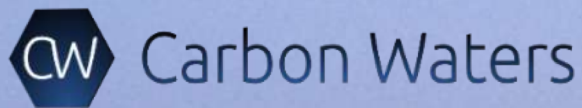


Graphene for a sustainable world



Alain Pénicaud

Centre de Recherche Paul Pascal, CNRS, Université Bordeaux-I

penicaud@crpp-bordeaux.cnrs.fr



université
de **BORDEAUX**

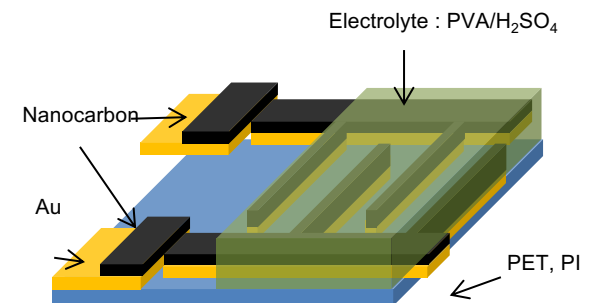
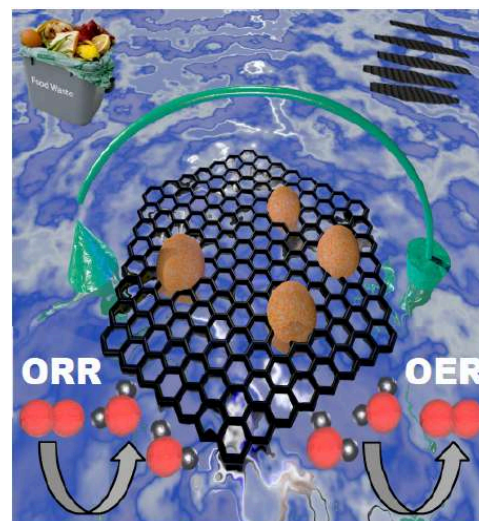
Graphene for a sustainable world

Less pollution

- Replace metals
- No surfactants added
- water based formulations

Energy saving

- anti corrosion coatings
- supercapacitors
- electrocatalysts



Outline

- Eau de Graphène (graphene water):
Additive free Single layer graphene in water



- Carbon Waters : the Company

- Eau de Nanotubes

- Multi layer Graphene from food waste ?

- electrocatalysts
- supercapacitors

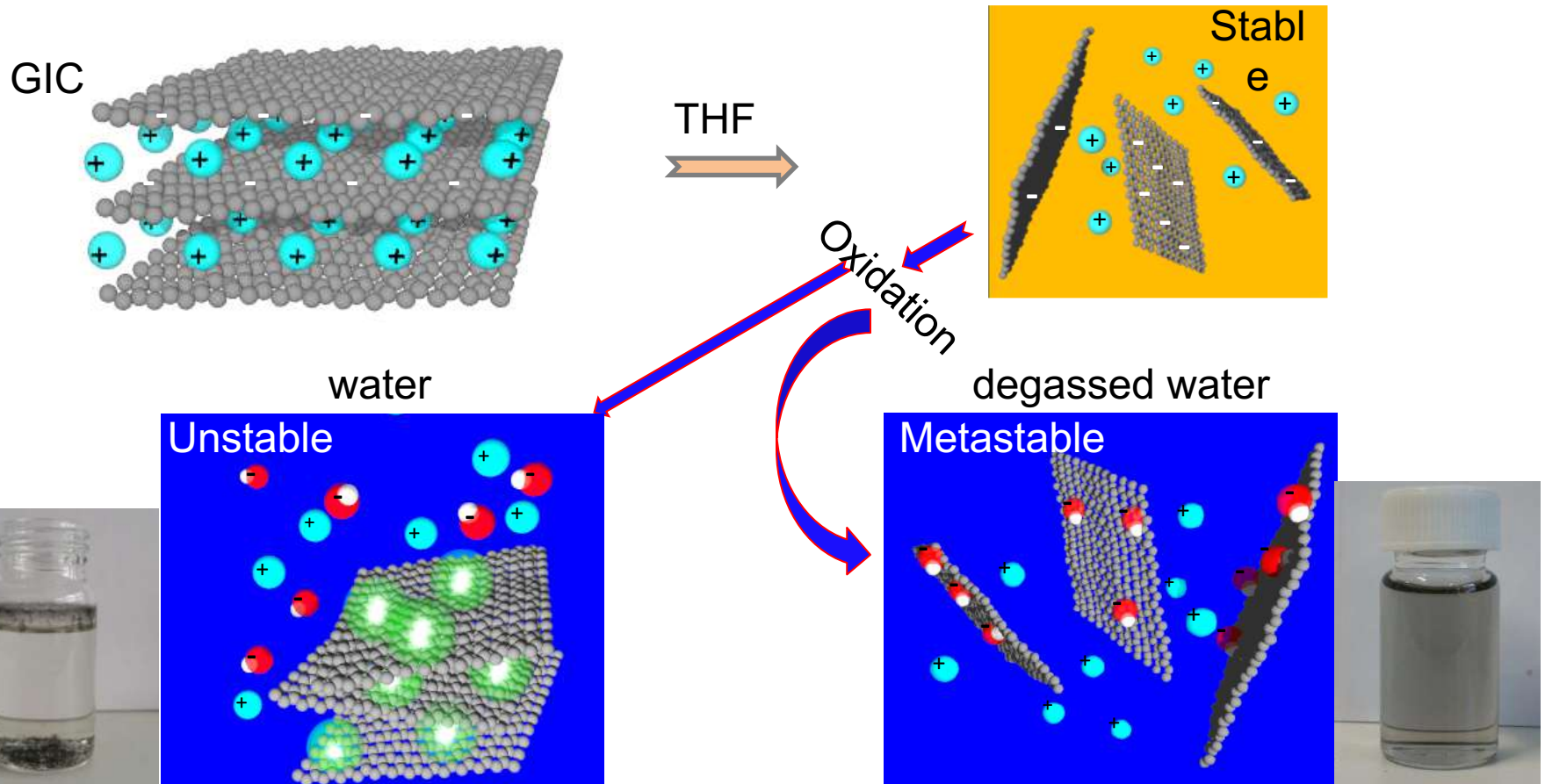


Stabilizing SLG in water : Eau de Graphène

(no surfactant, no organic solvent)



George Bepete



Valles *et al.*, JACS. (2008), Catheline *et al.*, Soft Matter. (2012)

Bepete *et al.*, Nature Chem, 2017, J. Phys. Chem C, 2016, Phys. Status Solidi RRL 2016

99,95% pure & 100% safe liquid graphene

for high-performance materials

INDUSTRY SOLUTIONS

GRAPHENE DISPERSIONS

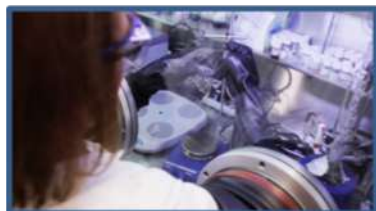
<https://www.carbon-waters.com/>

Carbon Waters



3 patents
Nature Chemistry 2017

- No nanoparticles steps
- No additives
- Extremely stable and homogeneous dispersions
- Very strong control on graphene quality
- Industrialization compatible



Dispersed nanocarbons
in aqueous solution



Surface preparation, printing
& treatment



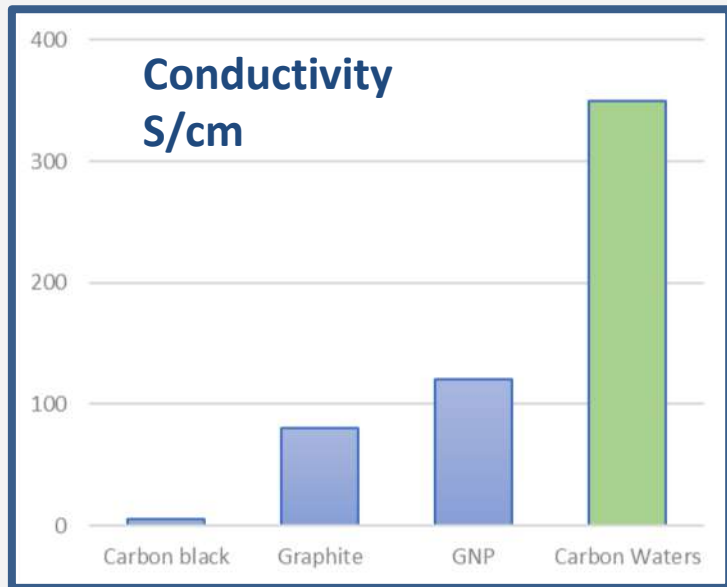
Surfaces with enhanced
and reinforced properties

Design and production of graphene dispersions and graphene-treated surfaces

Example of on-going development : liquid dispersions

Our advantages :

- High conductivity compatible with an EMI Shielding effect
- Stable on epoxy and latex
- Compatible with flexible and transparent substrates
- Fully printable with several processes

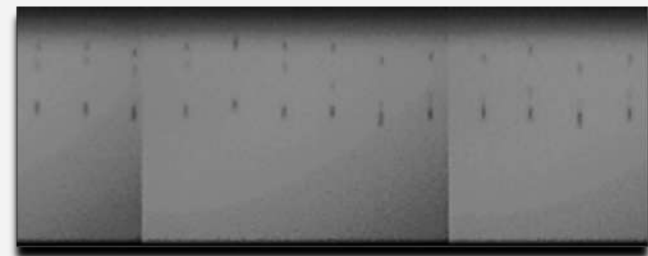


(1)

(2)

(1) *IDTechEx Graphene 2016*

(2) *In house measurement*

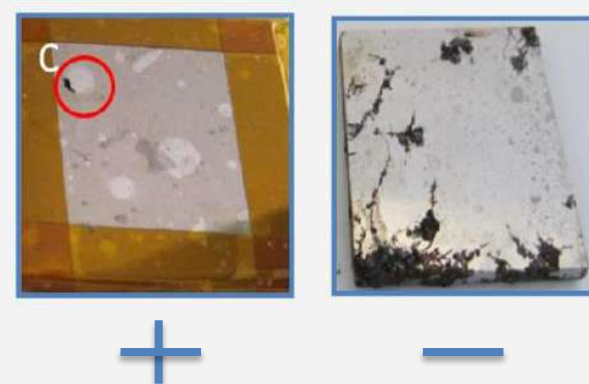


50 μm size drops

We print and develop transparent conductive films for conductive composites and EMI Shielding applications

Example of on-going development : printed surfaces

- Water-based graphene dispersions
- surface deposition analyzed by RAMAN
- Specific pre-treatments
- Filtration or electrodeposition



Steel treated surface

Our advantages :

- Thinnest material known for corrosion protection
- Efficient and time-saving printing
- Resistance to harsh conditions
- REACH compliant (Cr(VI) and cadmium replacement)
- Improved thermal management

We print our nanomaterials on steel surfaces in order to decrease corrosion and aging

Carbon Waters

Operational team



Alban Chesneau
CEO
PhD-MBA
12 yo technology marketing & innovation management



Dr Fabienne Dragin
Application manager
PhD nanotechnology
8 yo in nanomaterials and composites



Dr Julia Messner
Process manager
PhD Chemical engineering
5 yo in chemical processes

Total team of 6 persons

Industrial network



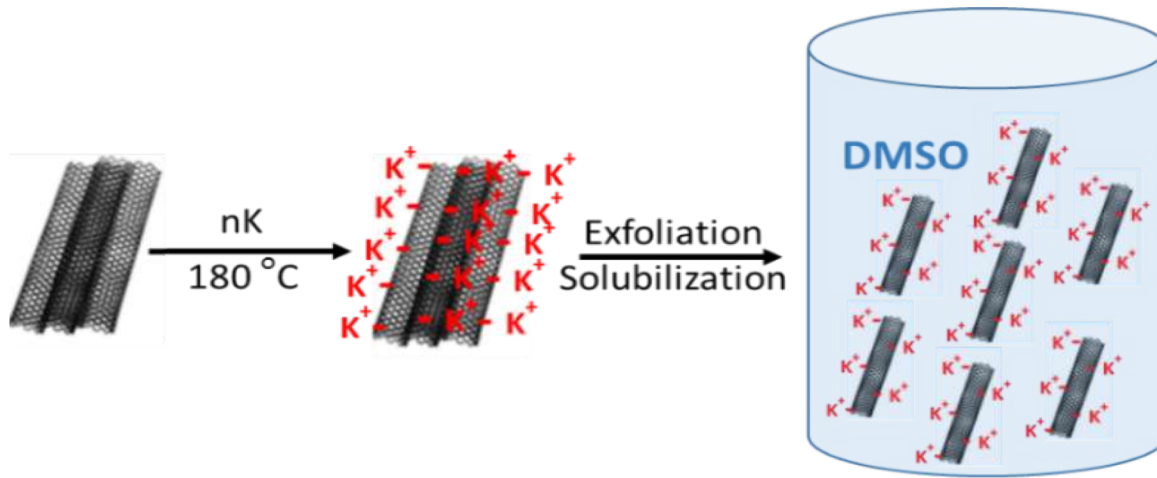
Patrick MAESTRO
Industrial advisor



Kevin ARTHUR
Serial technological entrepreneur



Preparation of Eau de Nanotubes



Inert atmosphere



Air



1.



Unstable
Closed SWNTs in
degassed water

2.



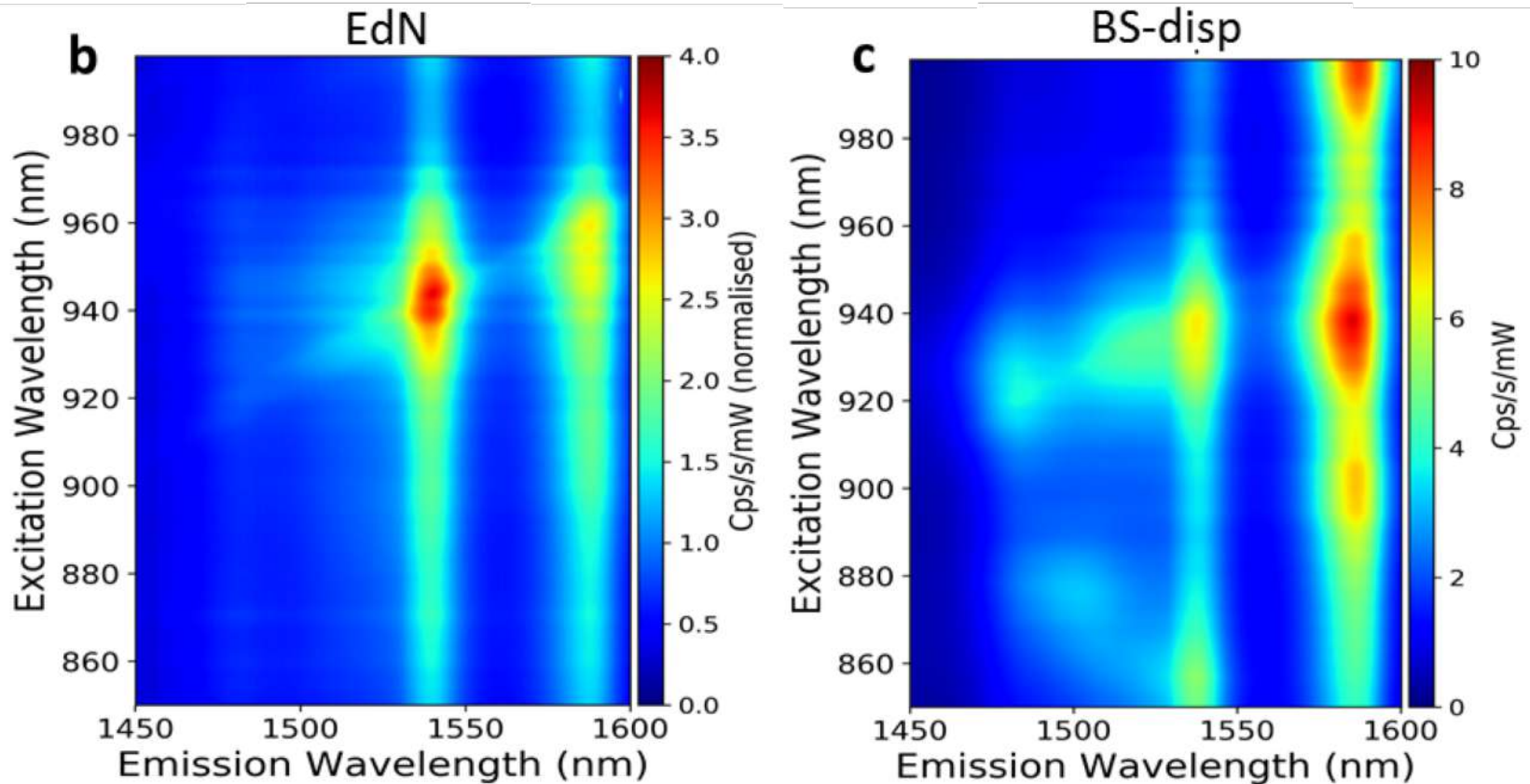
Unstable
in non degassed water

3.



“Eau de nanotubes”
Metastable (for months)
in degassed water/DMSO ($c=25\text{ mg/L}$)
No surfactant/polymer additives

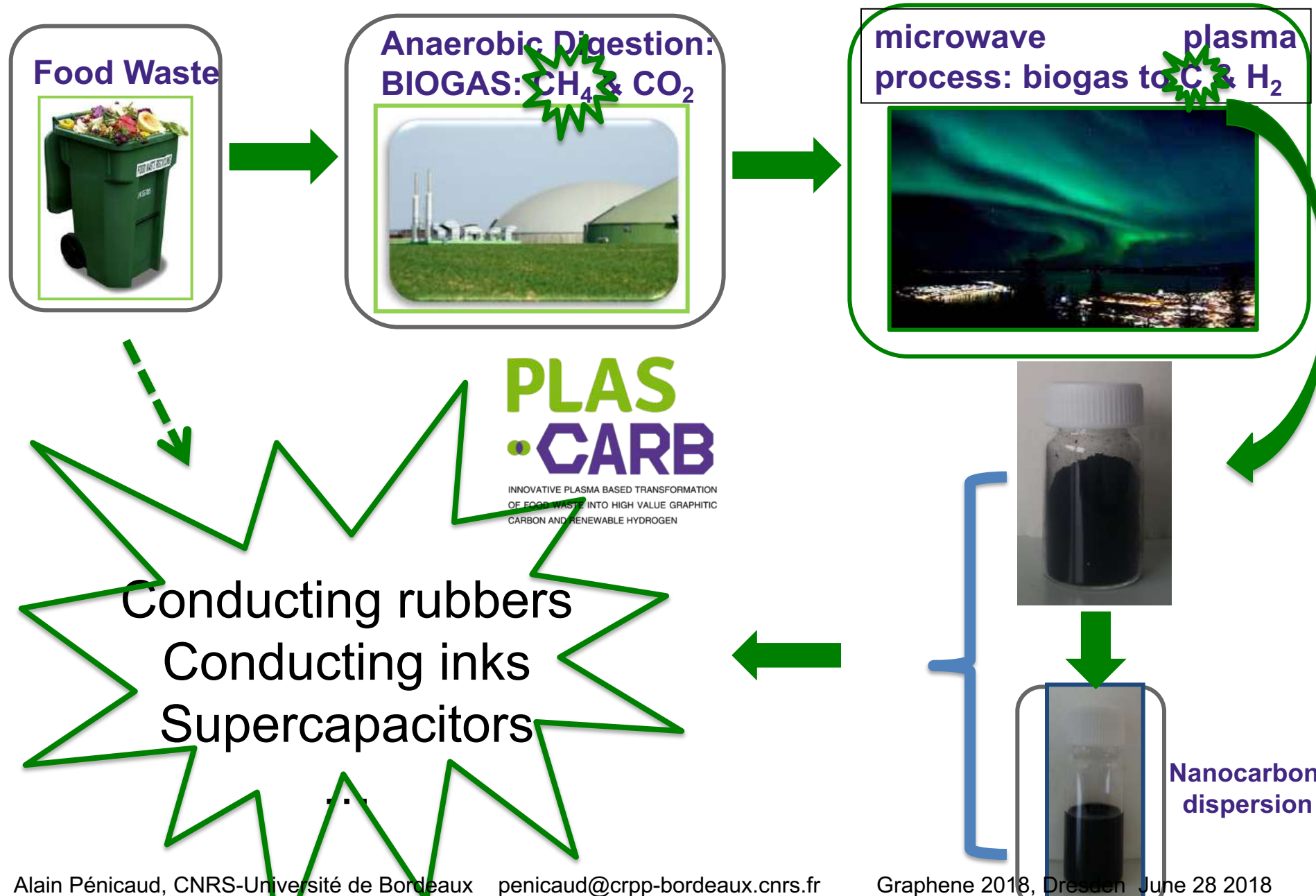
PL spectroscopy of EdN



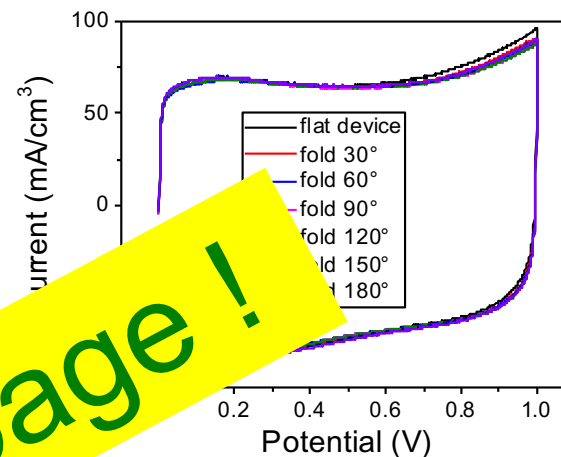
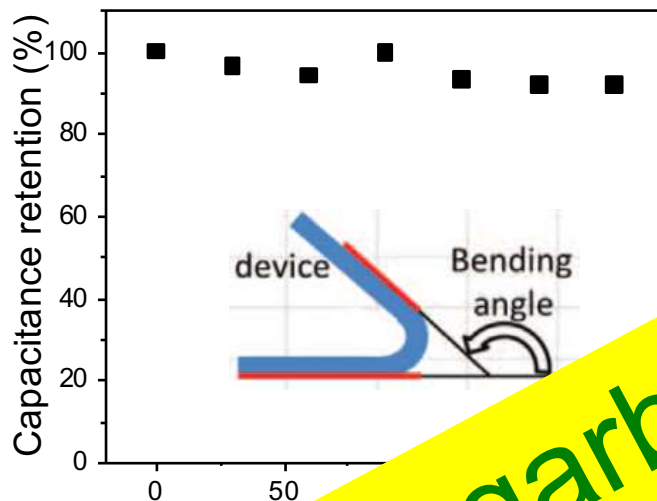
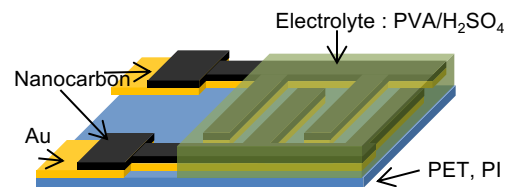
PLE profiles are close for **EdN** and **bile salt** suspensions
Redshift of ≈ 10 nm of S₂₂ peaks for **EdN** \rightarrow different environments

\rightarrow Bepete et al. ACS Nano 2018

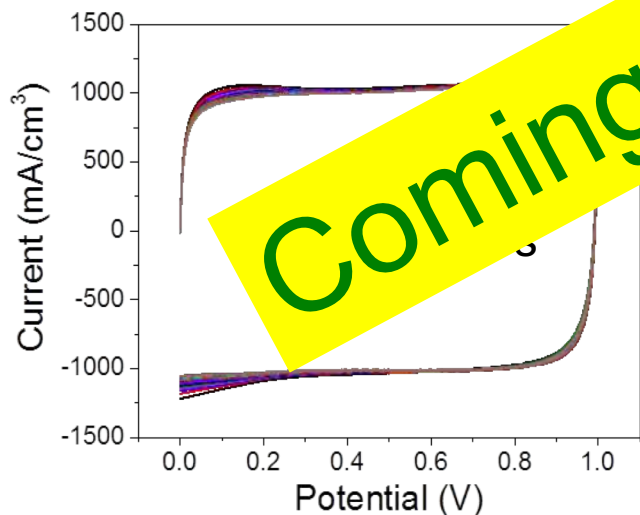
Multilayer graphene from food waste



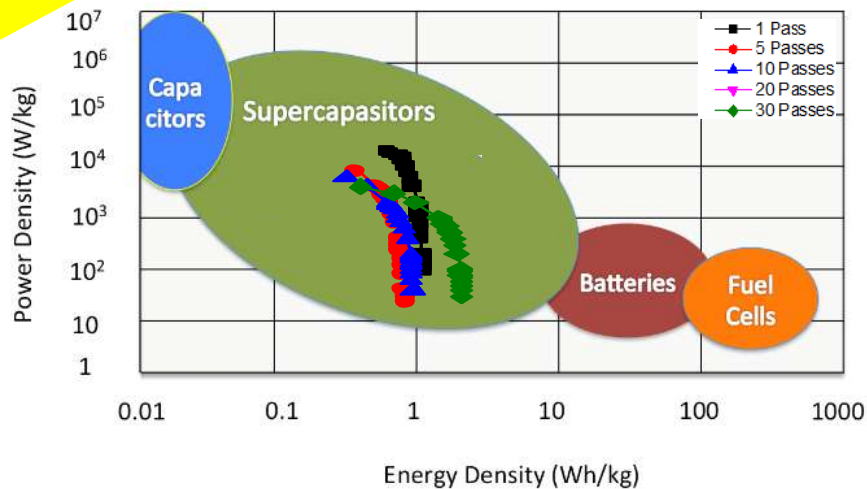
Microsupercapacitors

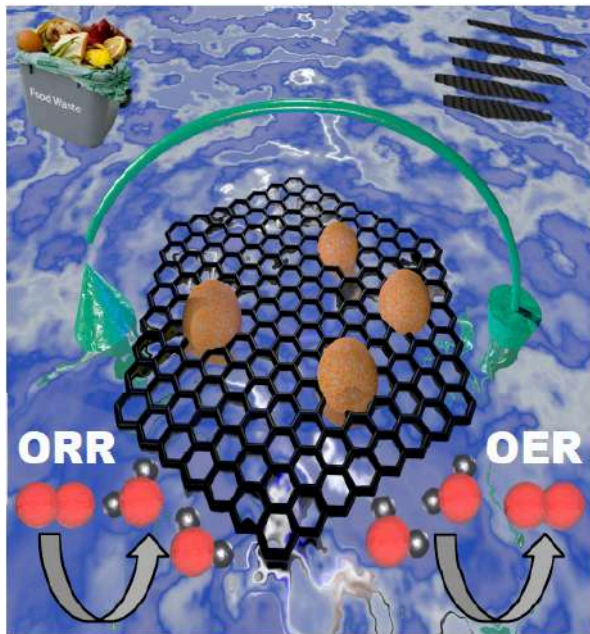


$$C = 10 \text{ F/cm}^{-3}$$



Coming from garbage!

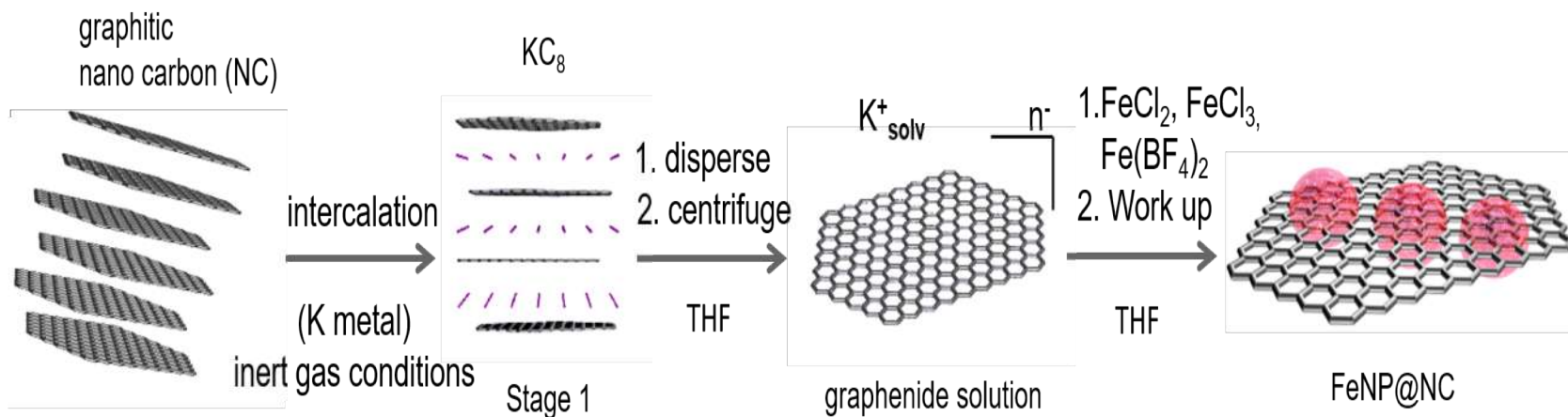




From Food Waste to Efficient Bifunctional Nonprecious Electrocatalyst



Ferdinand Hof

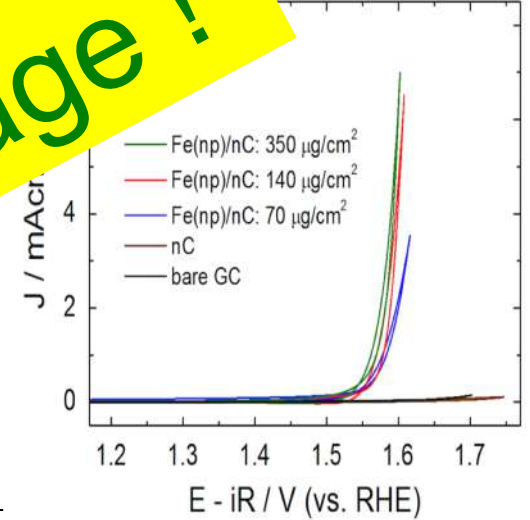
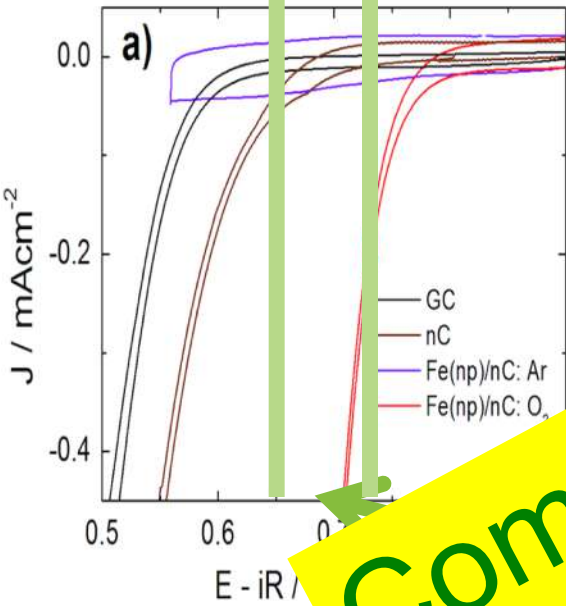
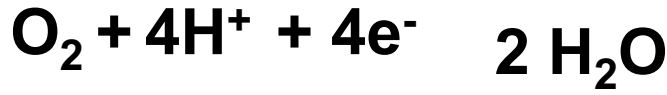


Hof, F. et al. Chem. Eur. J. 2017

Hof, F. et al. Chem. Eur. J. 2018

Oxygen reduction reaction (ORR)

Oxygen evolution reaction (OER)



Coming from garbage!

Commercial Pt catalysts
 for fuel cells exhibit an overpotential
 of 0.5 - 0.6 V vs RHE
 theoretical value: $E^\circ = 1.22 \text{ V vs RHE}$

GC bare glassy carbon
nC graphitic nano carbon

Hof, F. et al. Chem. Eur. J. 2017



Plascarb project :

Dr Katerina Kampioti

Dr Ferdinand Hof

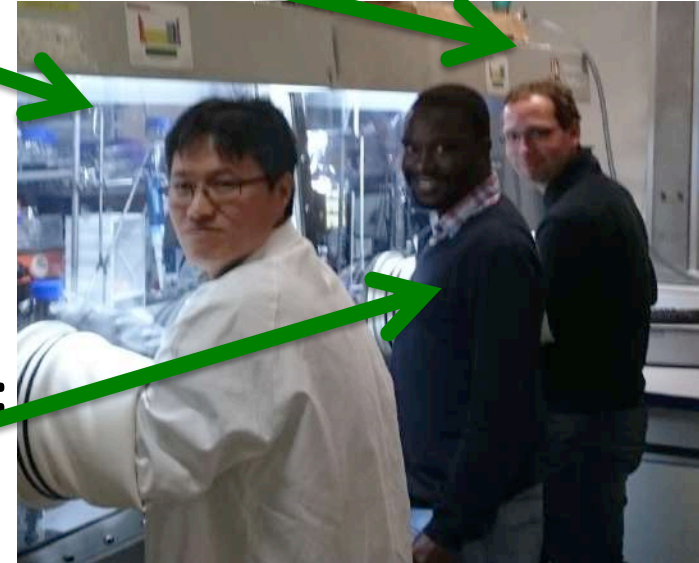
Dr Kai Huang



Rubber composites

Prof. A. Zarbin (Curitiba)

Dr Carolina Matos (“)



Eau de Graphène & nanotubes:

Dr George Bepete

Dr Carlos Drummond

Dr Fernando Torres cañas

Prof. Eric Anglaret (Montpellier University)


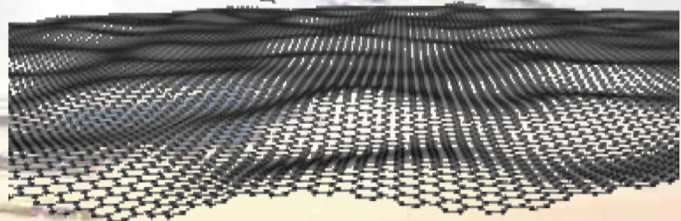
Dr Nicolas Izard (“)

Carbon Waters:

Dr A. Chesneau, Dr. F. Dragin, Dr. J. Messner



Chem  nTubes 2020



International
Meeting on the
Chemistry of
Graphene and
Carbon Nanotubes

19-23 or 26-30 April 2020

San Sebastián (Spain)

*Open Registration from January
2019*

Email: acriado@cicbiomagune.es