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Reaction Mechanism and Performances of Pioneer 2D Si_xGe_{1-x}H Electrodes in Lithium-ion Batteries

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Abstract

2D monoelemental materials, such as silicene and germanene, have recently received a lot of attention due to their tunable structures and remarkable physicochemical properties [1,2]. The addition of covalently bonded hydrogen in the z-orientation of each silicene and germanene atom results in hybridized bonding site changes from sp² to sp³ with new terms of silicane (*i.e.*, hydrogenterminated silicene, SiH) and germanane (*i.e.*, hydrogen-terminated germanene, GeH) [3,4]. However, challenges in the synthesis of such materials impede their detailed study, exploitation, and applications. In this work, we present pioneer 2D silicane and germanane materials with different compositions, namely, Si_{0.25}Ge_{0.75}H, Si_{0.5}Ge_{0.5}H, and Si_{0.75}Ge_{0.25}H that benefit from the simultaneous presence of both Si (high capacity) and Ge (high-rate performance and capacity retention). The three samples were synthesized through simple and efficient chemical exfoliation of bulk Zintl phases. The presence of both SiH and GeH in a single structure offers a synergistic beneficial effect on the electrochemical performances of $Si_xGe_{1-x}H$ (with x values from 0.25 to 0.75). Among them, the Si_{0.5}Ge_{0.5}H electrode shows the highest capacity, up to 1526 mAh g⁻¹ after 20 cycles. The Si_{0.5}Ge_{0.5}H electrodes were respectively, subjected to 0, 10, 100, and 200 half-cell cycles followed by ex-situ SEM and SEM-EDX analysis to assess the effects of lithiation/delithiation and alloying/dealloying on the Si_{0.5}Ge_{0.5}H flakes. Figure 1 shows the variations in morphology after different cycles.

References

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Figure 1: SEM and SEM-EDX images depicting the progression of deformation caused by lithiation and delithiation on the Si_{0.5}Ge_{0.5}H morphology upon cycling.