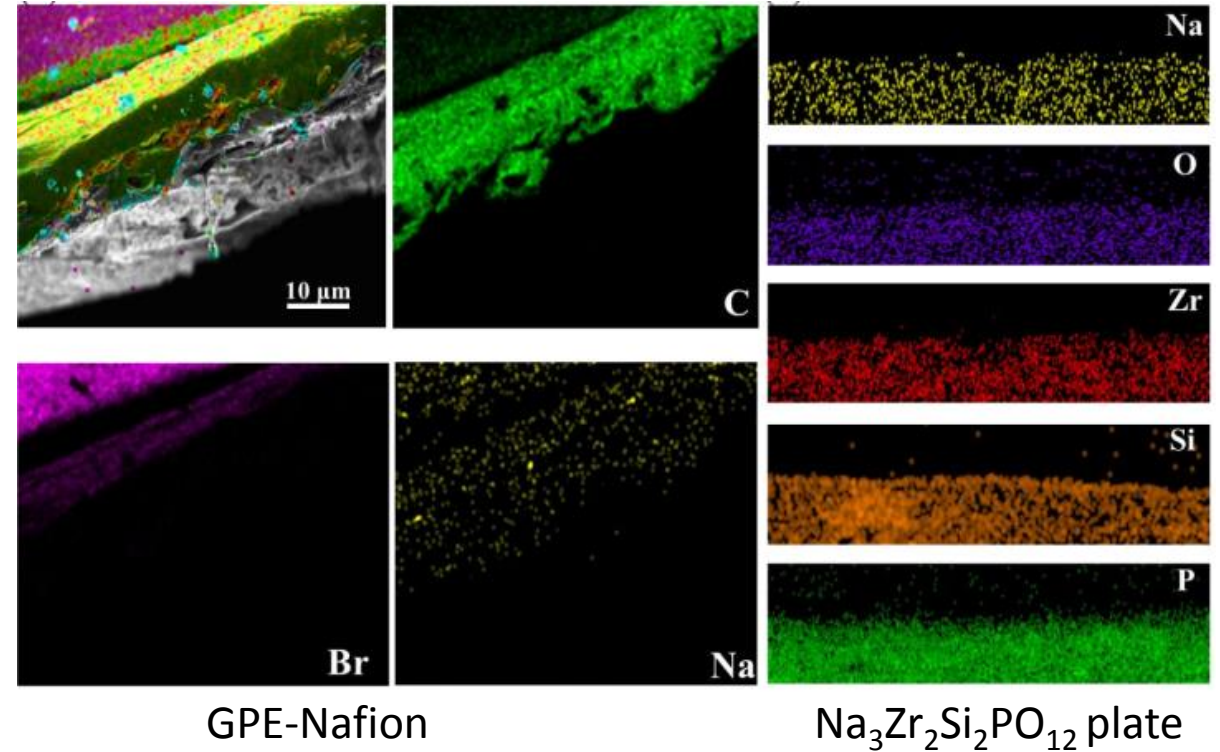
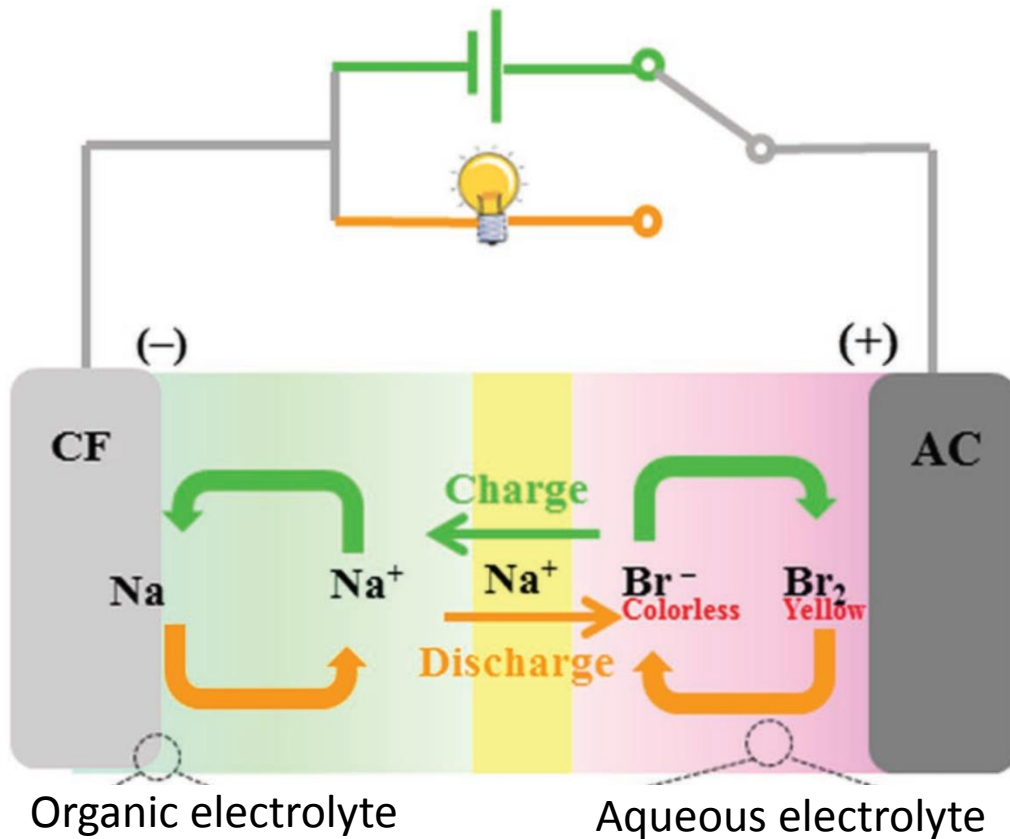
Nat. Commun. 2017, 8, 14083.

C//I₂ battery



Nat. Commun. 2017, 8, 527.

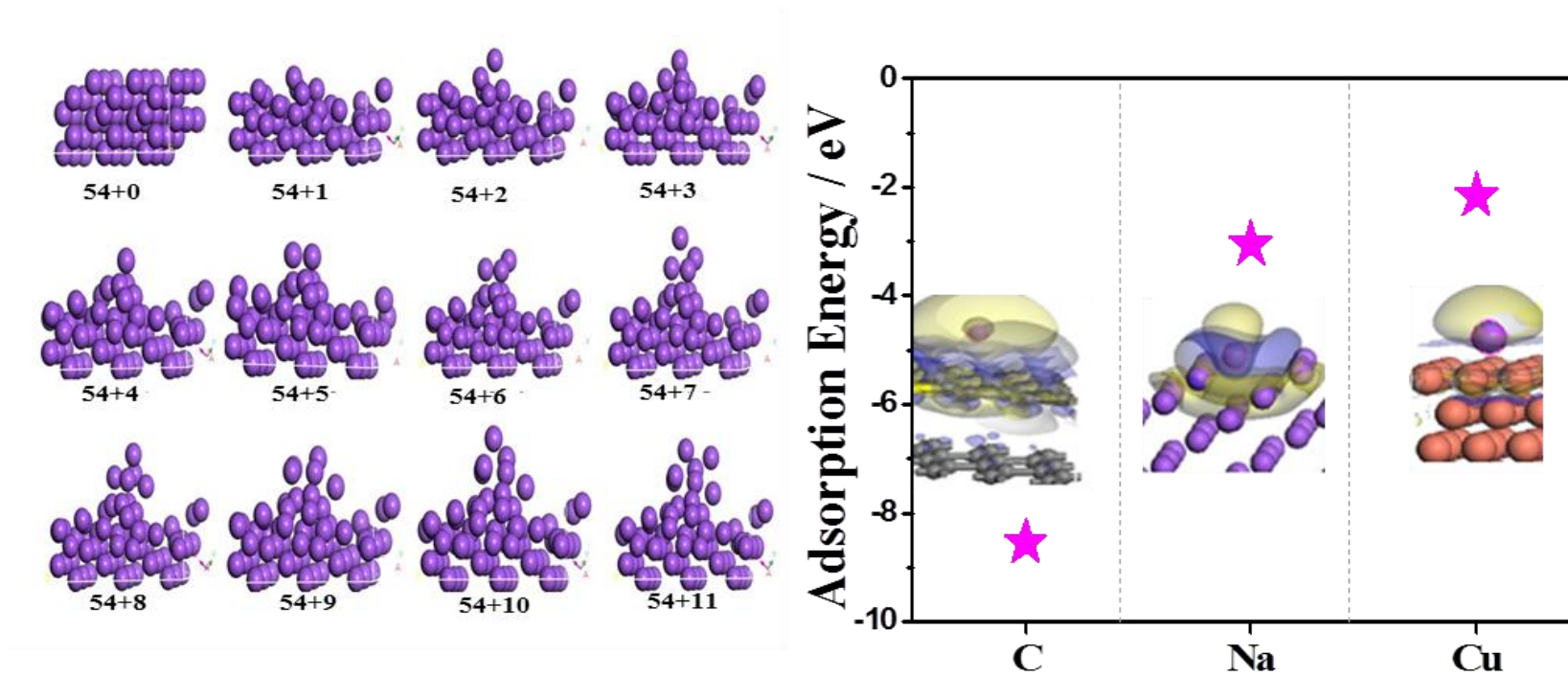
Our design strategies of sodium bromine battery



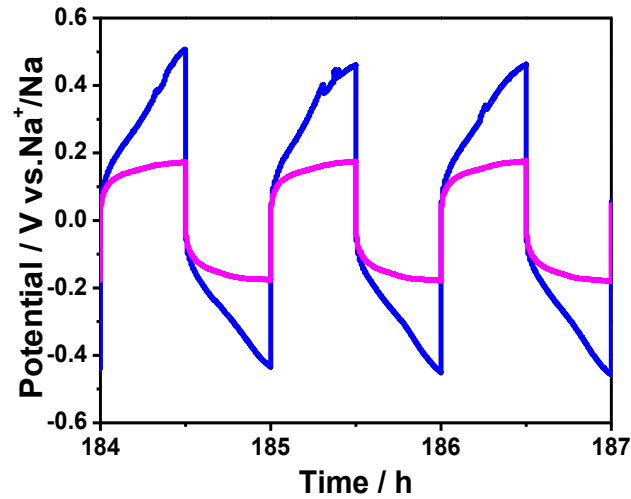
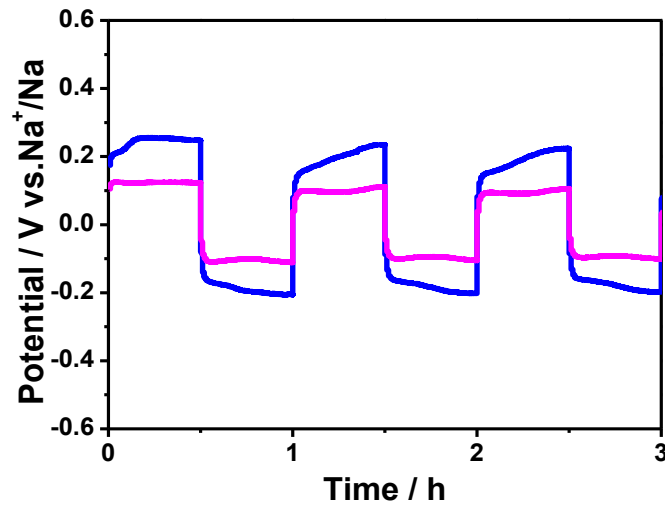
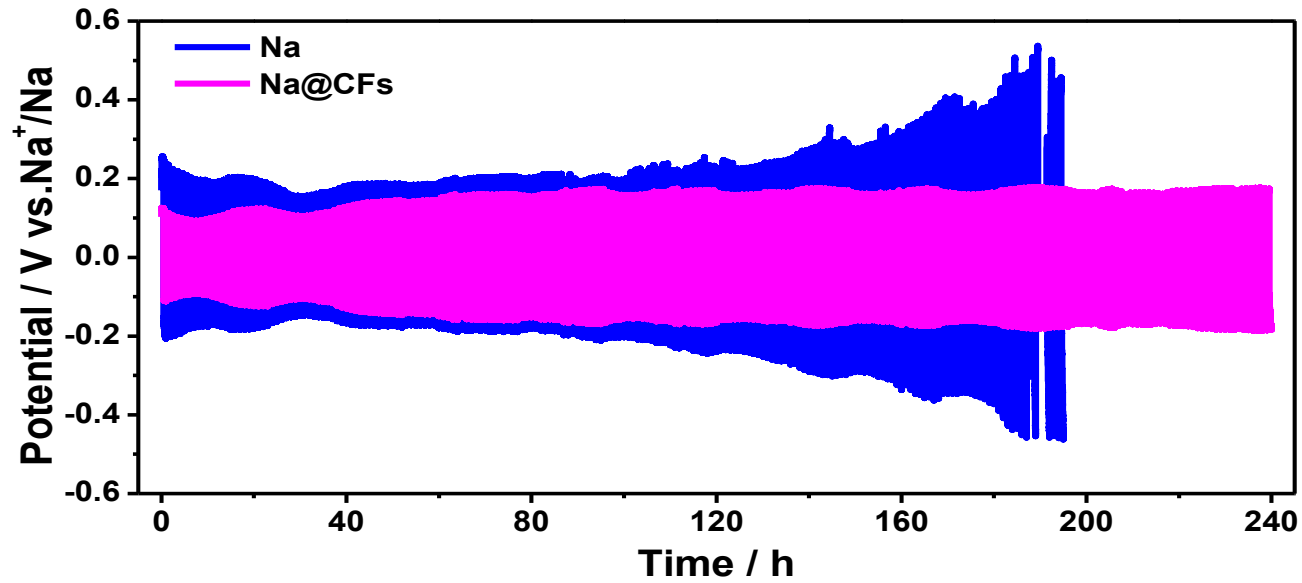
- (+) Cathode: Br⁻/Br₂ in aqueous electrolyte, theoretical capacity of ~330 mAh g⁻¹.
- (-) Anode: Na@CFs in organic electrolyte, theoretical capacity of ~1160 mAh g⁻¹.
- Separators: Na ion conductor (Na₃Zr₂Si₂PO₁₂ plate), only allow Na ion transportation.

DFT Calculations

CASTEP (version 5.4), Universal force field with the Forcite package,
PBE exchange-correlation functions,
A super-cell with 3 layer slab

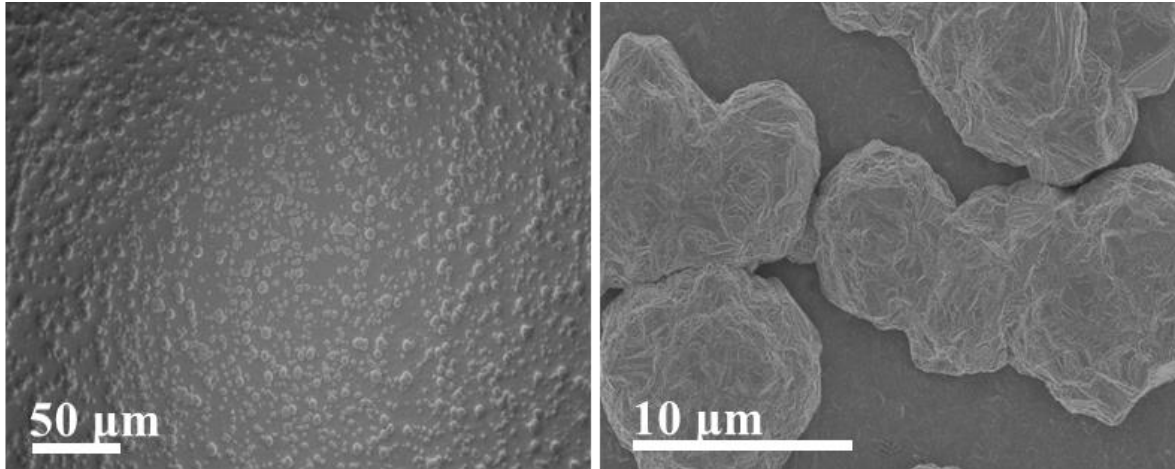


- Unsmooth surface during Na ion deposition process on the Na surface seems to be intrinsic.
- Na atoms are strongly polarized on the CF surface in comparison with those on the Na and copper surfaces.
- For Na@CFs composites, are Na ions preferentially deposited around the CF surface rather than Na protrusions?



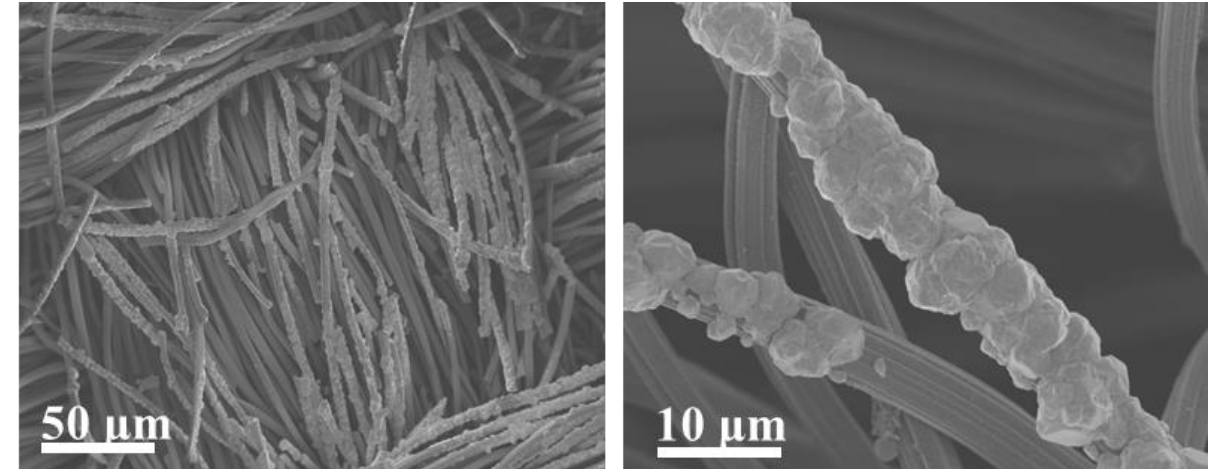
- Bare Na electrode failed after 190 hours cycles
- Na@CF electrode is stable over 240 hours cycles.

Pure Na anode



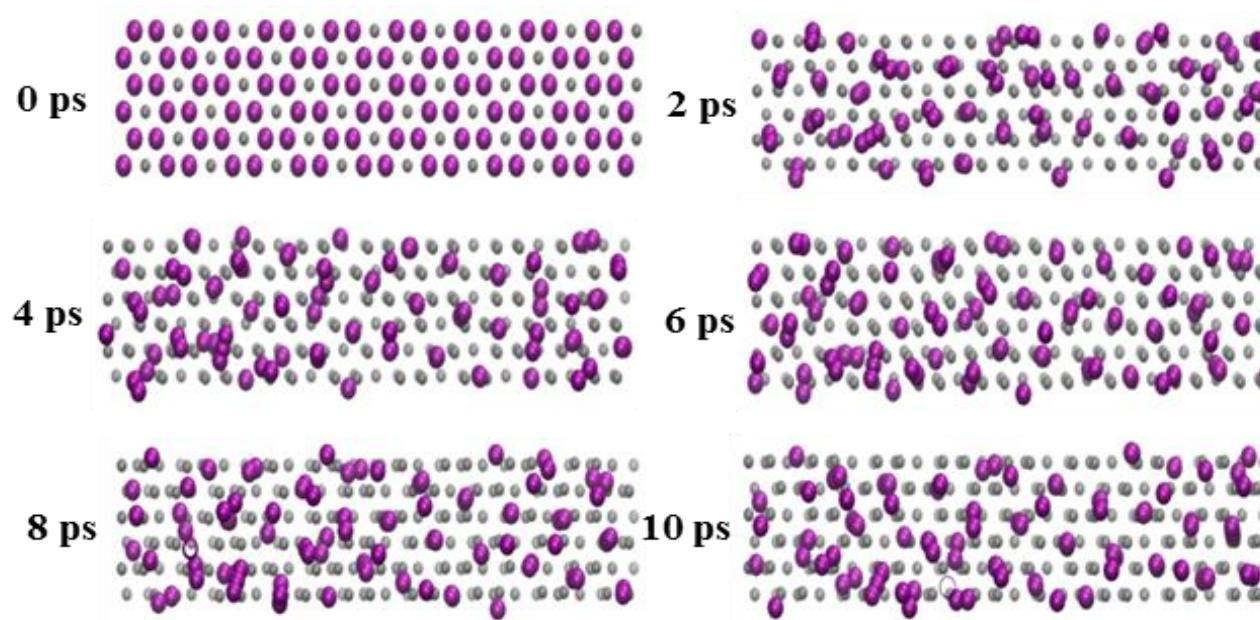
- Na dendrites were clearly from the top-view SEM image of bare Na metal after 100 cycles.
- Uneven growth of Na would ultimately penetrate through the separator and cause internal short circuits.

Na@CFs anode



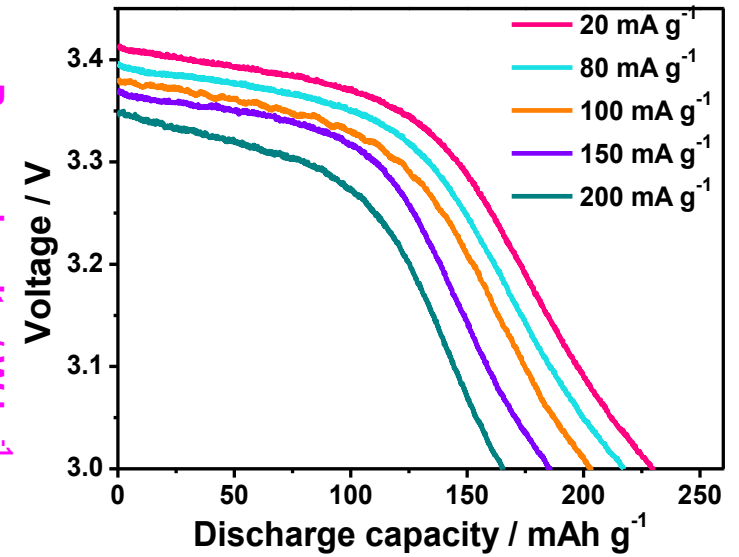
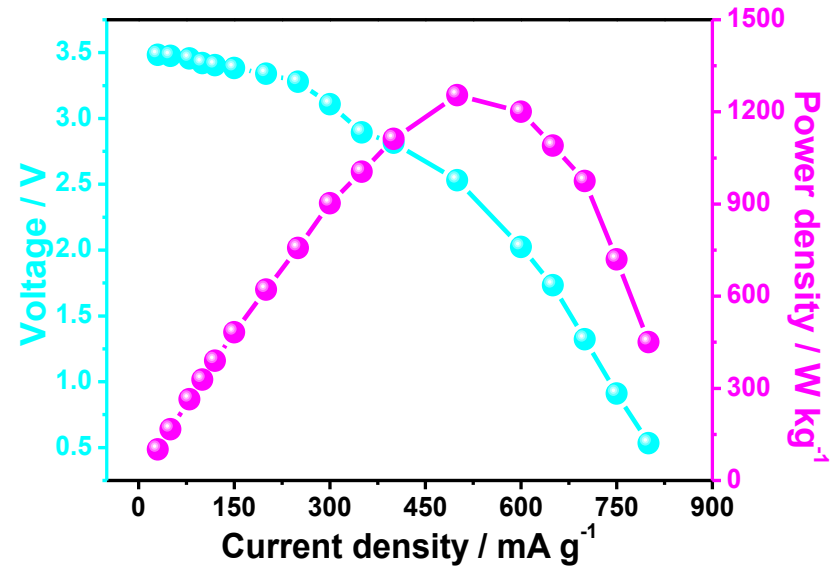
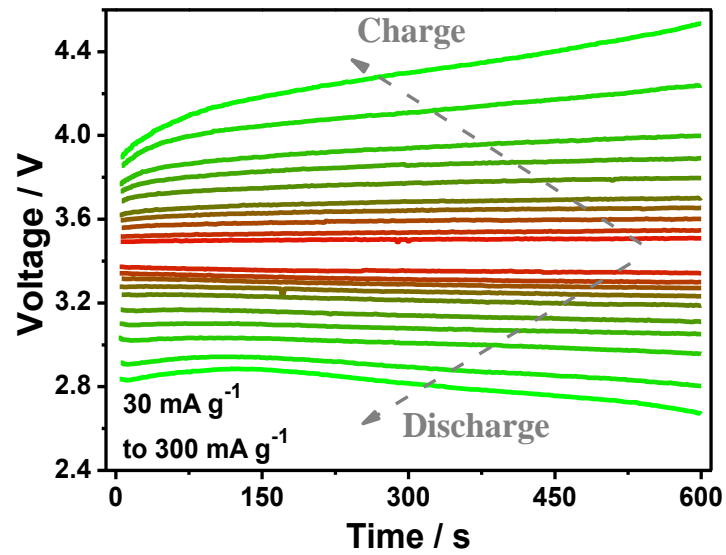
- Smooth and conformal Na deposition on the surface of CFs after 100 cycles
- Horizontal growth instead of vertical growth. The suppressed Na dendrite growth.

MD simulations

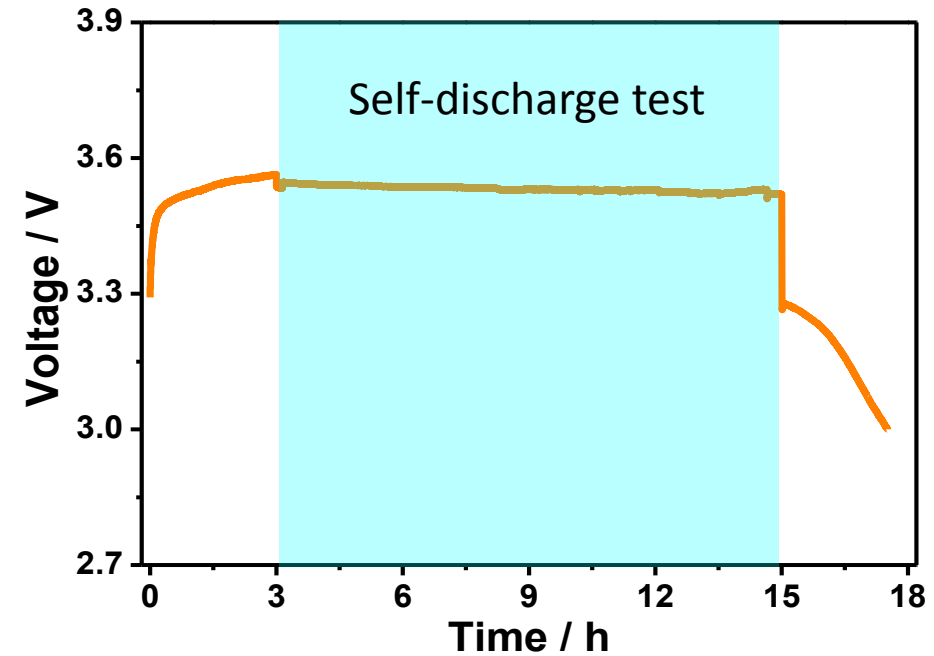
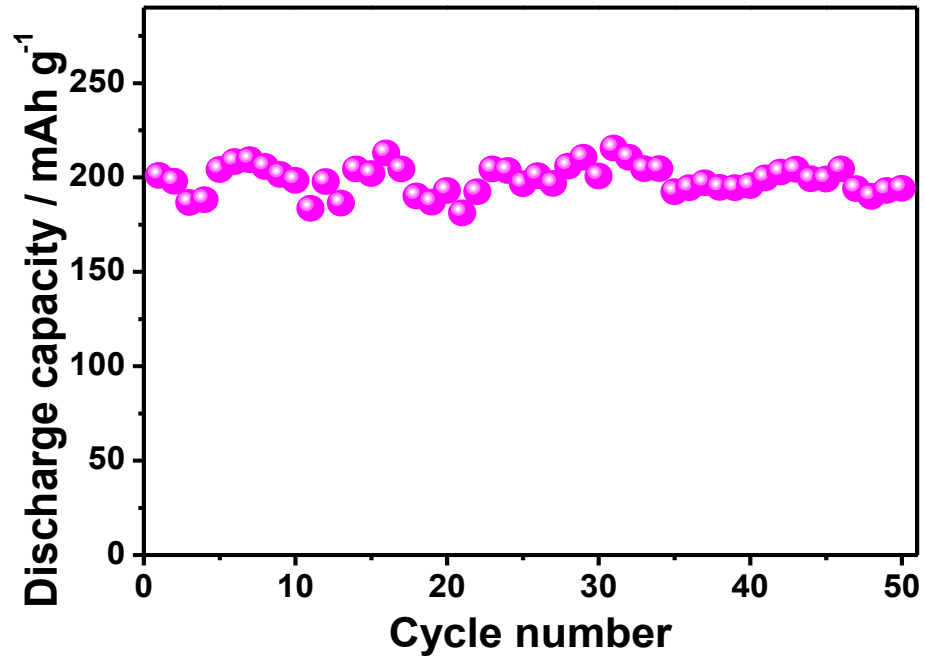


- Na atoms tend to uniformly distribute on the CF surface without forming large cluster

VASP (version 5.4), PBE exchange-correlation functions, A (3x16) super cell

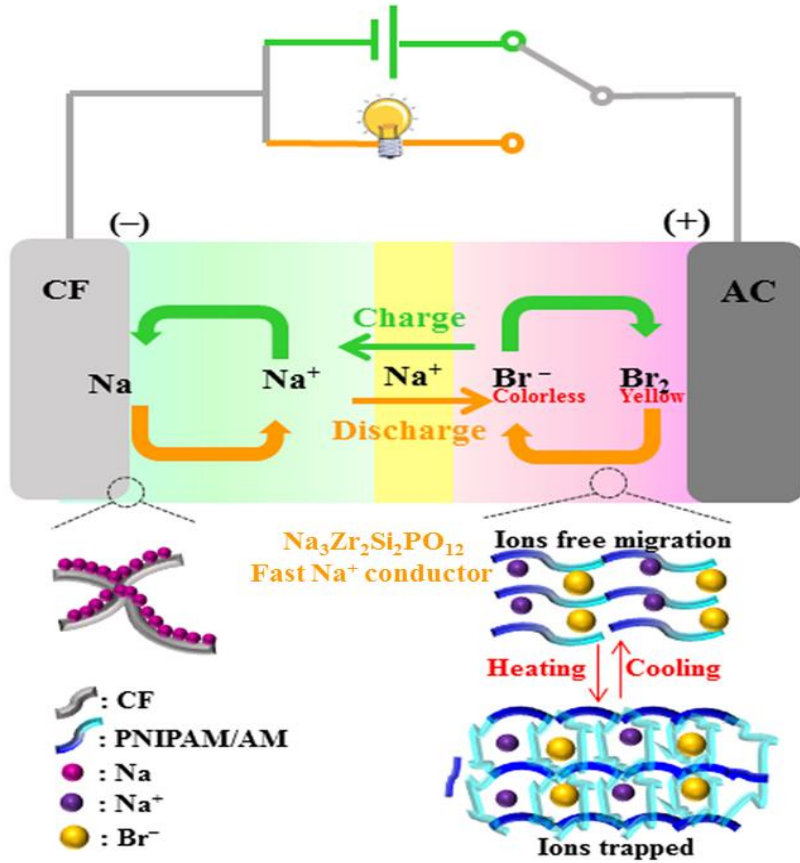


- Maximum operating voltage: 3.4V at 0.02 A g⁻¹.
- Peak power density: 1200 W kg⁻¹.
- Maximum energy density: 760 Wh kg⁻¹.



- Stable 50 cycles at 0.1 A g⁻¹.
- No clear self-discharge.

Our design strategies of dual-stimuli-responsive sodium bromine battery

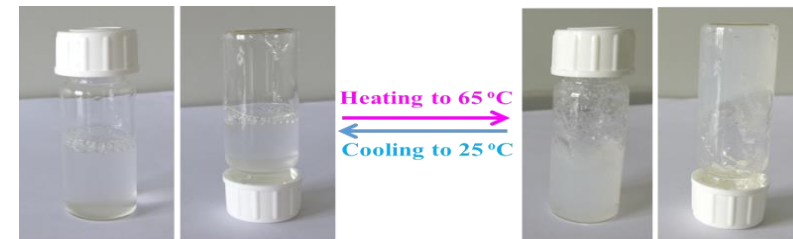


Charge

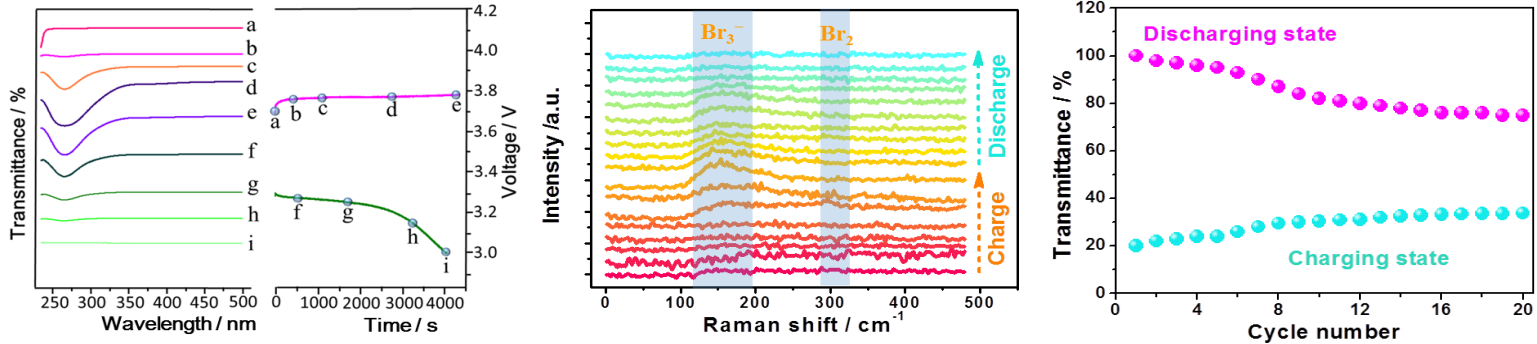


Discharge

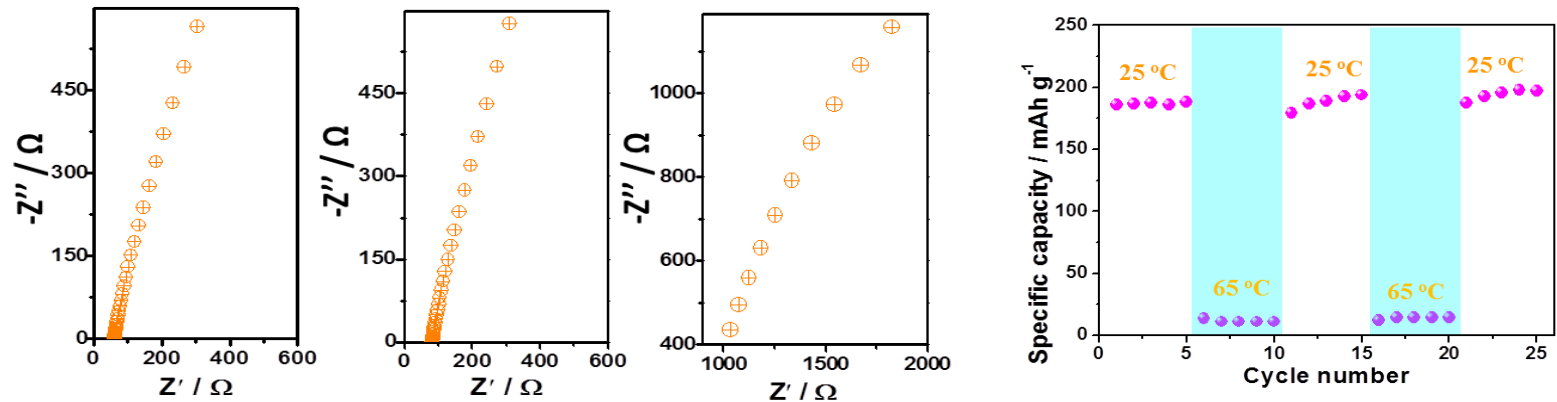
Electrochromic (colorless Br⁻/yellowish Br₂)



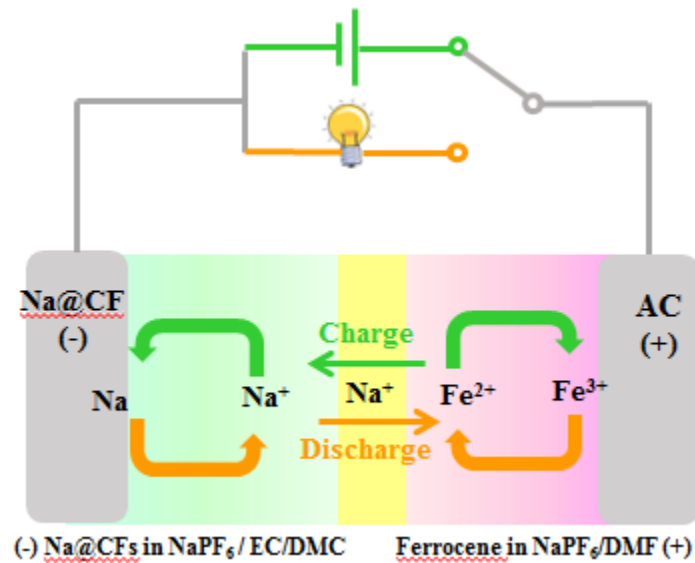
Thermal response (PNIPAAm/MC based smart electrolyte)



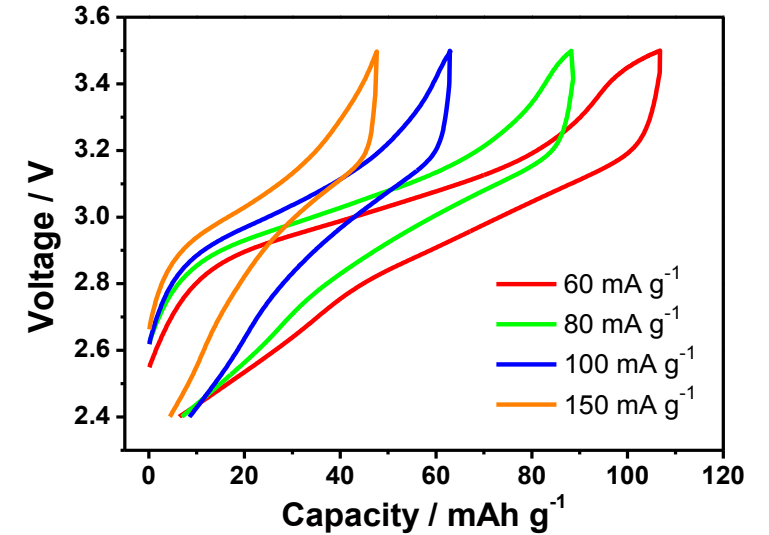
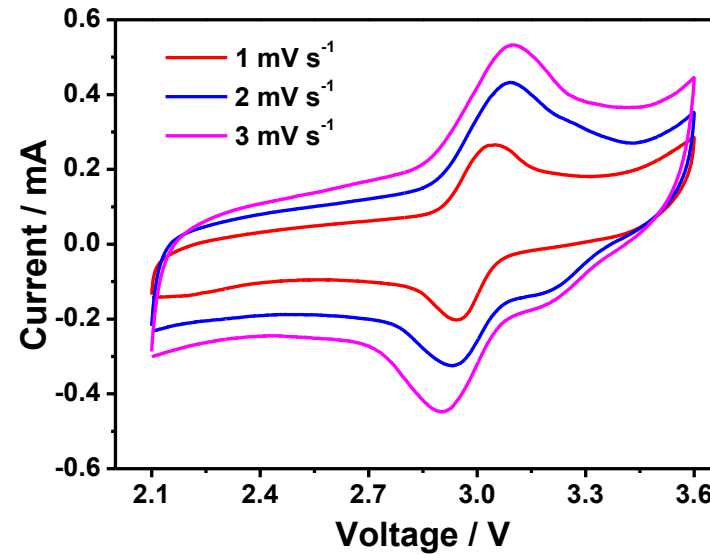
- The transmittance changes in response to the charge/discharge voltage
- The cycle of colored/bleached transmittances still need to be improved



- The sol-gel transition in PNIPAAm/MC lead to increasing bulk resistance
- Fast switch-off at ≥ 65 °C and recovery after cooling to room temperature

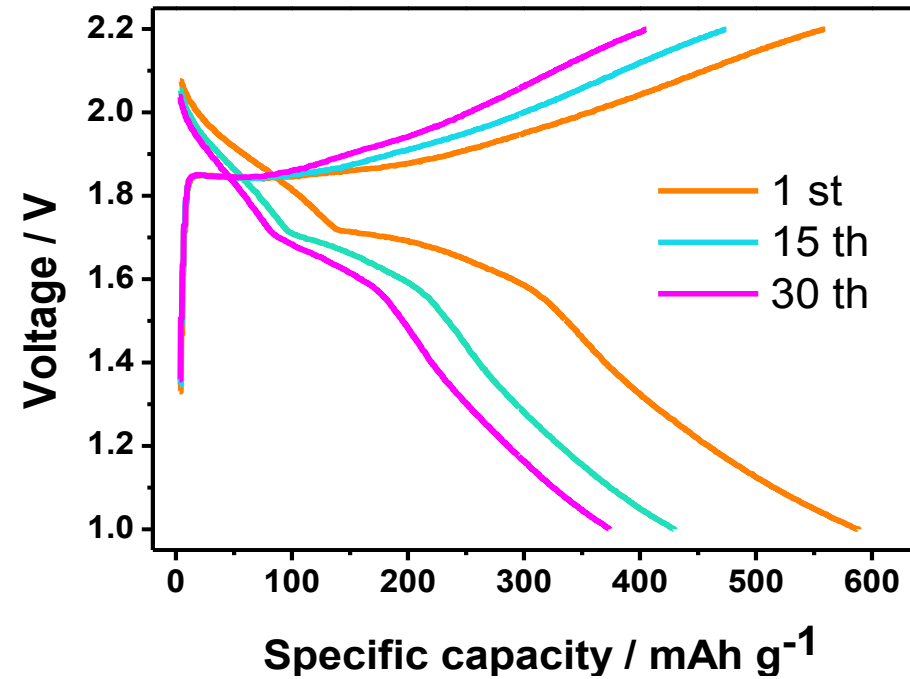
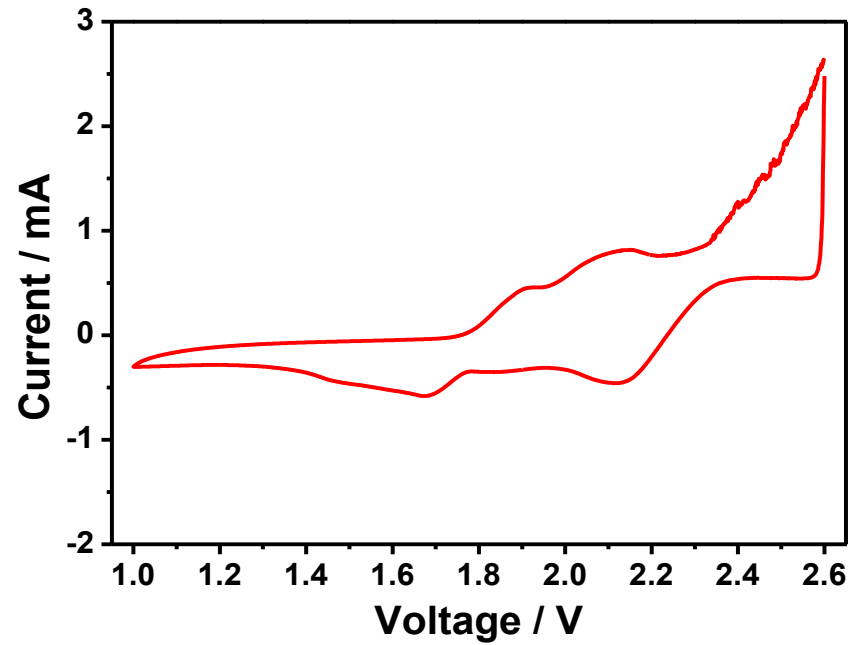


Na//Ferrocene battery



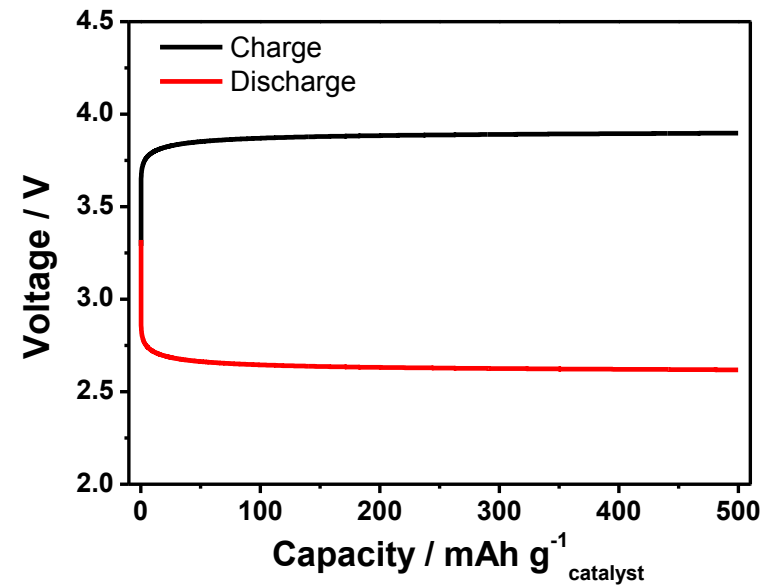
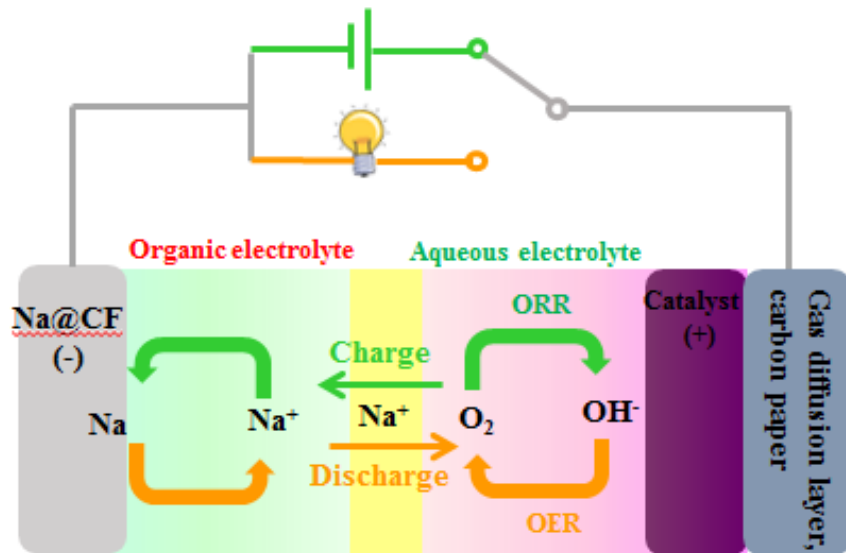
➤ The cathode in the Na//Br₂ battery can be extendable to other redox couple, like Na//Ferrocene battery.

Na//S battery with Na@CFs as anode



➤ The anode in the Na//Br₂ battery can be extendable to other Na-metal battery, like Na//S battery.

Na//air battery with Na@CFs as anode



➤ The anode in the Na//Br₂ battery can be extendable to other Na-metal battery, like Na//air battery.

Summary

- The electrochromic and thermal-responsive and Na//Br₂ battery was fabricated.
- This dual-stimuli-responsive battery achieved an energy density $> 700 \text{ Wh kg}^{-1}$.
- The used cathode can be extendable to other redox couples.
- The used anode can be extendable to sulfur and air batteries.

Thanks for your attention

