

Boosting the Electrocatalytic Activity of Chemically Exfoliated MoS₂ by Surface Functionalization

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The vibrant research area on ultrathin two-dimensional (2D) layered materials has continually shown that the unique and superlative properties firstly observed for graphene find a significant equivalence in other 2D materials as the transition metal dichalcogenides (TMDs). The MoS₂ is an important example which has triggered broad and intense investigations. The 2H-to-1T phase conversion is a semiconductor-to-(semi)metal transition with huge impact on its applications, mainly for electrochemical technologies. The surface functionalization of MoS₂ with a wide variety of species such as metal atoms, organic molecules, metallic nanoparticles, and polymers has been explored as an efficient strategy to modify its intrinsic properties [1,2], for multiple purposes and applications, including phase engineering. In this study, by an experimental-theoretical approach, we show how the chemical functionalization of 1T'-MoS₂ with iodoacetic acid molecules improves both the structural stability and the electrocatalytic properties of the metallic nanosheets for the hydrogen evolution reaction (HER) (Figure 1). The presence of -CH₂COOH groups on the MoS₂ surface boosts the electrocatalytic effects for the hydrogen evolution reaction, reducing the charge transfer resistance of the interface and preventing the oxidation and deactivation of the active sites.

References

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Figures

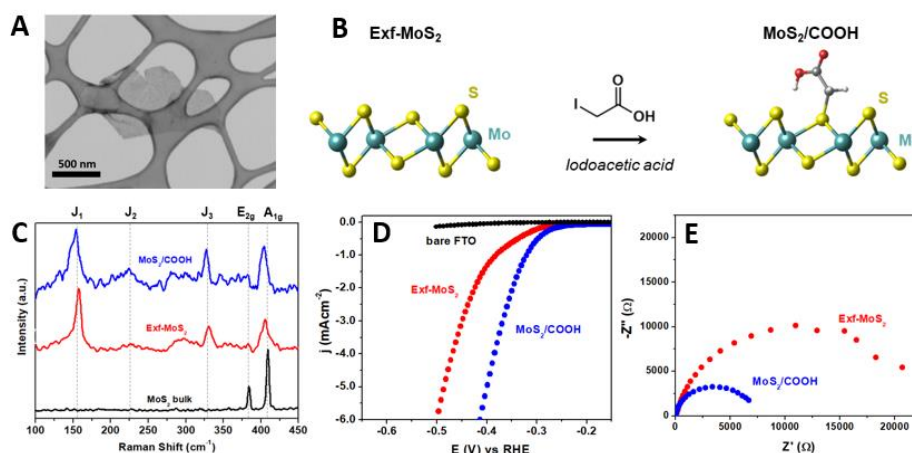


Figure 1: (A) STEM image of chemically exfoliated MoS₂; (B) schematic illustration of the functionalization of MoS₂ nanosheets; (C) Raman spectra of pristine and functionalized MoS₂ and (D-E) electrochemical HER studies of MoS₂ and MoS₂/COOH samples.

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