

Epitaxial synthesis of boron-containing Lewis acid material

Naoki Takahashi, Yuta Nishina

Graduate School of Environmental, Life, Natural Science and Technology, Okayama University,
 3-1-1 Tsushimanaka, kita-ku, Okayama, Japan
 n-takahashi@s.okayama-u.ac.jp

Boron-containing materials have been developed, and generally show Lewis acidity. Thus, those materials applied to a wide variety of applications such as anion sensors, electronic devices, and catalysts.^[1-3] Although those materials have unique properties, the proportion of boron is low and to control the structure is difficult. In our study, we aimed to synthesize a novel 2D material with high boron content in order to increase Lewis acidity. We expected that a lot of Lewis acid sites and high surface area of the 2D material improved the performance in application such as catalysts. We adopted bottom-up synthesis to tune material structure precisely. Triethynylborane-pyridine (teb·py) complex was chosen as monomer because of its high boron content (B:C = 1:6). poly(phenyl)borane ((BC₆)_n) was synthesized by cyclotrimerization of alkynes (Figure 1). The release of pyridine was observed at 300 °C in TG-MS measurement (Figure 2). This result indicated that pyridine was strongly coordinated to Lewis acid sites on (BC₆)_n, and it was found that the Lewis acidity of (BC₆)_n emerged by this heat treatment.

References

- [1] V. M. Suresh, A. Bandyopadhyay, S. Roy, S. K. Pati, T. K. Maji, *Chem. Eur. J.*, 21 (2015), 10799
- [2] S. Kawai, S. Saito, S. Osumi, S. Yamaguchi, A. S. Foster, P. Spijker, E. Meyer, *Nat. Commun.*, 6, (2015), 6.
- [3] Y. Lin, S. Wu, W. Shi, B. Zhang, J. Wang, Y. A. Kim, M. Endo, D. S. Su, *Chem. Commun.* 51, (2015), 13086

Figures

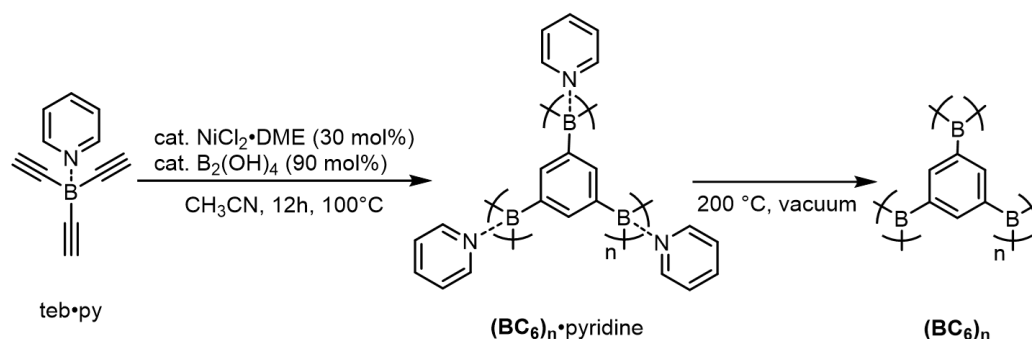


Figure 1: Synthesis route of (BC₆)_n

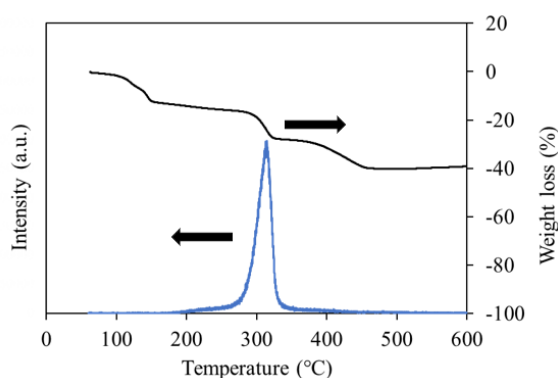


Figure 2: TG-MS measurement of (BC₆)_n·pyridine