

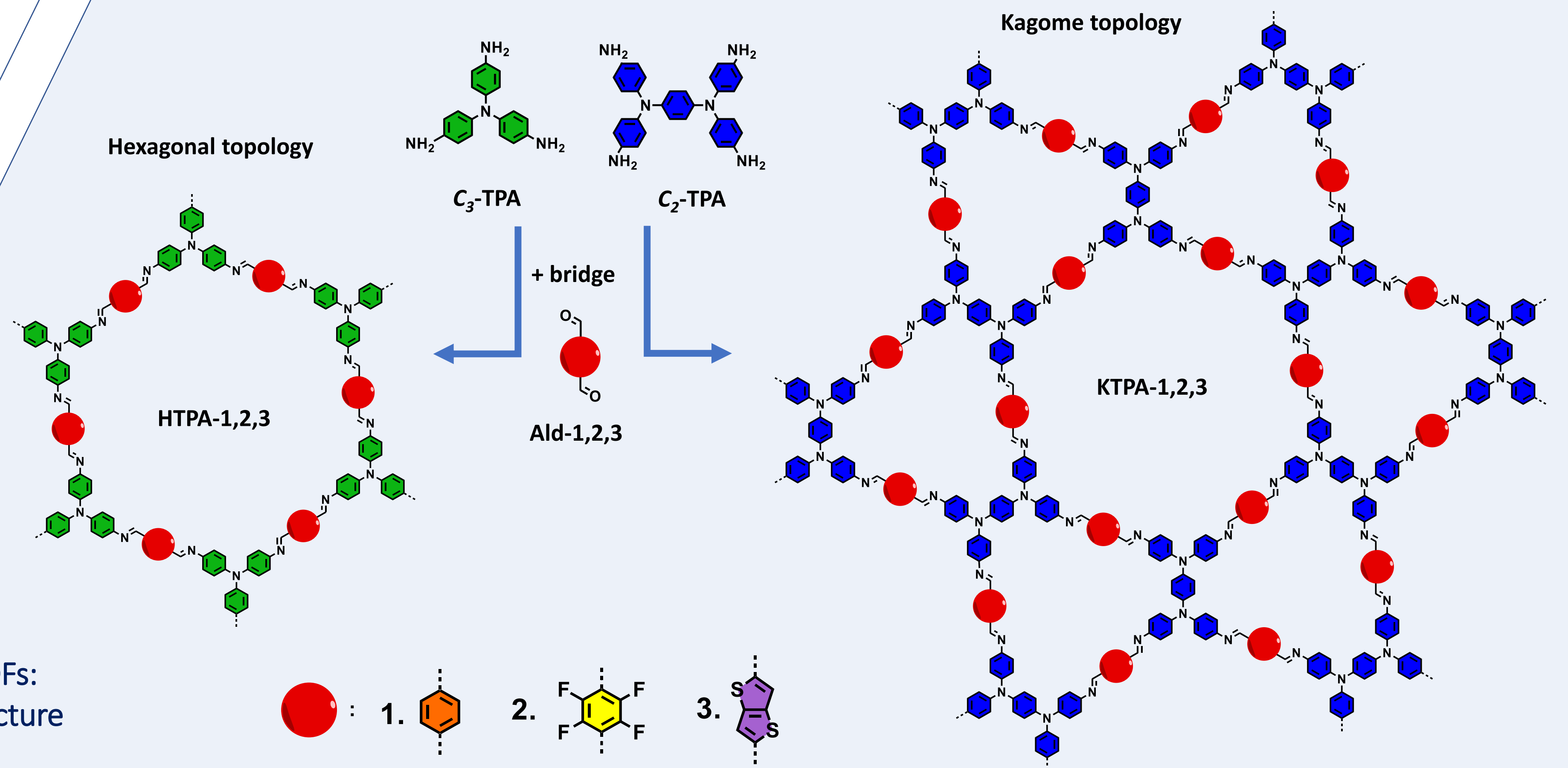
## Topology-related optoelectronic processes in triphenylamine-based 2D covalent organic frameworks

2D COFs are appealing candidates for advanced applications in optoelectronic devices. The combination between the vast choice of building-blocks and linkages available enable the fabrication of polymeric materials with specific functionalities and adjustable plane-extended  $\pi$ -conjugation.[1]

Recently, triphenylamine (TPA)-based moieties have emerged as possible building-blocks in 2D COFs. TPA is a thermally stable, propeller-shaped molecule which exhibits interesting photoactive and electroactive behaviors. These features are related to the stability of corresponding radical cation, easily generated by mono-electron oxidation. TPA-based advanced materials such as molecular derivatives, linear and branched polymers have been developed for various optoelectronic applications such as photoconductive, light-emitting, electrochromic devices and especially as hole transporting materials.[2]

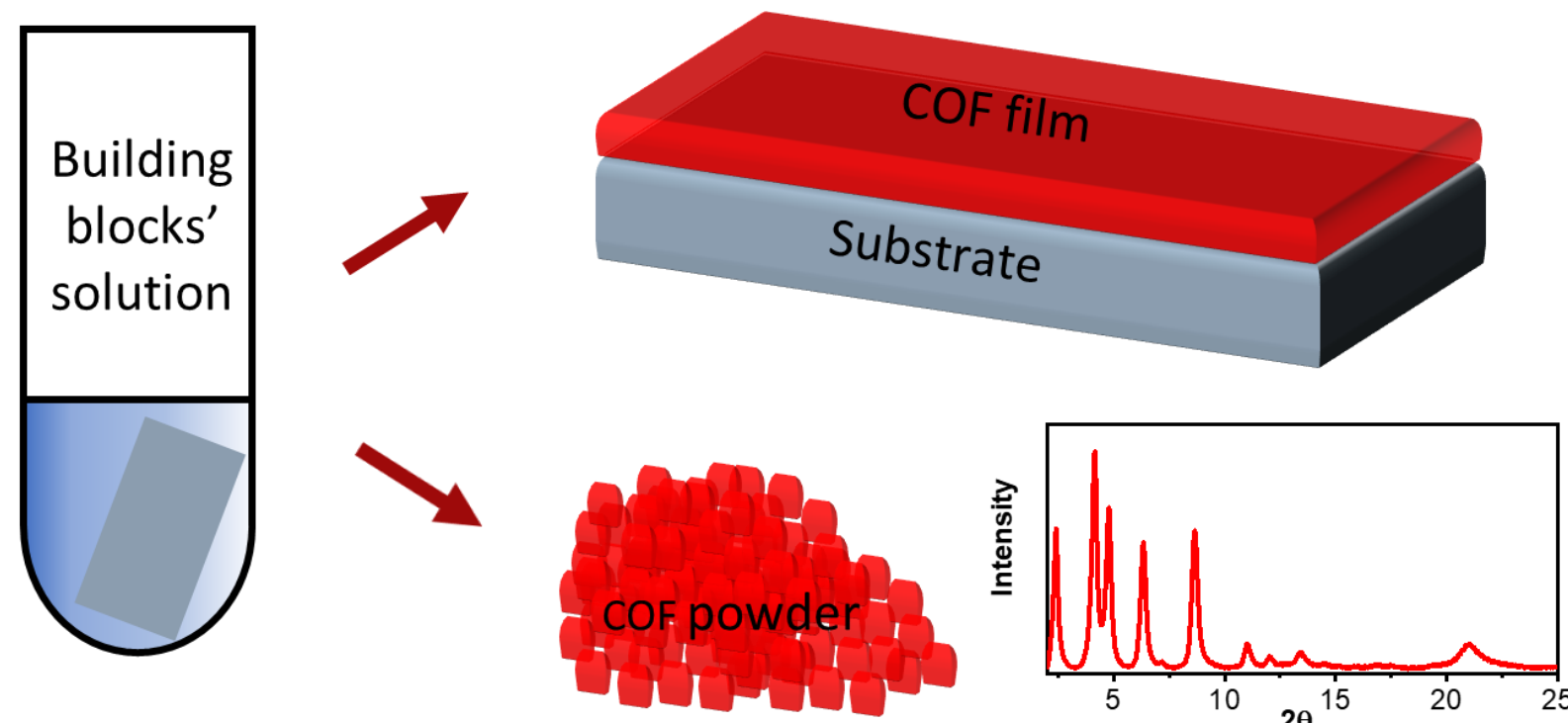
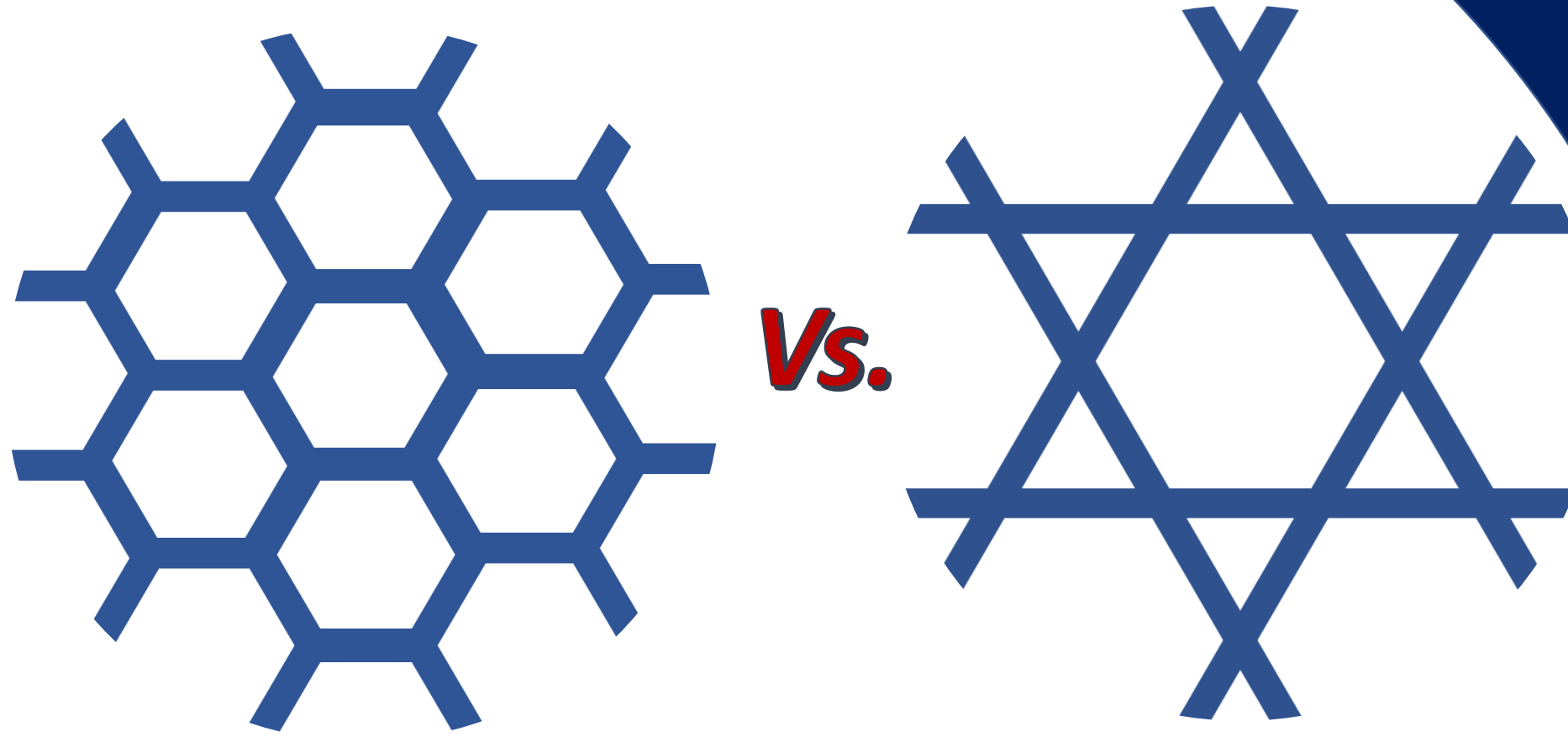
Herein, we report the integration of two types of TPA moieties in 2D COF thin films. Polymerization of  $C_3$ -TPA with linear dialdehydes leads to the formation of a hexagonal topology while the reaction of the  $C_2$ -TPA with the same bridges leads to the formation of a Kagome topology.[3] Our results show photoconductive and hole-transporting properties of the resulting materials and suggest that the topology of the framework might have an impact on the final behavior.[4]

TPA-based COFs:  
chemical structure

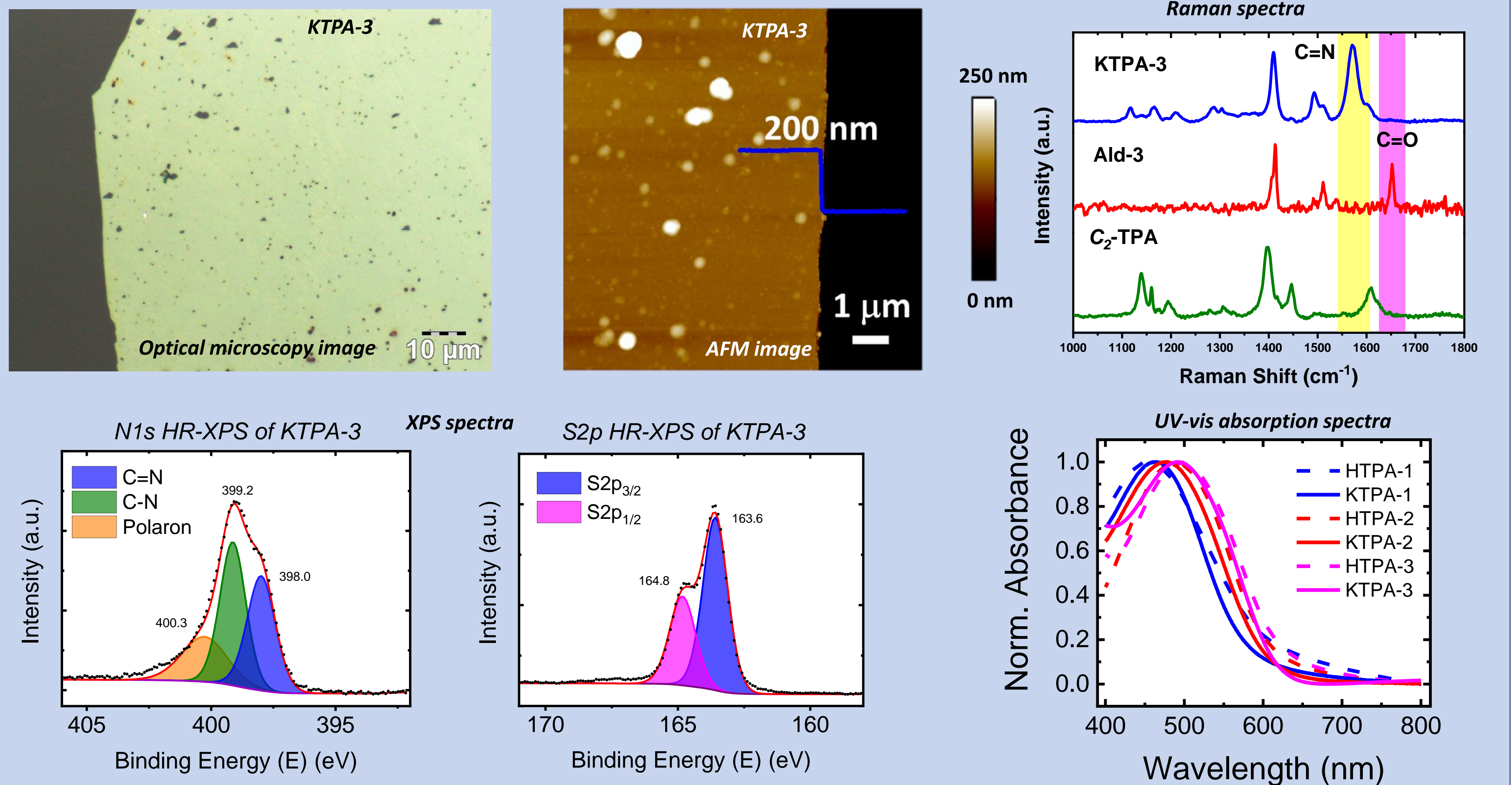


### Aim

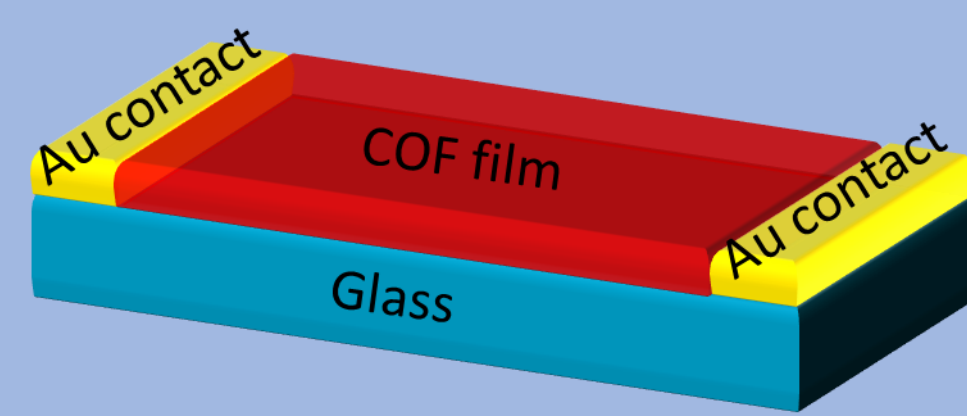
Study and compare the (opto)electronic properties of TPA-based COFs with hxl and kgm topologies



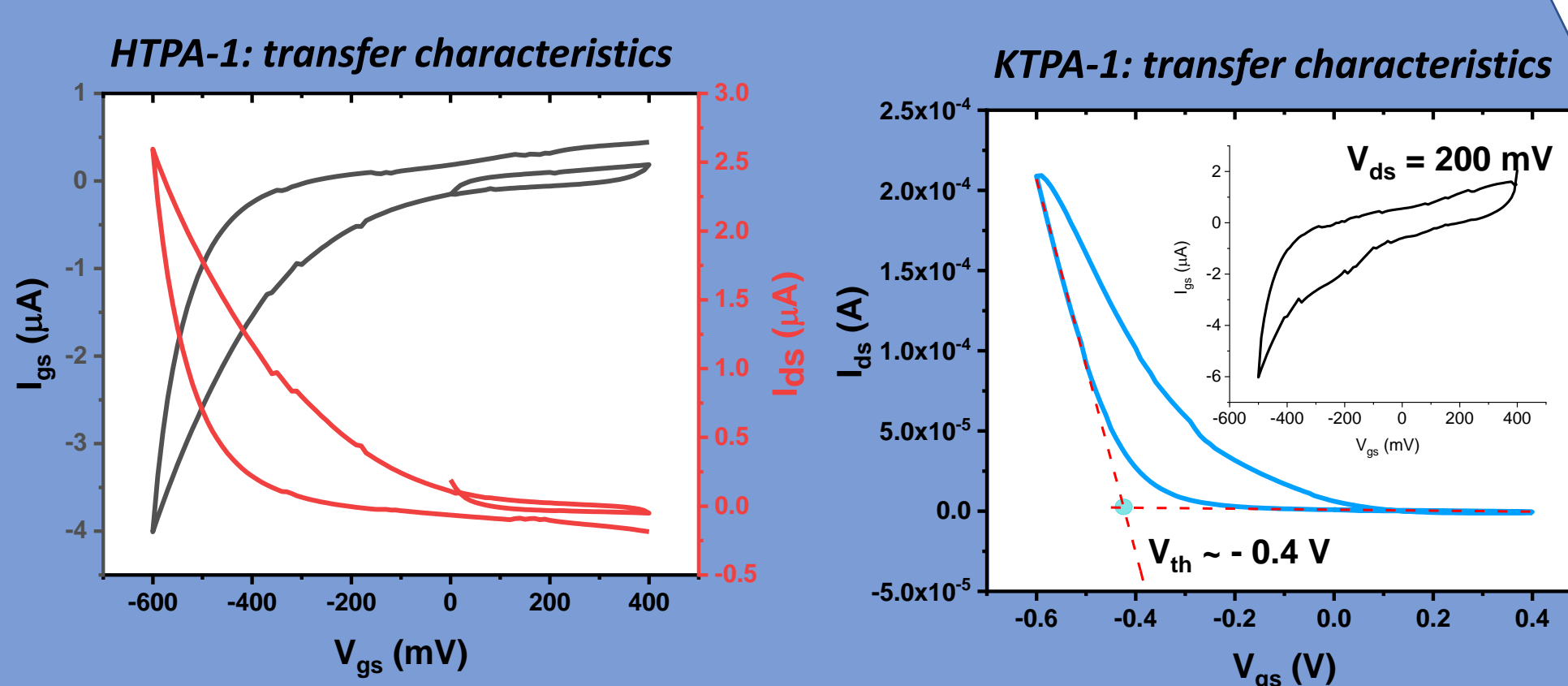
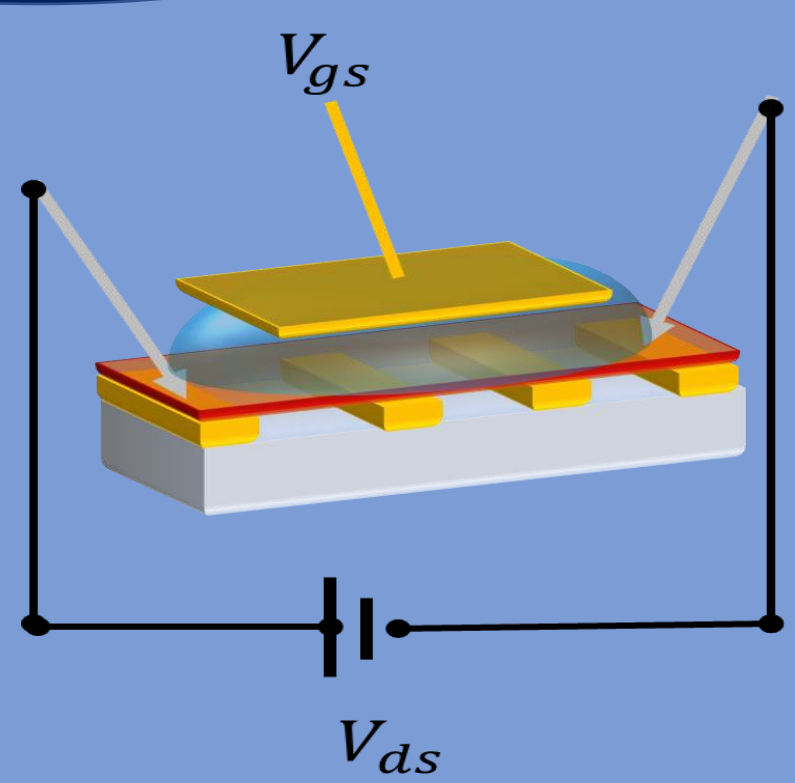
### Thin films: morphological and chemical characterization



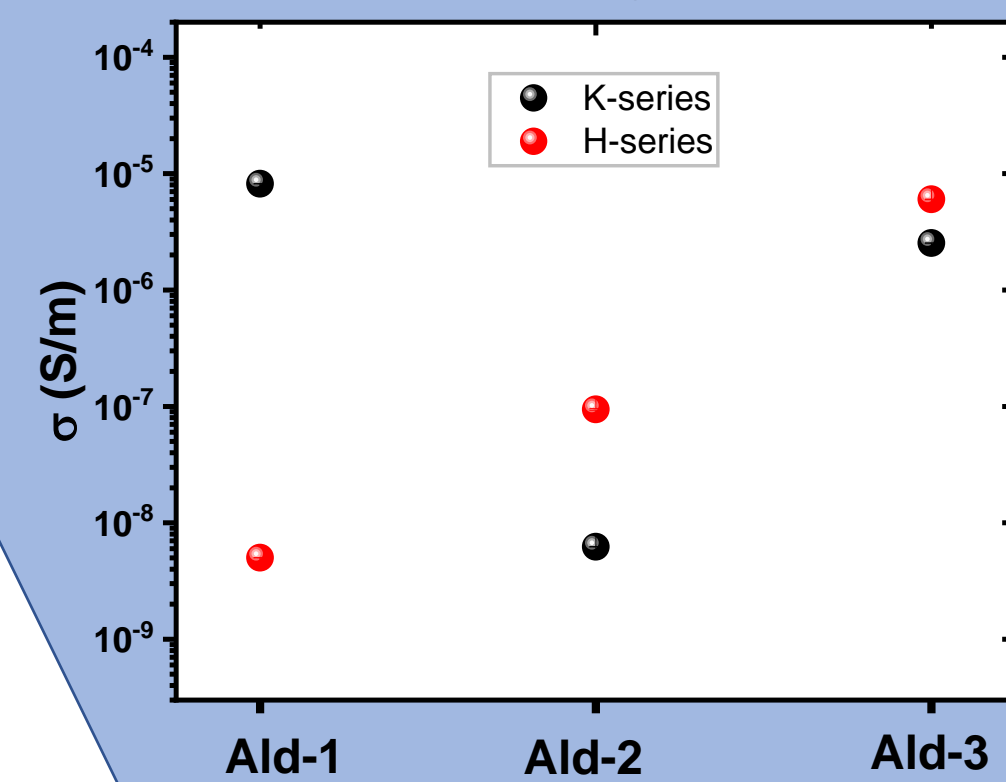
### Electrical characterization: 2-terminal device



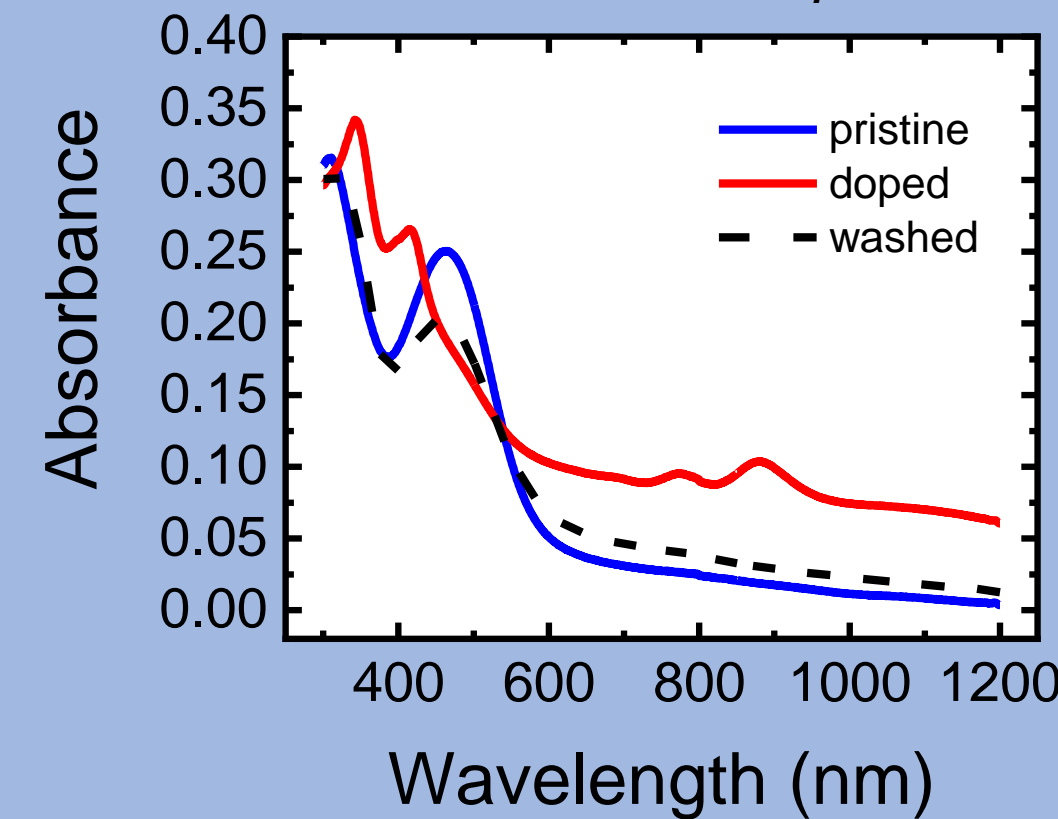
### Electrical characterization: 3-terminal device



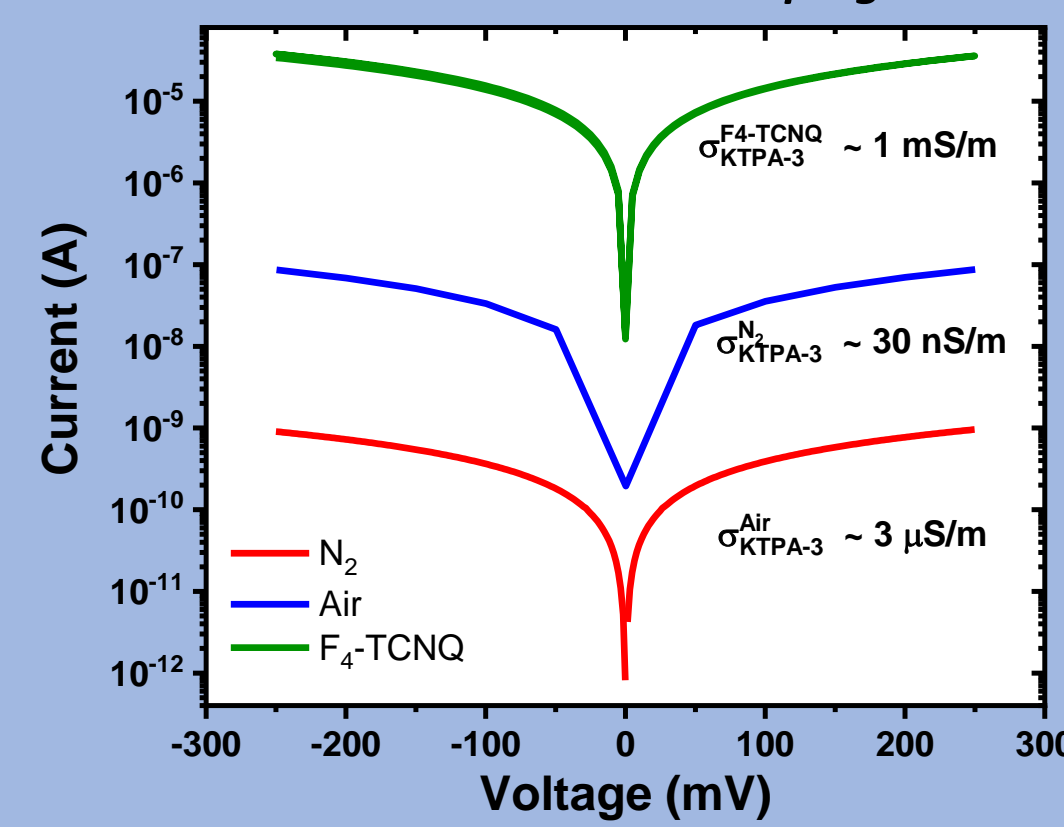
### Conductivity



### Doping KTPA-1: F4TCNQ



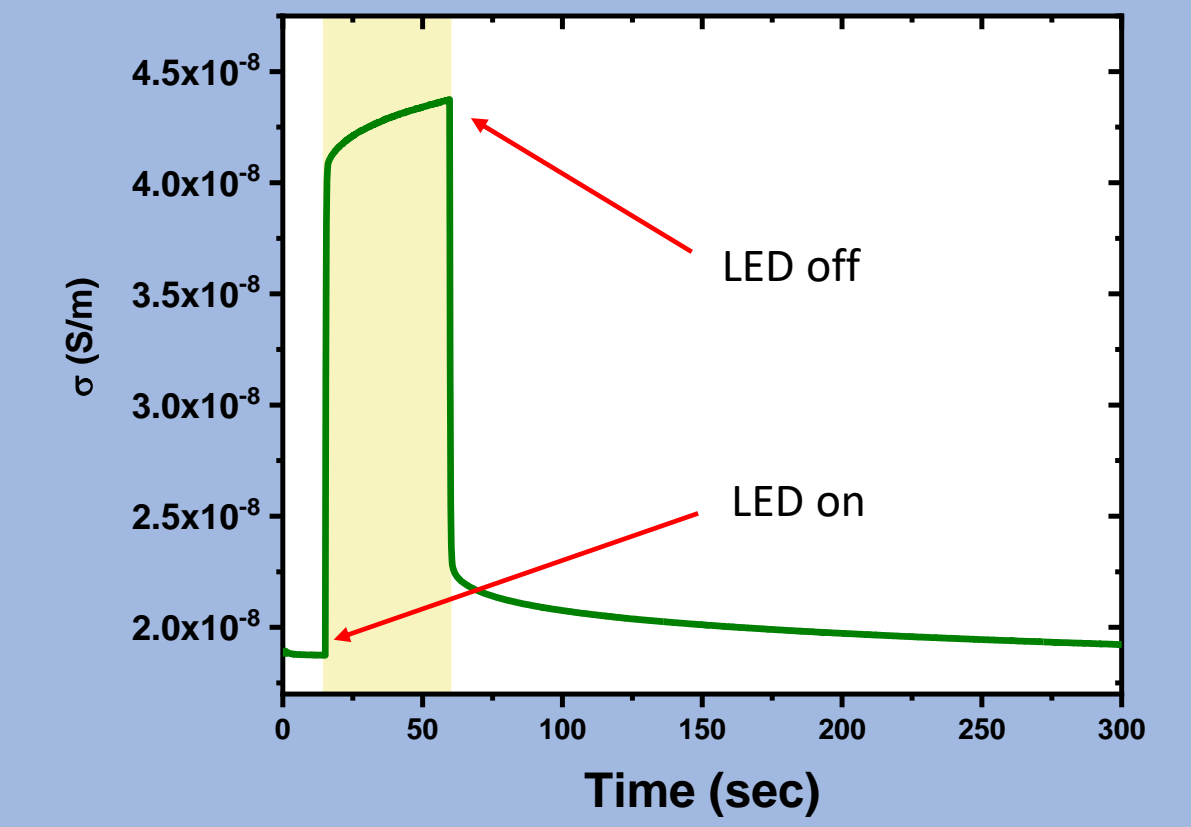
### KTPA-3: oxidative doping



### Optical bandgaps

Thin film	$\lambda_{max,abs}$ (nm)	$E_{g,opt}$ (eV)
HTPA-1	454	2.13
KTPA-1	463	2.24
HTPA-2	487	2.13
KTPA-2	476	2.18
HTPA-3	497	2.06
KTPA-3	493	2.11

### KTPA-3: photoconductivity



### CONTACT PERSON

lcusin@unistra.fr

### REFERENCES

- [1] N. Keller, T. Bein, *Chemical Society Reviews*, 50 (2021) 1815-1845
- [2] H. Yen, G. Liou, *Progress in Polymer Science*, 89 (2019) 250-287
- [3] J. Rotter, R. Guntermann, *et. al.*, *Chemical Science*, 11 (2020) 12843-12853
- [4] E. Jin, K. Geng, *et. al.*, *Angewandte Chemie International Edition*, 59 (2020) 12162-12169