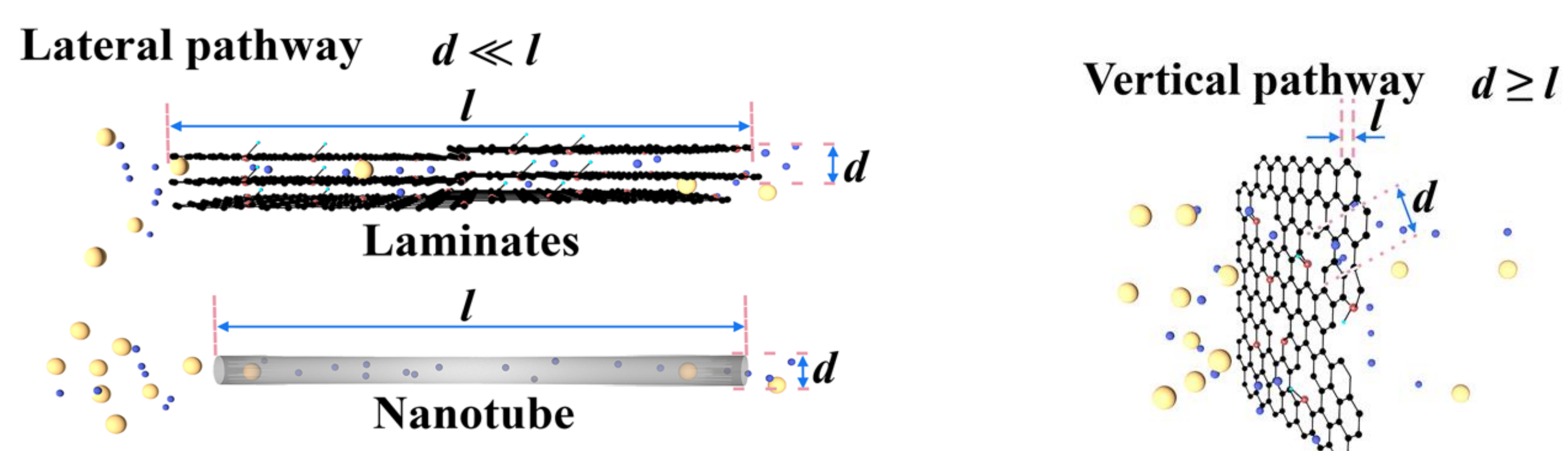


A proton conductive graphene composed membrane for direct methanol fuel cell (DMFC)

Weizhe Zhang, Xue Liu, Gregory F. Schneider

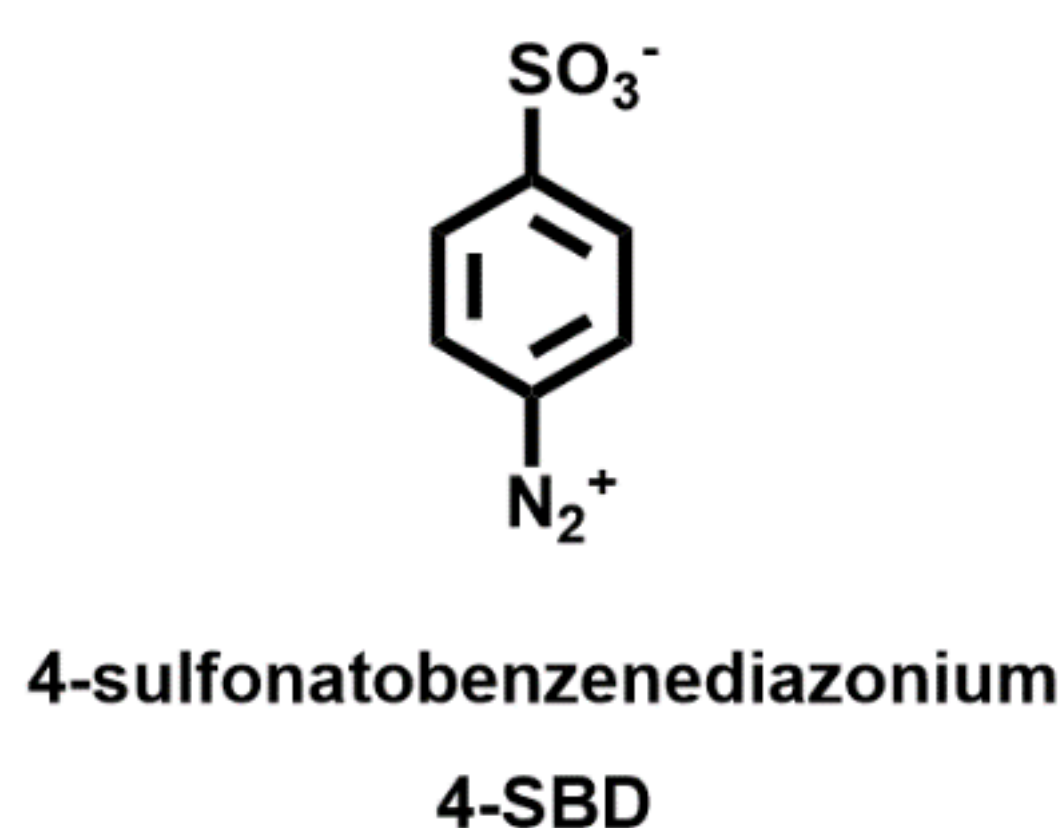
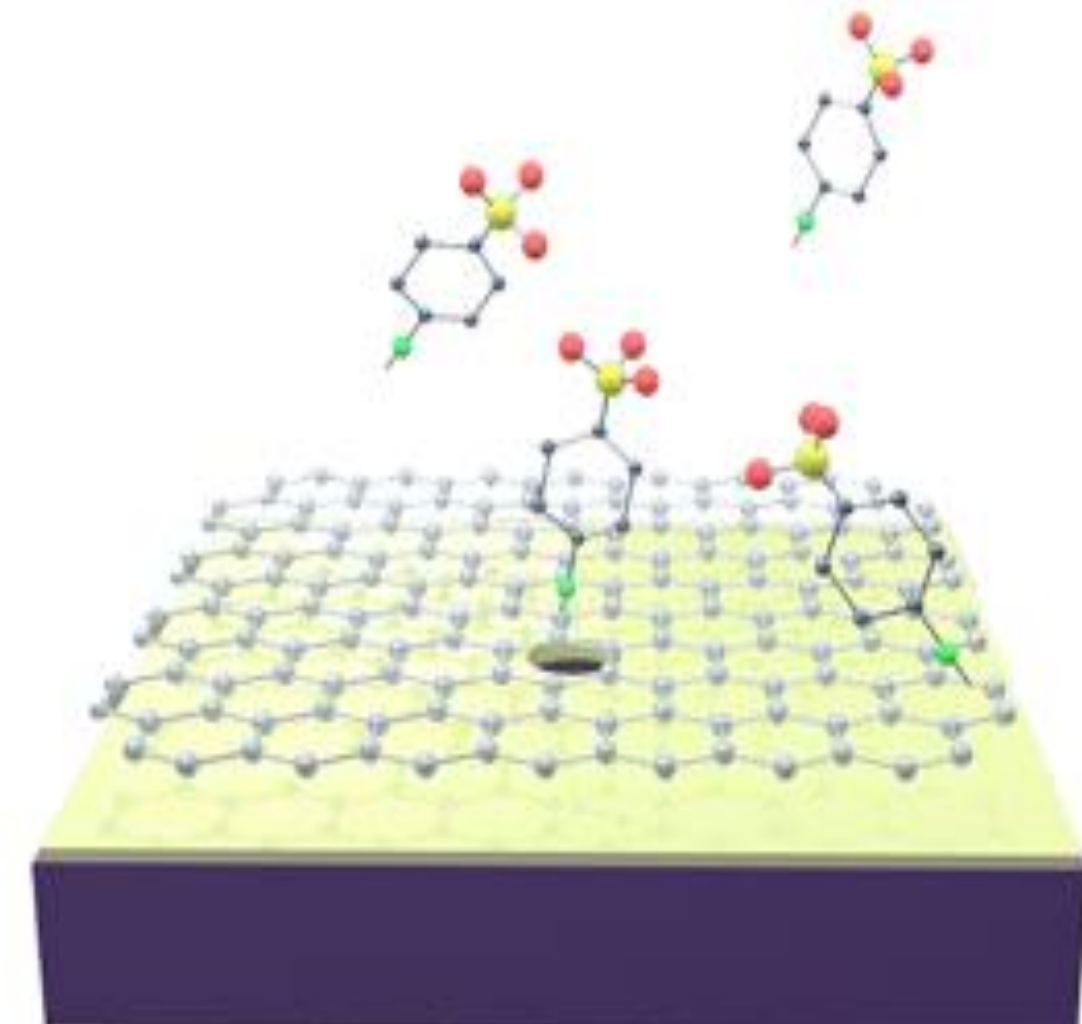
Leiden Institute of Chemistry, Leiden University

2D porous membrane

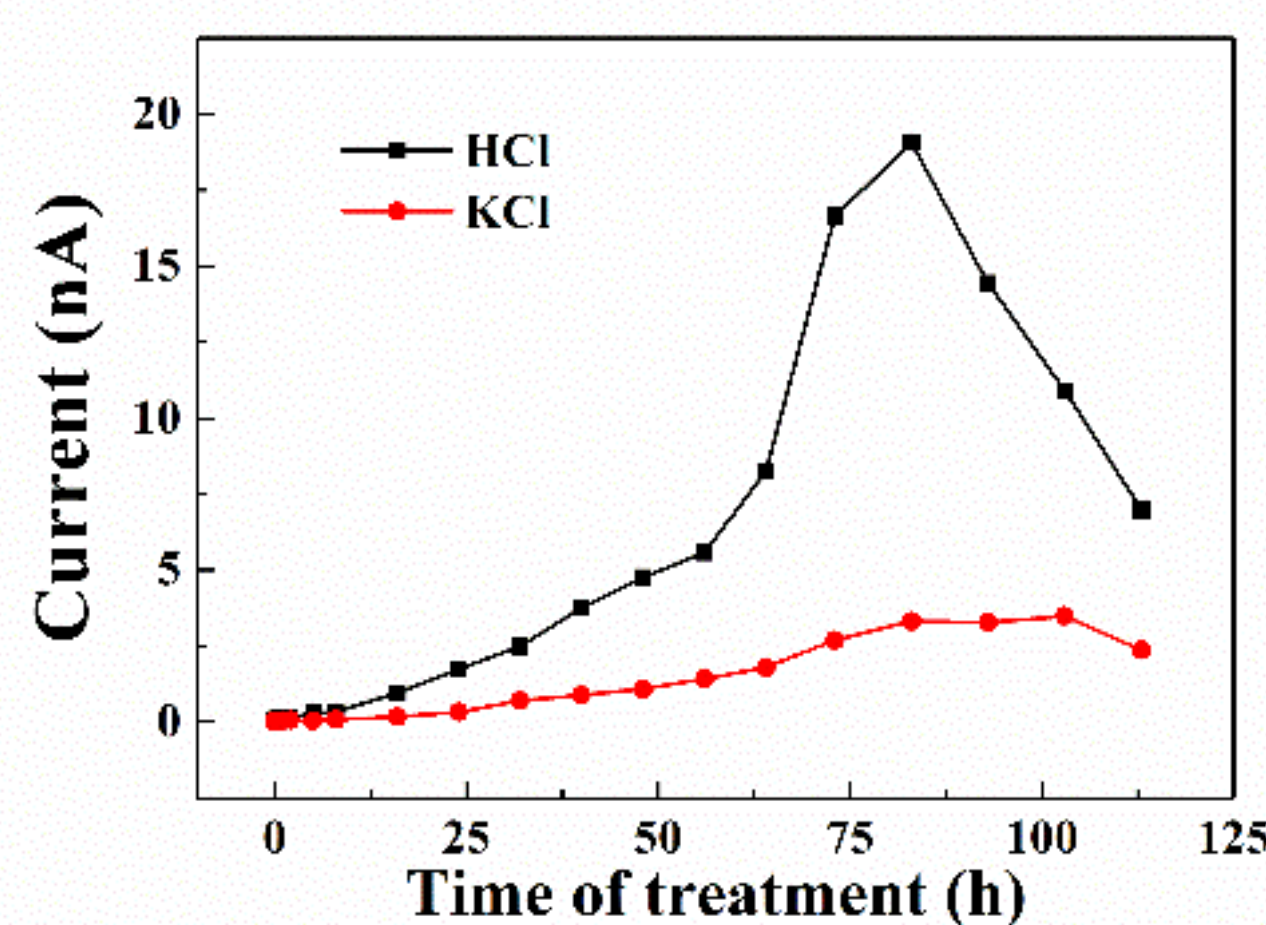
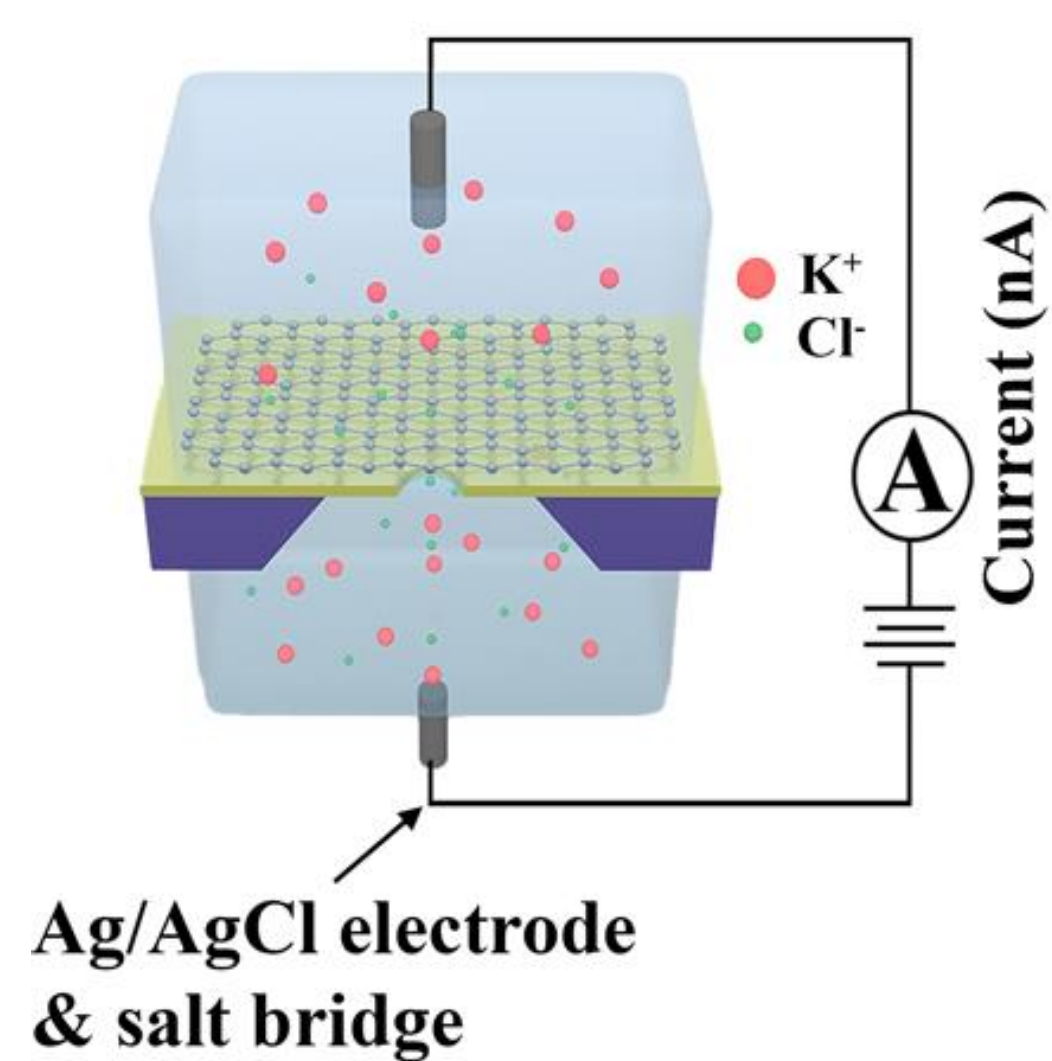


Shortening transport distance to **atomic size** endows possibilities for membrane of new generation while 2D membrane with high proton permselectivity remains challenging¹⁻⁵.

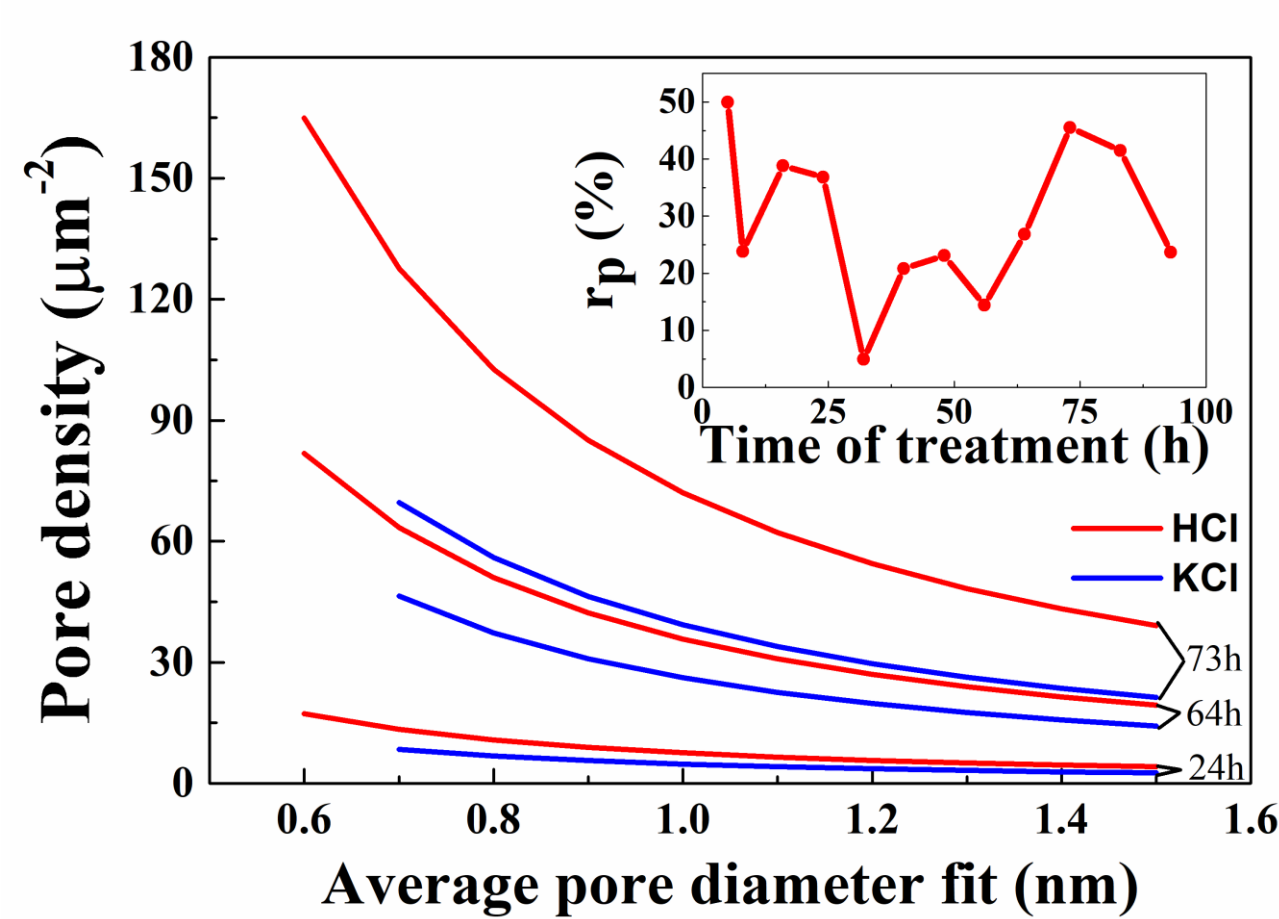
Proton conductive graphene



Illustrate: SBD treated graphene. High quality graphene fixed at SiN/Si chip with 1 μm aperture.



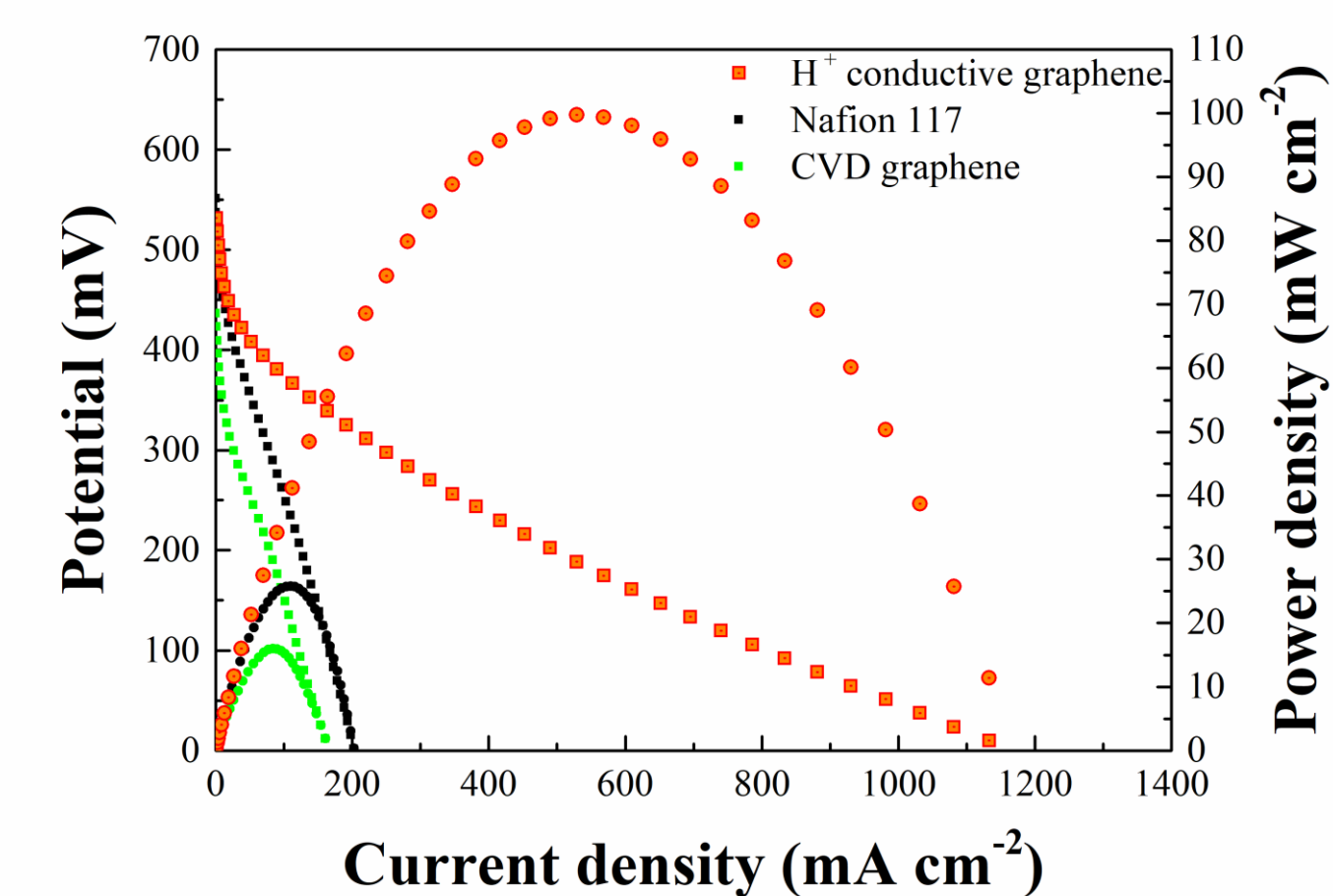
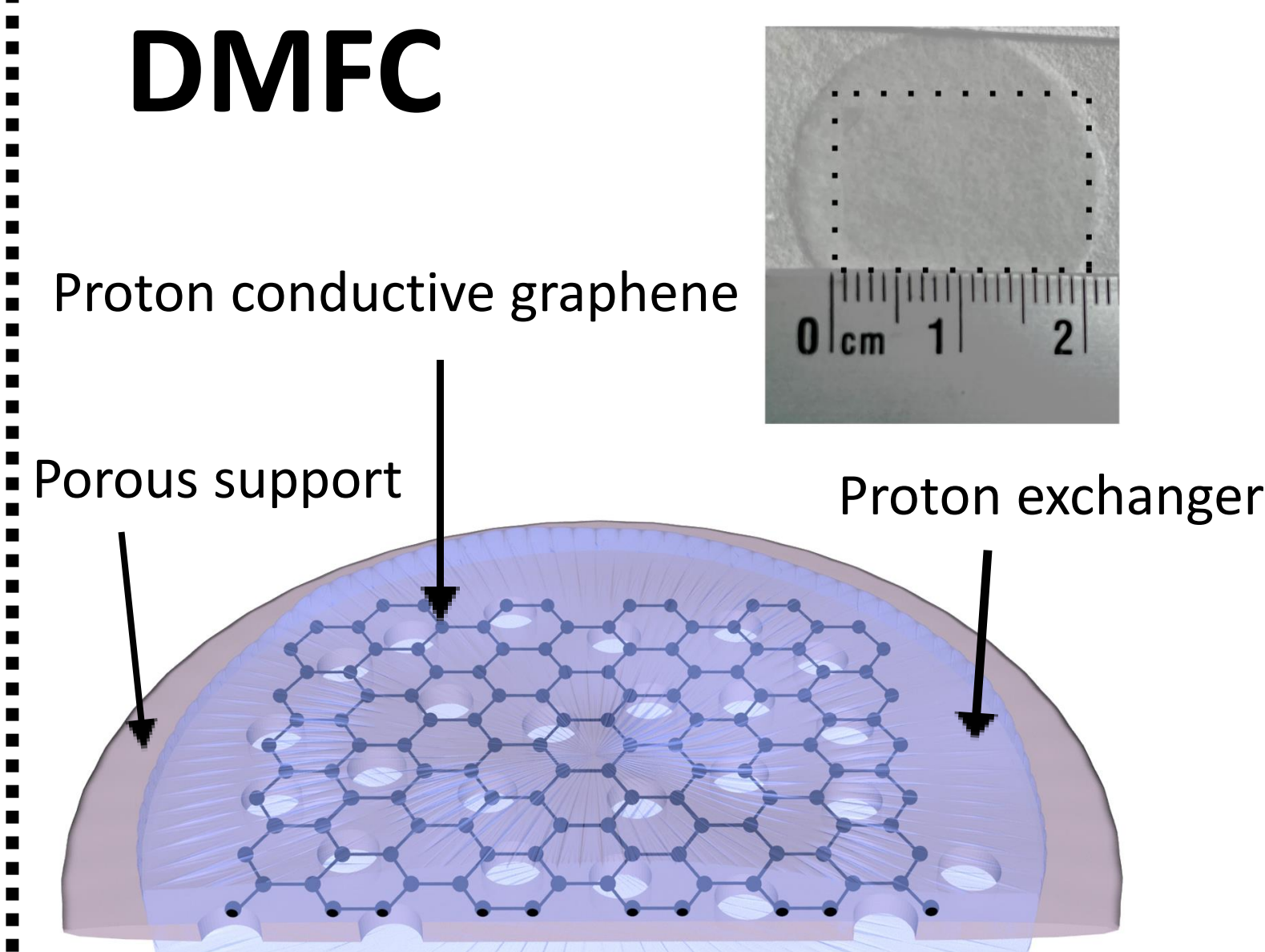
Ionic current measurements tracked the conductance variation with treatment time in 0.1 M solutions.



Pore density. a, Pore density as a function to assumed pore diameter, Inset: ratio of pores only allow proton passage which indicates the H⁺/K⁺ selectivity.

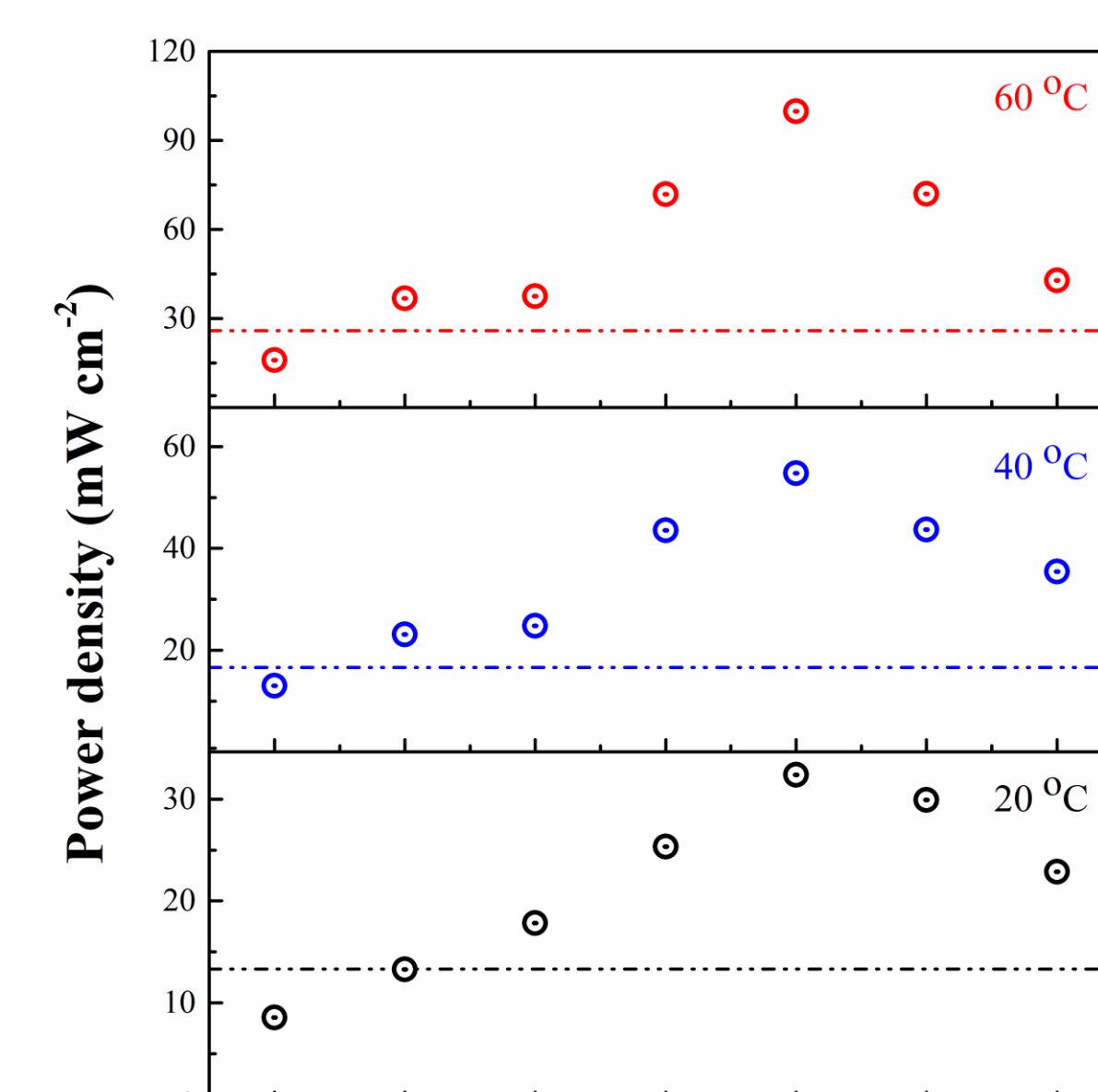
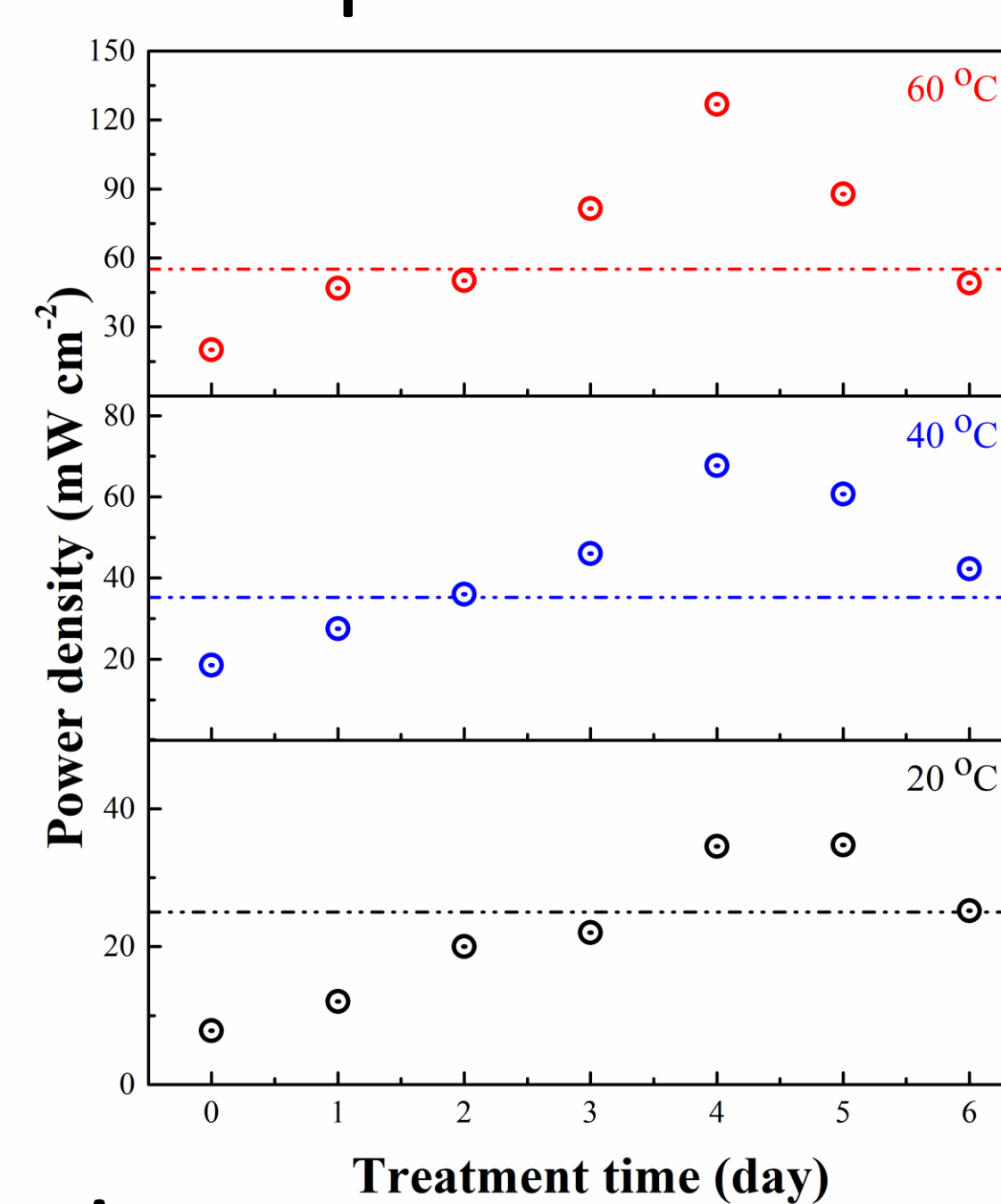
Proton selectivity up to **~48%** with respect to K⁺
Conductance reaches **18 s cm⁻²** in 0.1 M HCl

Proton conductive graphene membrane in DMFC

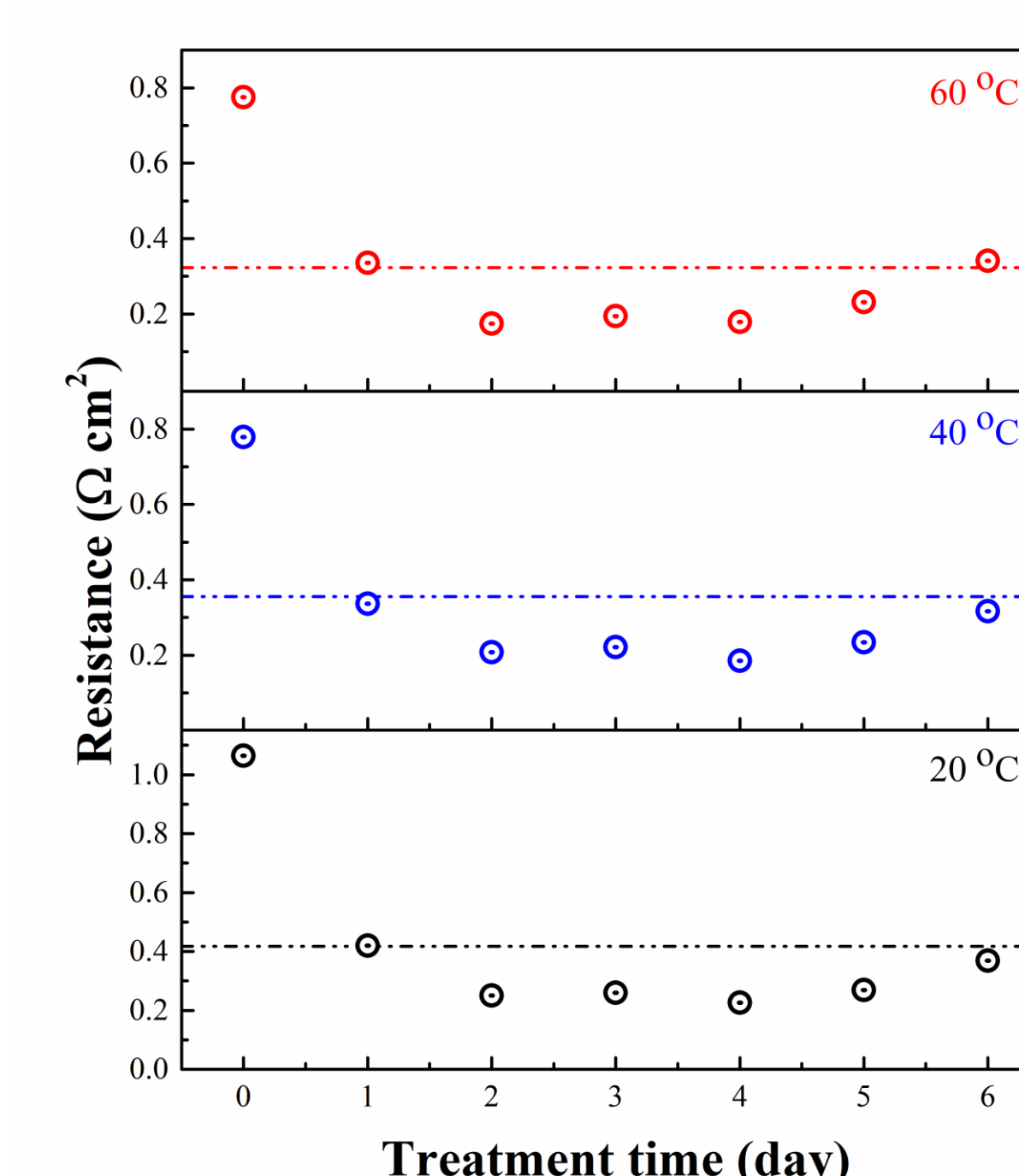
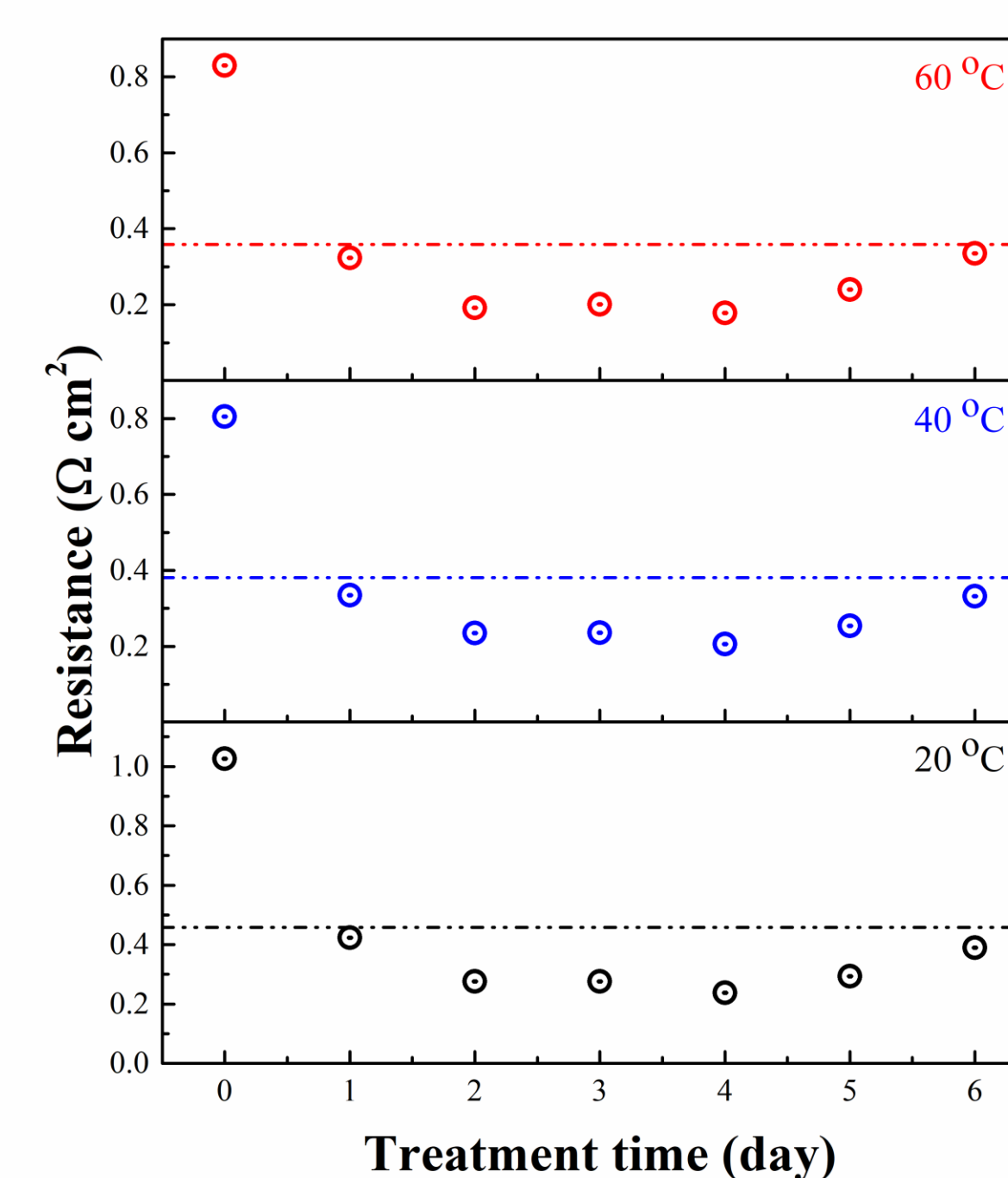


Polarization and power density at 60 °C with 5 M methanol

Max. power output



Proton resistance



Dash lines indicate data obtained from Nafion 117

Conclusion

In this work, diazonium treatment as a facile method was applied for proton conductive graphene fabrication which is suitable for the large-scale membrane applications. As-treated graphene reaches 18 s cm⁻² in 0.1 M HCl with high proton selectivity. Such proton conductive graphene composed membrane boosts ~4 times power output comparing with Nafion 117, and it is also less sensitive to methanol concentration.

CONTACT PERSON

Weizhe Zhang

w.zhang@lic.leidenuniv.nl

REFERENCES

1. Thiruraman, J. P., et al. *Advanced Functional Materials* **29**, 9, doi:10.1002/adfm.201904668 (2019).
2. Koenig, S. P., et al. *Nature Nanotechnology* **7**, 728-732, doi:10.1038/nnano.2012.162 (2012).
3. Jang, D., et al. *ACS Nano* **11**, 10042-10052, doi:10.1021/acsnano.7b04299 (2017).
4. Hu, S. et al. *Nature* **516**, 227+, doi:10.1038/nature14015 (2014).
5. Chaturvedi, P. et al. *ACS Nano* **13**, 12109-12119, doi:10.1021/acsnano.9b06505 (2019).