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2D transition metal carbide for catalysis

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Transition-metal carbide (TMC), owing to its electronic conductivity, chemical stability and physical properties, has aroused widespread interests in catalysis. Here, we have systematically studied the photocatalytic hydrogen (H₂) evolution of metallic cobalt carbide (Co₂C) by a combination of theoretical and experimental investigations. In term of intrinsic proton reduction property of Co₂C (020) facet and facile interfacial electron transfer, the assembled architecture of QDs/Co₂C can give an rate of ~18000 µmol g⁻¹ h⁻¹ (λ = 450 nm) using TMC as cocatalysts and an apparent quantum yield of ~2.7% of photocatalytic H₂ evolution, a ~10-fold enhancement compared with bare QDs under identical conditions. Our results indicate that Co₂C with suitable morphology and facet exposure can work as a cocatalyst to achieve photocatalytic H₂ evolution.

References

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Figures

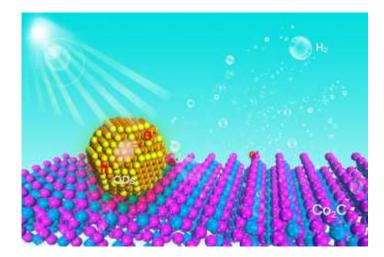


Figure 1: The application of ultrathin Co₂C nanosheets in photocatalytic hydrogen evolution.