

# Raman signatures of SLG dispersed in *degassed* water ("eau de graphene")





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### Why studying SLG in water ?



Flow/friction of water between GR flakes Falk et al, Nanolett. 2010





Aqueous precursors for GR-based films/materials @extremetech.com



Harvesting energy from waterflow over GR *Dhiman et al, Nanolett. 2011* 

### How to disperse/stabilize SLG in water ?



→ Dissolution of graphite intercalation compounds (GIC) in polar aprotic solvent Drummond & Penicaud., Acc. Chem. Res. 2013

→ +Transfer and stabilization of graphenide ("SLG anions") in degassed water Bepete et al., Nature Chemistry, 28 Nov. 2016

### Stabilizing SLG in water





### Evidences of SLG in "eau de graphene"







SLG in *degassed* water (no surfactant/organic additives)

c=0.16 g/L≈400 m²/L , pH=8

Typical flake size 100 nm-1  $\mu$ m

Bepete et al., Nature Chemistry, 28 Nov. 2016



### Raman signatures of "eau de graphene"





Metastable in *degassed* water (c=0.16 mg/mL, pH=8) No surfactant/organic additives









Intensity of SLG bands comparable to that of OH bending Relative intensities of SLG increase at low laser energies



### Raman signatures of SLG in "eau de graphene"





Bepete et al, J. Phys. Chem. C 2016



### Raman signatures of SLG in "eau de graphene"



#### FLG in sodium cholate SLG in EdG stabilized aqueous suspensions 2.33 eV 2.33 eV Intensity (a.u.) G,D' **2D** D 1.94 eV .94 eV ntensity (a.u.) 1.58 eV .58 eV Water 1.17 eV 1.17 eV Water absorption absorption 1600 1200 2000 2400 2800 1200 1500 1800 2400 2700 2100 Raman shift (cm<sup>-1</sup>) Raman shift (cm<sup>-1</sup>)

## Intrinsic signature of (an ensemble of) SLG in water :

a narrow and intense 2D band

Bepete et al, J. Phys. Chem. C 2016



### **Raman signatures of SLG in "eau de graphene"** Single layerness





Intrinsic signature of (an ensemble of) SLG in water : a symmetric, narrow and intense 2D band  $27 \text{ cm}^{-1} < \text{FWHM}_{2D} < 30 \text{ cm}^{-1}$  $2.7 < \text{A}_{2D}/\text{A}_{G} < 3.5$ 



### *SLG in "eau de graphene"* Defects





Exposition of graphenide to moisture  $\rightarrow$  functionnalisation by –H or –OH groups  $\rightarrow I_D$  is dominated by the contribution of point defects  $L_D$ =8-10 nm (300-400 ppm)



Upshift, broadening and decrease of 2D  $\rightarrow$  interlayer interactions  $I_D/I_G$  decreases ( $I_D/I_{D'}$  as well)  $\rightarrow$  curing of point defects

Bepete et al, J. Phys. Chem. C 2016





**I**<sub>D</sub> in annealed films is dominated by edge defects

 $< L_{flake} >_{measured} \approx 300 \text{ nm} \approx < L_{flake} >_{actual}$ 



### **SLG in "eau de graphene"** Doping and strain ?





Bepete et al, J. Phys. Chem. C 2016 \* Con

\* Considering a dispersion of 97 cm<sup>-1</sup>.eV<sup>-1</sup> for 2D









### Single layerness Narrow, intense 2D band $\rightarrow$ FWHM<sub>2D</sub>=28±2 cm<sup>-1</sup>

### Defects

D band dominated by sp<sup>3</sup> defects in EdG $\rightarrow L_D \approx 8-10$  nm (300-400 ppm) D band dominated by edge defects in annealed films $\rightarrow <L_a > \approx 300$  nm

### Electronic and mechanical interactions

(Heterogeneous ?) compressive strain (≈-0.1%)

Weak doping (<4.10<sup>12</sup> cm<sup>2</sup>)

### Coming soon...

Microscopic origin of strain ? Influence of charge density and pH ? Extend results to larger flakes